Access DB# 106833

SEARCH REQUEST FORM

Scientific and Technical Information Center

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Requester's Full Name: HELEN		Examiner # : <u>700 (P</u> Serial Number: <u>/0/</u>	Date: 10/27/03	
Art Unit: 1913 Phone Nu Mail Box and Bldg/Room Location:	mber 30 <u>8- 23 93 4</u> CP3-8B16 Result	s Format Preferred (circle)	PAPER DISK E-MAI	IL
If more than one search is submit	ted, please prioritize	searches in order of n	eed. ********	**
Please provide a detailed statement of the se Include the elected species or structures, key utility of the invention. Define any terms th known. Please attach a copy of the cover sh	/words, synonyms, acronyl at may have a special mea	ns, and registry numbers, and ning. Give examples or releva	ant citations, authors, etc, if	
Title of Invention:	ATTACHED	Climent 1	additive), comes	ete,
Inventors (please provide full names):		montan		
<u> </u>				
Earliest Priority Filing Date:/2,		_		
For Sequence Searches Only Please include appropriate serial number.	all pertinent information (po	rent, child, divisional, or issued	patent numbers) along with the	
A polycaboxylic	acid cop	algories des	ined from.	
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2) a unsatmatet	Can box glic	and menor		
3) another palya	explene fly	cal undating	ted manamed	
	(horas or an all P	A GL (Almor) /	•	1 4
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malie acid (a	, , ,	,,	>	
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4) 3- mercaptoprop.	mie acid	, butanethia	₹.	
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STAFF USE ONLY	**************************************	Vendors and cost	where applicable	
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Searcher Location:	Structure (#)	Questel/Orbit		•
Date Searcher Picked Up:	Bibliographic	Dr.Link		
Date Completed: $\frac{10/28/03}{}$	Litigation	Lexis/Nexis		
Searcher Prep & Review Time:	Fulltext	Sequence Systems		
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Online Time:	Other	Other (specify)		•
PTO-1590 (8-01)				

=> FILE REG

FILE 'REGISTRY' ENTERED AT 16:32:50 ON 28 OCT 2003
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STRUCTURE FILE UPDATES: 27 OCT 2003 HIGHEST RN 609766-09-8 DICTIONARY FILE UPDATES: 27 OCT 2003 HIGHEST RN 609766-09-8

TSCA INFORMATION NOW CURRENT THROUGH JULY 14, 2003

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Experimental and calculated property data are now available. See HELP PROPERTIES for more information. See STNote 27, Searching Properties in the CAS Registry File, for complete details: http://www.cas.org/ONLINE/STN/STNOTES/stnotes27.pdf

-> FILE HCAPLUS

FILE 'HCAPLUS' ENTERED AT 16:32:55 ON 28 OCT 2003
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FILE COVERS 1907 - 28 Oct 2003 VOL 139 ISS 18 FILE LAST UPDATED: 27 Oct 2003 (20031027/ED)

This file contains CAS Registry Numbers for easy and accurate substance identification.

=> D QUE L23

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L2
12 SEA FILE=REGISTRY ABB=ON (106-91-2/BI OR 106-92-3/BI OR
125121-38-2/BI OR 26915-72-0/BI OR 31694-55-0/BI OR 37371-89-4/
BI OR 53694-15-8/BI OR 75-21-8/BI OR 760-93-0/BI OR 79-10-7/BI
OR 79-41-4/BI OR 9002-98-6/BI)

L5
2 SEA FILE=REGISTRY ABB=ON L2 AND 1-20/N
L6
5 SEA FILE=REGISTRY ABB=ON L2 AND PROPENOIC
L7
5 SEA FILE=REGISTRY ABB=ON L2 NOT (L5 OR L6)
L8
2 SEA FILE=REGISTRY ABB=ON L7 AND PROPENYL
L9
1 SEA FILE=REGISTRY ABB=ON L8 AND ALPHA
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PEZZUTO 10/025588
                     10/28/03
                                Page 2
              4 SEA FILE=REGISTRY ABB=ON L7 NOT L9
L10
              6 SEA FILE=REGISTRY ABB=ON L9 OR L6
L12
L13
          46474 SEA FILE=HCAPLUS ABB=ON L12
                                        L10
          20517 SEA FILE=HCAPLUS ABB=ON
L14
           8265 SEA FILE=HCAPLUS ABB=ON
L15
             14 SEA FILE=HCAPLUS ABB=ON L13 AND L14 AND L15
L16
          11421 SEA FILE=HCAPLUS ABB=ON (L13 OR L14 OR L15)(L)(PREP OR IMF OR
L17
                SPN)/RL
L18
              4 SEA FILE=HCAPLUS ABB=ON L16 AND L17
          23913 SEA FILE=HCAPLUS ABB=ON
                                        L12/D
L19
                                        L10/D
L20
           3864 SEA FILE=HCAPLUS ABB=ON
           2119 SEA FILE=HCAPLUS ABB=ON L5/D
L21
              6 SEA FILE=HCAPLUS ABB=ON L19 AND L20 AND L21
L22
              7 SEA FILE=HCAPLUS ABB=ON L18 OR L22
L23
=> D L23 ALL 1-7 HITSTR
    ANSWER 1 OF 7 HCAPLUS COPYRIGHT 2003 ACS on STN
L23
     2002:906328 HCAPLUS
AN
DN
     138:5635
TΙ
     Azlactone-functional reactive hydrophilic coatings and hydrogels and
     articles comprising coated substrates useful for immobilization of
    biological materials
    Haddad, Louis C.; Hembre, James I.; Rasmussen, Jerald K.; Sarpong, Daniel
IN
PA
     3M Innovative Properties Company, USA
SO
     PCT Int. Appl., 30 pp.
     CODEN: PIXXD2
DT
     Patent
LA
    English
IC
    ICM C08F008-32
         C08F220-56; C08F226-06; C09D133-26; C08J007-12; C08J003-24;
         C12N011-08; C08K005-17; C08K005-5455; C08L033-26
CC
     42-3 (Coatings, Inks, and Related Products)
     Section cross-reference(s): 9, 37
FAN.CNT 1
     PATENT NO.
                     KIND DATE
                                          APPLICATION NO. DATE
                     ----
    WO 2002094890
                                          WO 2002-US5433 20020222
ΡI
                     A1 20021128
        W: AE, AG, AL, AM, AT, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH,
            CN, CO, CR, CU, CZ, CZ, DE, DE, DK, DK, DM, DZ, EC, EE, EE, ES,
             FI, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG,
             KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW,
            MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SK,
             SL, TJ, TM, TN, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZM, ZW, AM,
            AZ, BY, KG, KZ
         RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, CH,
             CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR,
             BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG
    US 2003049435
                                          US 2001-860944 20010518
                      A1
                            20030313
PRAI US 2001-860944
                      Α
                            20010518
     Surface coatings from azlactone-functional hydrogels and articles
AB
     comprising a substrate which is a film, a plate, a particle, a fiber, a
     column, a bead, a web or a membrane with the coatings disposed thereon are
    disclosed. Methods of making the coating and controlling the gelation
     time of the hydrogels by providing a suitable crosslinking agent, e.g. a
     compound containing primary and secondary amino groups are also disclosed.
     coatings have residual azlactone functionality which can be used for
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covalent attachment (immobilization) of biol. or other functional materials. Thus, a 40% solids Me Et ketone solution of 80:20 weight/weight dimethylacrylamide-vinyldimethylazlactone copolymer prepared by standard free radical polymerization was diluted to 20% solids with isopropanol, formulated with enough ethylenediamine to provide a crosslink d. of ≈ 10% by weight, then applied to a com. 1536-well plate. Upon drying, a reactive, azlactone-functional polymeric coating was obtained within the wells. azlactone reactive vinyl polymer hydrogel coating immobilization ST IT RL: TEM (Technical or engineered material use); USES (Uses) (ambifunctional, primers; azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials) Polyoxyalkylenes, reactions ΙT RL: RCT (Reactant); RACT (Reactant or reagent) (amino-terminated, crosslinking agents; azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials) IT Hydrogels Immobilization, molecular Microtiter plates (azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials) IT Ionomers RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials) IT Myoglobins Proteins RL: MSC (Miscellaneous) (azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials) TΤ Oxides (inorganic), uses RL: TEM (Technical or engineered material use); USES (Uses) (colloidal dispersions, primers; azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials) Polyamines ΙT RL: RCT (Reactant); RACT (Reactant or reagent) (crosslinking agents; azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials) IT Polycarbonates, uses RL: TEM (Technical or engineered material use); USES (Uses) (films, substrate; azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials) Polyester fibers, uses IT Polyesters, uses Polypropene fibers, uses RL: TEM (Technical or engineered material use); USES (Uses)

(nonwovens, substrate; azlactone-functional reactive hydrophilic

coatings and hydrogels and articles comprising coated substrates useful

for immobilization of biol. materials) Amines, reactions ΙT RL: RCT (Reactant); RACT (Reactant or reagent) (polyamines, nonpolymeric, primary, crosslinking agents; azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials) IT Crosslinking agents (primary-secondary polyamines and polyether-polyamines; azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials) Albumins, miscellaneous ΙT RL: MSC (Miscellaneous) (serum, bovine; azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials) ΙT Glass, uses RL: TEM (Technical or engineered material use); USES (Uses) (slides, substrate; azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials) ΙT Nonwoven fabrics (substrate; azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials) IT 79-06-1DP, Acrylamide, derivs., polymers with azlactone-derived monomers 79-10-7DP, Acrylic acid, hydroxyalkyl esters, polymers with azlactone-derived monomers 79-39-0DP, Methacrylamide, derivs., polymers with azlactone-derived monomers 79-41-4DP, Methacrylic acid, polymers with azlactone-derived monomers 97-65-4DP, Itaconic acid, polymers with azlactone-derived monomers 100-43-6DP, 4-Vinylpyridine, polymers with azlactone-derived monomers 100-69-6DP, 2-Vinylpyridine, polymers with azlactone-derived monomers 105-16-8DP, 2-Diethylaminoethyl methacrylate, polymers with azlactone-derived monomers 110-16-7DP, 110-17-8DP, Maleic acid, polymers with azlactone-derived monomers Fumaric acid, polymers with azlactone-derived monomers 1121-55-7DP, 3-Vinylpyridine, polymers with azlactone-derived monomers 1746-03-8DP, Vinylphosphonic acid, polymers with azlactone-derived monomers 2426-54-2DP, 2-Diethylaminoethyl acrylate, polymers with azlactone-derived 15214-89-8DP, 2-Acrylamido-2-methyl-1-propanesulfonic acid, polymers with azlactone-derived monomers 18526-07-3DP, 3-Dimethylaminopropyl acrylate, polymers with azlactone-derived monomers 20602-77-1DP, 3-Dimethylaminopropyl methacrylate, polymers with 26914-43-2DP, Styrenesulfonic acid, polymers azlactone-derived monomers 36885-49-1DP, polymers with with azlactone-derived monomers azlactone-derived monomers 45021-77-0DP, (3-Acrylamidopropyl) trimethylammonium chloride, polymers with azlactone-derived monomers 87328-05-0DP, reaction products with crosslinked dimethylacrylamide-vinyldimethylazlactone copolymer 477273-94-2P 477273-95-3P 477273-96-4P 477273-97-5P 477273-98-6P RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials) 7631-86-9, Silica, uses IT

RL: TEM (Technical or engineered material use); USES (Uses)

(beads, substrate; azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials)

- 75-21-8D, Ethylene oxide, polymer, amino-terminated 75-56-9D, Propylene oxide, polymer, amino-terminated 109-76-2, 1,3-Propanediamine 124-09-4, 1,6-Hexanediamine, reactions 616-29-5, 1,3-Diamino-2-hydroxypropane 919-30-2, 3-Aminopropyltriethoxysilane 929-59-9 1760-24-3, N-[3-(Trimethoxysilyl)propyl]ethylenediamine 4097-89-6, Tris(2-aminoethyl)amine 4246-51-9, 4,7,10-Trioxa-1,13-tridecanediamine 9046-10-0 24991-53-5
 - RL: RCT (Reactant); RACT (Reactant or reagent)
 (crosslinking agent; azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials)
- IT 24968-12-5, Polybutylene terephthalate 25038-59-9, Polyethylene terephthalate, uses 25085-53-4, Isotactic polypropylene 26062-94-2, Polybutylene terephthalate
 - RL: TEM (Technical or engineered material use); USES (Uses) (nonwovens, substrate; azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials)
- 9002-85-1, Polyvinylidene chloride
 RL: TEM (Technical or engineered material use); USES (Uses)

(primer; azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials)

- IT 9002-98-6
 - RL: TEM (Technical or engineered material use); USES (Uses) (primers; azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials)
- IT 67-63-0, Isopropanol, uses 78-93-3, Methyl ethyl ketone, uses
 RL: NUU (Other use, unclassified); USES (Uses)
 (solvent; azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials)
- RE.CNT 6 THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS RECORD
- (1) 3m Innovative Properties Co; WO 9953319 A 1999 HCAPLUS
- (2) Chabrecek, P; WO 9828026 A 1998 HCAPLUS
- (3) Hubner, K; US 3583950 A 1971 HCAPLUS
- (4) Minnesota Mining & Mfg; WO 8807062 A 1988 HCAPLUS
- (5) Minnesota Mining & Mfg; WO 0026725 A 2000 HCAPLUS
- (6) Roehm & Haas Gmbh; DE 1936155 A 1971 HCAPLUS
- TT 79-10-7DP, Acrylic acid, hydroxyalkyl esters, polymers with azlactone-derived monomers 79-41-4DP, Methacrylic acid, polymers with azlactone-derived monomers
 - RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
 - (azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials)
- RN 79-10-7 HCAPLUS
- CN 2-Propenoic acid (9CI) (CA INDEX NAME)

0 HO- C- CH CH2

RN 79-41-4 HCAPLUS

2-Propenoic acid, 2-methyl- (9CI) (CA INDEX NAME) CN

CH₂ -с-со2н Me-

75-21-8D, Ethylene oxide, polymer, amino-terminated

RL: RCT (Reactant); RACT (Reactant or reagent) (crosslinking agent; azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials)

75-21-8 HCAPLUS RN

Oxirane (9CI) (CA INDEX NAME) CN

IT9002-98-6

> RL: TEM (Technical or engineered material use); USES (Uses) (primers; azlactone-functional reactive hydrophilic coatings and hydrogels and articles comprising coated substrates useful for immobilization of biol. materials)

9002-98-6 HCAPLUS RN

Aziridine, homopolymer (9CI) (CA INDEX NAME) CN

CM

CRN 151-56-4 CMF C2 H5 N

L23 ANSWER 2 OF 7 HCAPLUS COPYRIGHT 2003 ACS on STN

2002:521811 HCAPLUS ΑN

137:79391 DN

applicants Manufacture of polycarboxylic acid type copolymer and their use for cement TΤ additives

Yuasa, Tsutomu; Nishikawa, Tomotaka; Sakamoto, Noboru; Hirata, Tsuyoshi; IN Izukashi, Hiroko; Ueta, Tomiyasu; Tanaka, Hiromichi; Onda, Yoshiyuki; Uno,

PA Nippon Shokubai Co., Ltd., Japan

PCT Int. Appl., 114 pp.

CODEN: PIXXD2 DT Patent Japanese LΑ IC ICM C08F290-06 ICS C04B024-26; C04B028-02 35-4 (Chemistry of Synthetic High Polymers) CC Section cross-reference(s): 38, 58 FAN.CNT 1 APPLICATION NO. DATE PATENT NO. KIND DATE ____ WO 2001-JP11437 20011226 20020711 WO 2002053611 A1 PΙ W: CN, KR, SG RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR US 2001-25588 20011226-US 2002193547 20021219 Α1 JP 2003128738 JP 2001-395248 20011226 A2 20030508 Α 20001227 PRAI JP 2000-399466 20010817 Α JP 2001-248276 20010718 Α JP 2001-217582 A polycarboxylic acid type copolymer is produced by copolymg. a monomer AΒ component comprising a polyalkyleneimine type or a polyoxyalkylene type unsatd. monomer and an unsatd. carboxylic acid type monomer. The polycarboxylic acid type copolymer allows the production of a cement composition having improved water reducing property and workability and also yields a hardened product having excellent strength and durability, and thus can be suitably used in producing an ultrahigh strength concrete. Stirring ethoxylated polyethyleneimine with glycidyl methacrylate and polymerization with methoxy polyethylene glycol monomethacrylate, and methacrylic acid in water using 3-mercaptopropionic acid chain transfer agent and NH4S2O8 initiator gave a polycarboxylic acid copolymer additive with weight average mol. weight 17,400. polycarboxylic acid copolymer cement additive; methacrylic acid copolymer STcement additive; ethoxylated polyethyleneimine glycidyl methacrylate adduct; methoxy polyethylene glycol monomethacrylate copolymer Polyoxyalkylenes, preparation TT RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (acrylic; manufacture of polycarboxylic acid type copolymers and their use for cement additives) IT Cement Mortar Polyelectrolytes (manufacture of polycarboxylic acid type copolymers and their use for cement additives) IT Polyoxyalkylenes, preparation RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (polyamine-, acrylic; manufacture of polycarboxylic acid type copolymers and their use for cement additives) ΙT Polyamines RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (polyoxyalkylene-, acrylic; manufacture of polycarboxylic acid type copolymers and their use for cement additives) 75-21-8DP, Ethylene oxide, reaction product with IT

polyethyleneimine, glycidyl methacrylate, and unsatd. carboxylic acids 79-10-7DP, Acrylic acid, reaction product with methoxy polyethylene glycol monomethacrylate and ethoxylated polyethyleneimine methacrylate 79-41-4DP, Methacrylic acid, reaction product with methoxy polyethylene glycol monomethacrylate and ethoxylated polyethyleneimine methacrylate 106-91-2DP, Glycidyl methacrylate, reaction product with ethoxylated polyethyleneimine and unsatd. carboxylic acids 106-92-3DP, Allyl glycidyl ether, reaction product with ethoxylated triethylenetetraamine and unsatd. carboxylic acids 760-93-0DP, Methacrylic acid anhydride, reaction product with ethoxylated polyethyleneimine and unsatd. carboxylic acids 9002-98-6DP, ethoxylated, ester with glycidyl methacrylate, copolymers with unsatd. carboxylic acids 26915-72-0DP, Methoxy polyethylene glycol monomethacrylate, reaction product with ethoxylated polyethyleneimine and unsatd. carboxylic acids 31694-55-0DP, Polyoxyethylene glycerol ether, reaction product with glycidyl methacrylate, and unsatd. carboxylic acids 37371-89-4DP, Ethoxylated triethylenetetramine, reaction product with glycidyl methacrylate, and unsatd. carboxylic acids 53694-15-8DP, Polyethylene glycol sorbitol ether, reaction product with glycidyl methacrylate, and unsatd. carboxylic acids 125121-38-2P RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (manufacture of polycarboxylic acid type copolymers and their use for cement additives) RE.CNT 10 THERE ARE 10 CITED REFERENCES AVAILABLE FOR THIS RECORD (1) Kao Corp; JP 07247150 A 1995 (2) Kao Corp; JP 2001146447 A 2001 HCAPLUS (3) Kao Corp; JP 2001146449 A 2001 HCAPLUS (4) Kao Corp; JP 2001220440 A 2001 HCAPLUS (5) Lion Corp; JP 710943 A 1995 (6) Nippon Shokubai Co Ltd; JP 11269239 A 1999 HCAPLUS (7) Nippon Shokubai Co Ltd; JP 2000159555 A 2000 HCAPLUS (8) Nippon Shokubai Co Ltd; JP 2000319054 A 2000 HCAPLUS (9) Nippon Shokubai Co Ltd; JP 200063164 A 2000 (10) Toagosei Co Ltd; JP 2000247706 A 2000 HCAPLUS 75-21-8DP, Ethylene oxide, reaction product with polyethyleneimine, glycidyl methacrylate, and unsatd. carboxylic acids 79-10-7DP, Acrylic acid, reaction product with methoxy polyethylene glycol monomethacrylate and ethoxylated polyethyleneimine methacrylate 79-41-4DP, Methacrylic acid, reaction product with methoxy polyethylene glycol monomethacrylate and ethoxylated polyethyleneimine methacrylate 106-91-2DP, Glycidyl methacrylate, reaction product with ethoxylated polyethyleneimine and unsatd. carboxylic acids 106-92-3DP, Allyl glycidyl ether, reaction product with ethoxylated triethylenetetraamine and unsatd. carboxylic acids 760-93-0DP, Methacrylic acid anhydride, reaction product with ethoxylated polyethyleneimine and unsatd. carboxylic acids 9002-98-6DP, ethoxylated, ester with glycidyl methacrylate, copolymers with unsatd. carboxylic acids 26915-72-0DP, Methoxy polyethylene glycol monomethacrylate, reaction product with ethoxylated polyethyleneimine and unsatd. carboxylic acids 31694-55-ODP, Polyoxyethylene glycerol ether, reaction product with glycidyl methacrylate, and unsatd. carboxylic acids

37371-89-4DP, Ethoxylated triethylenetetramine, reaction product

with glycidyl methacrylate, and unsatd. carboxylic acids

> 53694-15-8DP, Polyethylene glycol sorbitol ether, reaction product with glycidyl methacrylate, and unsatd. carboxylic acids

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(manufacture of polycarboxylic acid type copolymers and their use for cement

75-21-8 HCAPLUS RN

Oxirane (9CI) (CA INDEX NAME) CN

RN 79-10-7 HCAPLUS 2-Propenoic acid (9CI) (CA INDEX NAME)

HO-C-CH=CH2

RN 79-41-4 HCAPLUS 2-Propenoic acid, 2-methyl- (9CI) (CA INDEX NAME) CN

CH₂ Me-C-CO2H

106-91-2 HCAPLUS ŔŊ 2-Propenoic acid, 2-methyl-, oxiranylmethyl ester (9CI) (CA INDEX NAME) CN

CH₂-O-C-C-Me

RN 106-92-3 HCAPLUS CN Oxirane, [(2-propenyloxy)methyl]- (9CI) (CA INDEX NAME)

CH2-O-CH2-CH=CH2

RN 760-93-0 HCAPLUS 2-Propenoic acid, 2-methyl-, anhydride (9CI) (CA INDEX NAME)

$$^{
m H_2C}_{\parallel}$$
 0 0 CH2 $^{
m H_2}_{\parallel}$ $^{
m H_2}_{\parallel}$ $^{
m Me-}$ C- C- C- Me

RN 9002-98-6 HCAPLUS

CN Aziridine, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 151-56-4 CMF C2 H5 N



RN 26915-72-0 HCAPLUS

CN Poly(oxy-1,2-ethanediyl), α -(2-methyl-1-oxo-2-propenyl)- ω -methoxy- (9CI) (CA INDEX NAME)

$$H_2C$$
 O H_2C O H_2C H_2C OMe

RN 31694-55-0 HCAPLUS

CN Poly(oxy-1,2-ethanediyl), $\alpha,\alpha',\alpha''-1,2,3-$ propanetriyltris[ω -hydroxy- (9CI) (CA INDEX NAME)

$$\begin{array}{c|c} & \text{CH}_2 & \hline & \text{O} - \text{CH}_2 - \text{CH}_2 \\ \hline & \text{HO} & \hline & \text{CH}_2 - \text{CH}_2 - \text{O} \\ \hline & \text{n} \end{array} \text{OH}$$

RN 37371-89-4 HCAPLUS

CN Poly(oxy-1,2-ethanediyl), α-hydro-ω-hydroxy-, ether with 3,6,9,12-tetrakis(2-hydroxyethyl)-3,6,9,12-tetraazatetradecane-1,14-diol (6:1) (9CI) (CA INDEX NAME)

PAGE 1-B

$$- CH_2 - CH_2 - OH$$
 $- CH_2 - CH_2 - OH$
 $- CH_2 - CH_2 - CH_2 - OH$
 $- CH_2 - CH_2 - OH$

RN 53694-15-8 HCAPLUS

CN Poly(oxy-1,2-ethanediyl), α -hydro- ω -hydroxy-, ether with D-glucitol (6:1) (9CI) (CA INDEX NAME)

HO
$$= \begin{array}{c|c} CH_2 - CH$$

RN 125121-38-2 HCAPLUS

CN 2-Propenoic acid, polymer with α -(3-methyl-3-butenyl)- ω -hydroxypoly(oxy-1,2-ethanediyl) (9CI) (CA INDEX NAME)

CM 1

CRN 110412-77-6

CMF (C2 H4 O)n C5 H10 O

CCI PMS

HO
$$CH_2 - CH_2 - O$$
 $CH_2 - CH_2 -$

CM 2

CRN 79-10-7 CMF C3 H4 O2

0

HO- C- CH CH2

L23 ANSWER 3 OF 7 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 1999:722766 HCAPLUS

DN 131:337562

TI Poly-perfluoroalkyl substituted polyamines as grease proofing agents for paper and foam stabilizers in aqueous fire-fighting foams

IN Jennings, John; Deisenroth, Ted; Haniff, Marlon

PA Ciba Specialty Chemicals Holding Inc., Switz.

SO Eur. Pat. Appl., 25 pp.

CODEN: EPXXDW

DT Patent

LA English

IC ICM C08G073-02

ICS D21H017-56; D21H021-16; A61C005-02; A62D001-02

CC 35-8 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 43, 50

FAN.CNT 1

LIM		TENT N	10.		KII	ND	DATE			AF	PLI	CATI	ON NO	o.	DATE			
PI		95532				2	1999			EF	19	99-8	1036	6	1999	0430		
	ΕP	95532			Α.	_	2000			an.	an.	T.M.		T 17	NIT	C E	MC	DШ
		R:							FR,	GB,	GR,	IT,	тт,	ъυ,	NL,	SE,	MC,	PI,
			ΙE,	SI,	LT,	L∨,	FI,	RO										
	US	61562	222		Α		2000	1205		US	19:	99-2	3425:	1	1999	0120		
	AU	99269	81		A.	1	1999	1118		ΑU	19	99-2	6981		1999	0506		
	AU	74317	70		B	2	2002	0117										
	BR	99020	062		А		2000	0411		BF	19	99-2	062		1999	0506		
	CN	12366	554		Α		1999	1201		CN	199	99-1	0637	0	1999	0510		
	JP	20000	266	01	' A.	2	2000	0125		JF	19	99-1	2800	8	1999	0510		
	US	63656	576		В:	1	2002	0402		US	20	00-6	87972	2	2000	1016		
PRAI	US	1998-	-848	15P	Р		1998	0508										
	US	1999-	-2342	251	A.	3	1999	0120										

- 10/28/03 Page 13 PEZZUTO 10/025588 Title polyamines, such as perfluoroalkyl-allyloxy- and AB perfluoroalkyl-iodopropyloxy-substituted polyaminoacids or poly-RF-fluoroallyl-substituted polyaminoacids (RF = monovalent perfluorinated alkyl or alkenyl, straight or branched organic radical having 4-20 fully fluorinated carbon atoms), comprise units of perfluoroalkyl-substituted amino groups, hydrophilically substituted amino or amido groups, and substituted amino or amido group, and are useful to provide oil repellency to paper and as foam stabilizers in alc. resistant-aqueous fire-fighting foam compns. Thus, Epomin SP 012 was reacted with allyl glycidyl ether to give a poly-(N-2-hydroxy-4-oxa-6,7-ene heptyl)polyethylenimine, which was further reacted with chloroacetic acid sodium salt, and Zonyl TELA-N perfluoroalkyl iodide, sodium metabisulfite, and 2,2'-azobisisobutyronitrile were added to give a poly-N-2-hydroxy-4oxa-[6,7-ene and 6-iodo]-7-RF-heptyl-N-carboxymethylene poly(ethylenimine) having good foam stabilizing effect on hot 2-propanol. perfluoroalkyl substituted polyamine oilproofing agent paper; aq firefighting foam stabilizer; Epomin allyl glycidyl ether reaction product; polyethylenimine chloroacetic acid reaction product; Zonyl perfluoroalkyl iodide polyethylenimine reaction product; azobisisobutyronitrile perfluoroalkyl polyamino acid prepn ITFoams Foams (fire-extinguishing; preparation of poly-perfluoroalkyl substituted polyamines as oilproofing agents for paper and foam stabilizers in aqueous fire-fighting foams) Fire extinguishers IT Fire extinguishers Stabilizing agents (foams; preparation of poly-perfluoroalkyl substituted polyamines as oilproofing agents for paper and foam stabilizers in aqueous fire-fighting foams)
- IT Sizes (agents)

(paper; preparation of poly-perfluoroalkyl substituted polyamines as oilproofing agents for paper and foam stabilizers in aqueous fire-fighting foams)

IT Oilproofing agents

Paper

(preparation of poly-perfluoroalkyl substituted polyamines as oilproofing agents for paper and foam stabilizers in aqueous fire-fighting foams)

IT Polyamines

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(reaction products; preparation of poly-perfluoroalkyl substituted polyamines as oilproofing agents for paper and foam stabilizers in aqueous fire-fighting foams)

IT 9002-98-6DP, Lupasol PR 8515, reaction products with allyl glycidyl ether, perfluoroalkyl iodides, and acids, amides, and/or glycidyl compds.

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(Lupasol PR 8515; preparation of poly-perfluoroalkyl substituted polyamines as oilproofing agents for paper and foam stabilizers in aqueous fire-fighting foams)

IT 78-67-1, 2,2'-Azobisisobutyronitrile 13472-08-7, Vazo 67
RL: CAT (Catalyst use); USES (Uses)

(free radical initiator; preparation of poly-perfluoroalkyl substituted polyamines as oilproofing agents for paper and foam stabilizers in aqueous fire-fighting foams)

79-07-2DP, 2-Chloroacetamide, reaction products with polyethylenimines and IT perfluoroalkyl iodides 79-10-7DP, Acrylic acid, reaction products with polyethylenimines and perfluoroalkyl iodides 106-92-3DP, Allyl glycidyl ether, reaction products with polyethylenimines and perfluoroalkyl iodides 126-83-0DP, reaction products with polyethylenimines and perfluoroalkyl iodides 355-43-1DP, Perfluorohexyl iodide, reaction products with polyethylenimines 556-52-5DP, Glycidol, reaction products with polyethylenimines and perfluoroalkyl iodides 3033-77-0DP, Quab 151, reaction products with polyethylenimines and perfluoroalkyl iodides 3039-83-6DP, Vinyl sulfonic acid sodium salt, reaction products with polyethylenimines and 3926-62-3DP, Chloroacetic acid sodium salt, perfluoroalkyl iodides reaction products with polyethylenimines and perfluoroalkyl iodides 7785-84-4DP, Sodium trimetaphosphate, reaction products with polyethylenimines and perfluoroalkyl iodides 250151-78-1DP, Zonyl TELA-N, reaction products with polyethylenimines 250151-80-5DP, Zonyl TELA-L, reaction products with polyethylenimines 250151-83-8DP, Zonyl TELB-L, reaction products with polyethylenimines RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (preparation of poly-perfluoroalkyl substituted polyamines as oilproofing agents for paper and foam stabilizers in aqueous fire-fighting foams) 9002-98-6DP, Lupasol PR 8515, reaction products with allyl ITglycidyl ether, perfluoroalkyl iodides, and acids, amides, and/or glycidyl compds. RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (Lupasol PR 8515; preparation of poly-perfluoroalkyl substituted polyamines as oilproofing agents for paper and foam stabilizers in aqueous fire-fighting foams) 9002-98-6 HCAPLUS RNAziridine, homopolymer (9CI) (CA INDEX NAME) CN CM CRN 151-56-4 CMF C2 H5 N



79-10-7DP, Acrylic acid, reaction products with polyethylenimines and perfluoroalkyl iodides 106-92-3DP, Allyl glycidyl ether, reaction products with polyethylenimines and perfluoroalkyl iodides RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(preparation of poly-perfluoroalkyl substituted polyamines as oilproofing agents for paper and foam stabilizers in aqueous fire-fighting foams)
RN 79-10-7 HCAPLUS
CN 2-Propenoic acid (9CI) (CA INDEX NAME)

RN 106-92-3 HCAPLUS

Oxirane, [(2-propenyloxy)methyl]- (9CI) (CA INDEX NAME) CN

L23 ANSWER 4 OF 7 HCAPLUS COPYRIGHT 2003 ACS on STN

1995:444025 HCAPLUS

DN 122:201289

TILight-sensitive recording material.

Kingma, Arend Jouke; Bronstert, Bernd; Scherr, Guenter; Steuerle, Ulrich ΙN PΑ SO

Eur. Pat. Appl., 14 pp. CODEN: EPXXDW

DT Patent

LΑ German

IC ICM G03F007-027

74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other CC

FAN.	CNT 1	ocesses)	1, and inotographic and
PI	PATENT NO. EP 599068	KIND DATE	APPLICATION NO. DATE
	EP 599068	A2 19940601 A3 19940824 ES, FR, GB, IT	EP 1993-117314 19931026
PRAI GI	JP 06202330 DE 1992-4239661	A1 19940601 A2 19940722 19921126	DE 1992-4239661 19921126 JP 1993-283176 19931112

$$\begin{bmatrix} c_n H_{2n} - N \\ R^1 \end{bmatrix}_p \begin{bmatrix} c_m H_{2m} - N \\ R^2 \end{bmatrix}_q$$

$$\begin{array}{c|c}
 & R^{X} & R^{X} \\
 & \downarrow \\
 & C_{n}H_{2n} - N \\
 & \downarrow \\
 & R^{Y}
\end{array}$$

$$\begin{array}{c|c}
 & R^{X} \\
 & \downarrow \\
 & \downarrow \\
 & R^{Y}
\end{array}$$

$$\begin{array}{c|c}
 & R^{X} \\
 & \downarrow \\
 & \downarrow \\
 & R^{Y}
\end{array}$$

$$\begin{array}{c|c}
 & R^{X} \\
 & \downarrow \\
 & \downarrow \\
 & R^{Y}
\end{array}$$

$$\begin{array}{c|c}
 & R^{X} \\
 & \downarrow \\
 & \downarrow \\
 & R^{Y}
\end{array}$$

The title material comprises ≥ 1 polymer binder, ≥ 1 ΑB polymerizable ethylenically unsatd.. organic compound from polyfunctional

ΙI

acrylate, a photoinitiator, and a softening agent from a polyalkylene imine derivative I or II or their mixture [n, m = 2-8; p, q = 10-10,000; R1] and R2 = R3CO; R3 = ethylenically unsatd. hydrocarbon; Rx = H, aliphatic, aromatic group; Ry = hydroxyalkyl, hydroxyoxaalkyl, poly(hydroxy)oxaalkyl, Rx, II; Y = RxCO2, monovalent anionic group.]. The composition is useful for production of relief printing plates. ST photosensitive compn printing plate softening agent; polyalkylene imine softening agent Linseed oil IT RL: DEV (Device component use); MOA (Modifier or additive use); USES (reaction product with polyethyleneimine and glycidyl methacrylate; photosensitive composition for relief printing plate) IT Printing plates (relief, photosensitive composition containing polyalkylene imine) 64-19-7D, Acetic acid, reaction product with polyethyleneimine IΤ Methyl iodide, reaction product with polyethyleneimine and ethylene oxide 75-21-8D, Ethylene oxide, reaction product with polyethyleneimine 75-56-9D, Propylene oxide, reaction product with polyethyleneimine 79-41-4D, Methacrylic acid, reaction product with polyethyleneimine and acetic acid 106-91-2D, Glycidyl methacrylate, reaction product with polyethyleneimine and linseed oil 7647-01-0D, Hydrochloric acid, reaction product with polyethyleneimine and ethylene oxide 9002-98-6D, modified 10563-26-5D, reaction product with dichloroethylene and acetic acid 25323-30-2D, Dichloroethylene, reaction product with ethylenedipropylenetetramine and 29722-29-0D, Butane diol monoacrylate, reaction product with acetic acid polyethyleneimine and methacrylic acid RL: DEV (Device component use); MOA (Modifier or additive use); USES (Uses) (photosensitive composition for relief printing plate) 75-21-8D, Ethylene oxide, reaction product with polyethyleneimine IΤ 79-41-4D, Methacrylic acid, reaction product with polyethyleneimine and acetic acid 106-91-2D, Glycidyl methacrylate, reaction product with polyethyleneimine and linseed oil 9002-98-6D, modified RL: DEV (Device component use); MOA (Modifier or additive use); USES (photosensitive composition for relief printing plate) RN 75-21-8 HCAPLUS CN Oxirane (9CI) (CA INDEX NAME)



RN 79-41-4 HCAPLUS CN 2-Propenoic acid, 2-methyl- (9CI) (CA INDEX NAME)

RN 106-91-2 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, oxiranylmethyl ester (9CI) (CA INDEX NAME)

RN 9002-98-6 HCAPLUS

CN Aziridine, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 151-56-4 CMF C2 H5 N



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L23 ANSWER 5 OF 7 HCAPLUS COPYRIGHT 2003 ACS on STN
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AN 1995:302816 HCAPLUS

DN 122:56836

TI Water-thinned resin dispersions with good storability and manufacture thereof

IN Uchida, Masaya; Nikashiwa, Toshiki; Minami, Kenji; Izumibayashi, Masuji

PA Nippon Catalytic Chem Ind, Japan

SO Jpn. Kokai Tokkyo Koho, 30 pp. CODEN: JKXXAF

DT Patent

Di Facenc

LA Japanese IC ICM C08F(

CC 35-4 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 42

FAN.CNT 1

L MM .	CNII				
	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
ΡI	JP 06192307	A2	19940712	JP 1993-211984	19930826
	JP 2698536	B2	19980119		
	JP 10114847	A2	19980506	JP 1997-197473	19930826
PRAI	JP 1992-229588		19920828		
	JP 1993-211984		19930826		

AB The title dispersions forming films with excellent appearances, water repellency, lubricity, etc. are prepared by radical polymerization of a monomer component containing C9-30 aliphatic hydrocarbyl group, dispersed as droplets of

volume-average diameter $\leq 1~\mu m$ in water with $\geq 3~\mu m$ -diameter droplet content ≤ 5 volume%. Polymerization of Me methacrylate, styrene, 2-ethylhexyl acrylate, glycidyl methacrylate, stearyl acrylate in water in the presence of AIBN, dodecyl mercaptan-terminated polyacrylic acid ammonium salt gave a 39.8%-solids dispersion with average particle diameter

0.35

```
acrylic polymer aq dispersion
ST
     Epoxides
IT
     RL: MOA (Modifier or additive use); USES (Uses)
        (C10-12-alkyl, reaction products with polyethylenimine; water-thinned
        resin dispersions with good storability)
     Fatty acids, preparation
IT
     RL: IMF (Industrial manufacture); PREP (Preparation)
        (soya, glycidyl methacrylate adducts, polymers; manufacture as stable
aqueous
        dispersions)
     Coating materials
IT
        (water-thinned, long-chain aliphatic group-containing polymers with good
        storability)
     106-92-3D, Allyl glycidyl ether, reaction products with olefin
IT
     epoxides and polyethylenimine 9002-98-6D, Epomin SP-006,
     reaction products with olefin epoxides 9003-01-4D, Poly(acrylic acid),
     dodecyl mercaptan-terminated, ammonium salt
                                                  38639-64-4D, Acrylic
     acid-2-hydroxyethyl acrylate copolymer, dodecyl mercaptan-terminated,
     ammonium salt
     RL: MOA (Modifier or additive use); USES (Uses)
        (dispersant; water-thinned resin dispersions with good storability)
     57-11-4DP, Stearic acid, reaction products with isopropenyloxazoline,
IT
     polymers 106-91-2DP, Glycidyl methacrylate, soya fatty acid
                        6498-82-4DP, soya fatty acid adducts, polymers
     adducts, polymers
     25703-24-6P
                 25986-77-0P, Poly(stearyl acrylate)
                                                          53640-78-1P
     63623-16-5P, Poly(nonyl acrylate)
                                         160311-26-2P
                                                         160311-27-3P
     RL: IMF (Industrial manufacture); PREP (Preparation)
        (manufacture as stable aqueous dispersions)
     577-11-7, Sodium dioctyl sulfosuccinate
     RL: MOA (Modifier or additive use); USES (Uses)
        (water-thinned resin dispersions with good storability)
ΙT
     106-92-3D, Allyl glycidyl ether, reaction products with olefin
     epoxides and polyethylenimine 9002-98-6D, Epomin SP-006,
     reaction products with olefin epoxides
     RL: MOA (Modifier or additive use); USES (Uses)
        (dispersant; water-thinned resin dispersions with good storability)
RN
     106-92-3 HCAPLUS
     Oxirane, [(2-propenyloxy)methyl]- (9CI) (CA INDEX NAME)
CN
     CH_2 - O - CH_2 - CH == CH_2
     9002-98-6 HCAPLUS
RN
    Aziridine, homopolymer (9CI) (CA INDEX NAME)
CN
     CM
          1
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CRN 151-56-4 CMF C2 H5 N

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H
N
/\
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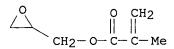
IT 106-91-2DP, Glycidyl methacrylate, soya fatty acid adducts,
polymers

RL: IMF (Industrial manufacture); PREP (Preparation)

(manufacture as stable aqueous dispersions)

RN 106-91-2 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, oxiranylmethyl ester (9CI) (CA INDEX NAME)



L23 ANSWER 6 OF 7 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 1993:540422 HCAPLUS

DN 119:140422

TI Tough polyamine-acrylate epoxy resin compositions

IN Yoshida, Masatoshi; Minami, Kenji; Namura, Ichiro; Izumibayashi, Masuji

PA Nippon Shokubai Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 10 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C08L063-00

ICS C08F002-28; C08F020-18; C08L033-10

CC 37-6 (Plastics Manufacture and Processing)

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
ΡI	JP 04370139	A2	19921222	JP 1991-174347	19910618
	JP 07015041	B4	19950222		
PRAI	JP 1991-174347		19910618		

AB The title compns., with good water resistance and heat resistance, are comprised of reaction products of reactive emulsifiers of modified polyamines and (meth)acrylates, and epoxy resins. Thus, an adhesive composition from Epikote 828 100, HV 953U (amidoamine hardener), and an emulsion prepared by the reaction of Epomin SP 006 and AOE-X 24 to give a modified polyamine and emulsion polymerization with Et acrylate, Me methacrylate,

and glycidyl methacrylate, showed shear strength 258 kg/cm2 initially and 175 kg/cm2 after 24 h in water at 80°.

ST emulsion polymn epoxy acrylate tough; polyoxyalkylene polyamine epoxy acrylate compn; emulsifier polymn acrylate polyamine tough

IT Emulsifying agents

(acrylate-modified polyamines, epoxy resins containing, tough)

IT Epoxy resins, uses

RL: USES (Uses)

(modified polyamine-containing, tough)

IT Polyamines

RL: USES (Uses)

(acrylic, emulsions, epoxy resins containing, tough)

TT Epoxides RL: USES (Uses) (alkenyl, reaction products with polyethylenimine, Et acrylate, Me methacrylate and glycidyl methacrylate, emulsions, eposy resin compns. containing) ΙT Amines, compounds RL: USES (Uses) (poly-, reaction products, with Bu acrylate, styrene and acrylonitrile, emulsions, epoxy resin compns. containing, tough and water-resistant) ΙT RL: USES (Uses) (reaction products, with allyl glycidyl ether, polyamines and acrylic monomers, emulsions, epoxy resin compns. containing) ΙT 25068-38-6, Epikote 828 RL: USES (Uses) (acrylate polyamine-containing, tough and water-resistant) 80-62-6D, reaction products with α -olefin epoxides, ΙT polyethylenimine, Et acrylate and glycidyl methacrylate 100-42-5D. reaction products with Bu acrylate, modified polyamines and styrene 106-91-2D, reaction products with α -olefin epoxides, polyethylenimine, Et acrylate and Me methacrylate 106-92-3D, reaction products with epoxides, polyamines and acrylic monomers 107-13-1D, 2-Propenenitrile, reaction products with Bu acrylate, modified polyamines and styrene 140-88-5D, reaction products with α -olefin epoxides, polyethylenimine, Me methacrylate and glycidyl methacrylate 141-32-2D, reaction products with acrylonitrile, modified polyamines and 2274-11-5D, polymers with modified polyamines and acrylic monomers 9002-98-6D, reaction products with $\alpha\text{-olefin}$ epoxides, Et acrylate, Me methacrylate and glycidyl methacrylate RL: USES (Uses) (emulsions, epoxy resin compns. containing, tough and water-resistant) IT 2426-08-6, Butyl glycidyl ether RL: USES (Uses) (epoxy resin compns. containing acrylate-modified polyamines and, tough) TΤ 39421-53-9, Araldite HV 953U RL: USES (Uses) (hardeners, epoxy resin compns. containing acrylate polyamines and, tough) IT 101-77-9 RL: USES (Uses) (hardeners, epoxy resin compns. containing acrylate-modified polyamines and, tough) 106-91-2D, reaction products with α -olefin epoxides, polyethylenimine, Et acrylate and Me methacrylate 106-92-3D, reaction products with epoxides, polyamines and acrylic monomers 9002-98-6D, reaction products with α -olefin epoxides, Et acrylate, Me methacrylate and glycidyl methacrylate RL: USES (Uses) (emulsions, epoxy resin compns. containing, tough and water-resistant) RN 106-91-2 HCAPLUS 2-Propenoic acid, 2-methyl-, oxiranylmethyl ester (9CI) (CA INDEX NAME) CN

RN 106-92-3 HCAPLUS

CN Oxirane, [(2-propenyloxy)methyl]- (9CI) (CA INDEX NAME)

СН2-0-СН2-СН=СН2

RN 9002-98-6 HCAPLUS

CN Aziridine, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 151-56-4 CMF C2 H5 N



L23 ANSWER 7 OF 7 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 1986:191928 HCAPLUS

DN 104:191928

TI Concrete admixture and its use

IN Teraji, Tsutomu; Kawada, Kazushige; Takeuchi, Toru; Sugita, Shinichi; Adachi, Toshikazu; Yagi, Hideo

PA Fujisawa Pharmaceutical Co., Ltd., Japan

SO Eur. Pat. Appl., 33 pp. CODEN: EPXXDW

DT Patent

LA English

IC ICM C04B024-28

CC 58-2 (Cement, Concrete, and Related Building Materials)
Section cross-reference(s): 38

FAN.CNT 1

	PAT	TENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI		174644 174644	A1 B1	19860319 19881117	EP 1985-111463	19850911
		R: AT, BE,	CH, DE	, FR, GB, IT	, LI, NL, SE	
	CN	85108262	А	19870325	CN 1985-108262	19850907
	ΑU	8547171	A1	19860320	AU 1985-47171	19850909
	ΑU	579887	B2	19881215		
	zA	8506897	Α	19860430	ZA 1985-6897	19850909
	CA	1247152	A1	19881220	CA 1985-490286	19850910
	DK	8504142	Α	19860313	DK 1985-4142	19850911
	JP	61083663	A2	19860428	JP 1985-202180	19850911
	JΡ	06099169	B4	19941207		
	BR	8504469	Α	19860715	BR 1985-4469	19850911
	AT	38657	E	19881215	AT 1985-111463	19850911
	US	4650522	A	19870317	US 1985-775390	19850912
PRAI	GB	1984-23054		19840912		
	ΕP	1985-111463		19850911		

AB Concrete admixts., improving the flow properties of the cement mix, comprise polyethyleneimine derivs. with acidic group, acidic

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PEZZUTO 10/025588
                     10/28/03
                                Page 22
     group-substituted alkyl and/or acyl groups or the substituted alkyl groups
     and carbamoylalkyl and hydroxy-substituted alkyl groups. Thus, 25% mol.
     equivalent maleic acid was added dropwise to a solution of polyethyleneimine
(average
    mol. weight 10,000) 43.0 g in water 130 mL at 25-40° with stirring and
     stirred 2 h at room temperature, NaOH 20.0 g in 100 mL water was added with
     stirring and the mixture stirred 18 h under reflux to give an aqueous solution
of
    poly(disodium 1,2-dicarboxylatoethyl)polyethyleneimine 83.0 g with acidic
     substitution 11.2%. Thus, mortars from portland cement 520, sand 1040,
    the admixt. 1.1 g, and water 364 mL had flow 209 and 172 mm in cone tests
     immediately and 60 min after mixing, resp., vs. 169 and 144 without the
     additive.
     acid group substituted polyethyleneimine concrete; carboxylic acid
ST
     reaction product polyethyleneimine
TΤ
     Concrete
    Mortar
        (polyethyleneimine reaction product admixts. in, for improved rheol.)
     75-21-8D, reaction products with acrylic acid or derivs. and
TΤ
                       75-56-9D, reaction products with acrylic acid or
     polyethyleneimine
     derivs. and polyethyleneimine
                                    79-06-1D, reaction products with
     polyethyleneimine 79-10-7D, reaction products with
     polyethyleneimine 79-11-8D, reaction products with polyethyleneimine and
     sodium hydroxide 79-41-4D, reaction products with
     polyethyleneimine 85-44-9D, reaction products with polyethyleneimine and
     sodium hydroxide
                        97-65-4D, reaction products with polyethyleneimine
     107-13-1D, reaction products with polyethyleneimine 110-16-7D, reaction
    products with polyethyleneimine
                                      1305-62-0D, reaction products with
     acrylamide and polyethyleneimine 1310-73-2D, reaction products with
     carboxylic acid or derivs. and polyethyleneimine 9002-98-6D,
     reaction products with carboxylic acids
                                             40618-18-6D, reaction products
     with polyethyleneimine
     RL: USES (Uses)
        (in concrete mix, for improved rheol.)
ΤТ
     75-21-8D, reaction products with acrylic acid or derivs. and
     polyethyleneimine 79-10-7D, reaction products with
     polyethyleneimine 79-41-4D, reaction products with
     polyethyleneimine 9002-98-6D, reaction products with carboxylic
     acids
     RL: USES (Uses)
        (in concrete mix, for improved rheol.)
     75-21-8 HCAPLUS
RN
     Oxirane (9CI) (CA INDEX NAME)
CN
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RN 79-10-7 HCAPLUS CN 2-Propenoic acid (9CI) (CA INDEX NAME)

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PEZZUTO 10/025588
                     10/28/03 Page 23
RN
    79-41-4 HCAPLUS
     2-Propenoic acid, 2-methyl- (9CI) (CA INDEX NAME)
CN
   CH<sub>2</sub>
Me-C-CO2H
    9002-98-6 HCAPLUS
RN
    Aziridine, homopolymer (9CI) (CA INDEX NAME)
CN
     CM
          1
     CRN
         151-56-4
     CMF C2 H5 N
=> => D QUE
             12 SEA FILE=REGISTRY ABB=ON (106-91-2/BI OR 106-92-3/BI OR
L2
                125121-38-2/BI OR 26915-72-0/BI OR 31694-55-0/BI OR 37371-89-4/
                BI OR 53694-15-8/BI OR 75-21-8/BI OR 760-93-0/BI OR 79-10-7/BI
                OR 79-41-4/BI OR 9002-98-6/BI)
              2 SEA FILE=REGISTRY ABB=ON L2 AND 1-20/N
L5
             5 SEA FILE=REGISTRY ABB=ON L2 AND PROPENOIC
L6
             5 SEA FILE=REGISTRY ABB=ON L2 NOT (L5 OR L6)
L7
              2 SEA FILE=REGISTRY ABB=ON
                                         L7 AND PROPENYL
rs
              1 SEA FILE=REGISTRY ABB=ON
                                         L8 AND ALPHA
L9
              4 SEA FILE=REGISTRY ABB=ON
                                         L7 NOT L9
L10
              6 SEA FILE=REGISTRY ABB=ON L9 OR L6
L12
L13
          46474 SEA FILE=HCAPLUS ABB=ON L12
          20517 SEA FILE=HCAPLUS ABB=ON
                                         L10
L14
           8265 SEA FILE=HCAPLUS ABB=ON
L15
                                        L5
             14 SEA FILE=HCAPLUS ABB=ON L13 AND L14 AND L15
L16
L17
          11421 SEA FILE=HCAPLUS ABB=ON (L13 OR L14 OR L15)(L)(PREP OR IMF OR
                SPN)/RL
              4 SEA FILE=HCAPLUS ABB=ON L16 AND L17
L18
          23913 SEA FILE=HCAPLUS ABB=ON L12/D
L19
                                        L10/D
           3864 SEA FILE=HCAPLUS ABB=ON
L20
           2119 SEA FILE=HCAPLUS ABB=ON
                                        L5/D
L21
L22
              6 SEA FILE=HCAPLUS ABB=ON
                                        L19 AND L20 AND L21
L23
              7 SEA FILE=HCAPLUS ABB=ON L18 OR L22
                STR /
L24
       5
       0
```

NODE ATTRIBUTES:

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PEZZUTO 10/025588 10/28/03 Page 24
      I'S RC
NSPEC
                  AΤ
                       1
      IS RC
NSPEC
                  AT
                        2
                  AT
                       3
      IS RC
NSPEC
                        4
        IS RC
                  AT
NSPEC
DEFAULT MLEVEL IS ATOM
DEFAULT ECLEVEL IS LIMITED
GRAPH ATTRIBUTES:
RING(S) ARE ISOLATED OR EMBEDDED
NUMBER OF NODES IS
STEREO ATTRIBUTES: NONE
L25
                STR 2
N \rightarrow C \rightarrow C
1 2 3
NODE ATTRIBUTES:
NSPEC IS RC
                  ΑT
NSPEC
        IS RC
                  AΤ
NSPEC
        IS RC
                  AT
DEFAULT MLEVEL IS ATOM
DEFAULT ECLEVEL IS LIMITED
GRAPH ATTRIBUTES:
RING(S) ARE ISOLATED OR EMBEDDED
NUMBER OF NODES IS 3
STEREO ATTRIBUTES: NONE
                STR 3
L26
C \rightarrow C \rightarrow 0
1 2 3
NODE ATTRIBUTES:
NSPEC IS RC
                  AT
                       1
NSPEC
        IS RC
                  ΑT
                        2
                                               84,334 polymers
from shurtures
1 and 2 and 3
        IS RC
                  AT
NSPEC
DEFAULT MLEVEL IS ATOM
DEFAULT ECLEVEL IS LIMITED
GRAPH ATTRIBUTES:
RING(S) ARE ISOLATED OR EMBEDDED
NUMBER OF NODES IS 3
STEREO ATTRIBUTES: NONE
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          36791 SEA FILE=REGISTRY ABB=ON 46.150.18/RID AND L30
L33
          47543 SEA FILE=REGISTRY ABB=ON L30 NOT L33
L34
          43604 SEA FILE=REGISTRY ABB=ON L34 NOT 1-10/SI
L35
          41681 SEA FILE=REGISTRY ABB=ON L35 NOT TRIAZIN?
L36
L37
          40716 SEA FILE=REGISTRY ABB=ON L36 NOT MORPHOLIN?
L38
          36224 SEA FILE=REGISTRY ABB=ON L37 NOT (PYRROLI? OR AZEPIN?)
L39
                STR
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NODE ATTRIBUTES:

DEFAULT MLEVEL IS ATOM
DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RSPEC I

NUMBER OF NODES IS 3

STEREO ATTRIBUTES: NONE

L40

STR

 $\begin{array}{ccccc} \text{CH2-CH2-N----} & \text{CH2-CH2} \\ 1 & 2 & 3 & 4 & 5 \end{array}$

NODE ATTRIBUTES:

DEFAULT MLEVEL IS ATOM

DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RSPEC I

NUMBER OF NODES IS 5

STEREO ATTRIBUTES: NONE

L43 L44 L44 L45 L44 L45 L44 L45 L45 L46 L45 L46 L46 L47 L46 L47 L47 L47 L48 L48 L48 L48 L48 L49 L48 L49 L49 L49 L49 L49 L40	L42	13218	SEA	FILE=REGISTRY	SUB=L30	SSS FUL (L39 OR L40)
L45	L43	6332	SEA	FILE=REGISTRY	ABB=ON	L38 AND L42
L46	L44	5022	SEA	FILE=REGISTRY	ABB=ON	L43 NOT AMINIUM
L47 L48 L48 L48 L48 L49 L49 L49 L49 L50 L40 L40 L40 L40 L40 L40 L40 L40 L40 L4	L45					
L48 1353 SEA FILE=HCAPLUS ABB=ON L47(L)(PREP OR IMF OR SPN)/RL L49 16 SEA FILE=HCAPLUS ABB=ON L48 AND POLYCARBOXYLIC L50 1038 SEA FILE=HCAPLUS ABB=ON L48 AND COPOLYMER? L51 19 SEA FILE=HCAPLUS ABB=ON L48 AND CEMENT# L54 55 SEA FILE=HCAPLUS ABB=ON L50 AND POLYAMINES/IT L55 8 SEA FILE=HCAPLUS ABB=ON L54 AND POLYOXYALKYLENE?/IT L56 42 SEA FILE=HCAPLUS ABB=ON L49 OR L51 OR L55 L57 30 SEA FILE=HCAPLUS ABB=ON L50 AND POLYAMINE/IT L58 8 SEA FILE=HCAPLUS ABB=ON L50 AND POLYAMINE/IT L58 44 SEA FILE=HCAPLUS ABB=ON L57 AND POLYOXYALKYLENE?/IT L59 44 SEA FILE=HCAPLUS ABB=ON L56 OR L58	L46	4031	SEA	FILE=REGISTRY	ABB=ON	L45 NOT 1-30/F
L49 16 SEA FILE=HCAPLUS ABB=ON L48 AND POLYCARBOXYLIC L50 1038 SEA FILE=HCAPLUS ABB=ON L48 AND COPOLYMER? L51 19 SEA FILE=HCAPLUS ABB=ON L48 AND CEMENT# L54 55 SEA FILE=HCAPLUS ABB=ON L50 AND POLYAMINES/IT L55 8 SEA FILE=HCAPLUS ABB=ON L54 AND POLYOXYALKYLENE?/IT L56 42 SEA FILE=HCAPLUS ABB=ON L49 OR L51 OR L55 L57 30 SEA FILE=HCAPLUS ABB=ON L50 AND POLYAMINE/IT L58 8 SEA FILE=HCAPLUS ABB=ON L57 AND POLYOXYALKYLENE?/IT L59 44 SEA FILE=HCAPLUS ABB=ON L56 OR L58	L47	3323	SEA	FILE=HCAPLUS	ABB=ON	
L50	L48	1353	SEA	FILE=HCAPLUS	ABB=ON	L47(L)(PREP OR IMF OR SPN)/RL
L51 19 SEA FILE=HCAPLUS ABB=ON L48 AND CEMENT# L54 55 SEA FILE=HCAPLUS ABB=ON L50 AND POLYAMINES/IT L55 8 SEA FILE=HCAPLUS ABB=ON L54 AND POLYAMINES/IT L56 42 SEA FILE=HCAPLUS ABB=ON L49 OR L51 OR L55 L57 30 SEA FILE=HCAPLUS ABB=ON L50 AND POLYAMINE/IT L58 8 SEA FILE=HCAPLUS ABB=ON L57 AND POLYAMINE/IT L59 44 SEA FILE=HCAPLUS ABB=ON L56 OR L58	L49	16	SEA	FILE=HCAPLUS	ABB=ON	L48 AND POLYCARBOXYLIC
L54 55 SEA FILE=HCAPLUS ABB=ON L50 AND POLYAMINES/IT L55 8 SEA FILE=HCAPLUS ABB=ON L54 AND POLYOXYALKYLENE?/IT L56 42 SEA FILE=HCAPLUS ABB=ON L49 OR L51 OR L55 L57 30 SEA FILE=HCAPLUS ABB=ON L50 AND POLYAMINE/IT L58 8 SEA FILE=HCAPLUS ABB=ON L57 AND POLYOXYALKYLENE?/IT L59 44 SEA FILE=HCAPLUS ABB=ON L56 OR L58	L50	1038	SEA	FILE=HCAPLUS	ABB=ON	L48 AND COPOLYMER?
L55 8 SEA FILE=HCAPLUS ABB=ON L54 AND POLYOXYALKYLENE?/IT L56 42 SEA FILE=HCAPLUS ABB=ON L49 OR L51 OR L55 L57 30 SEA FILE=HCAPLUS ABB=ON L50 AND POLYAMINE/IT L58 8 SEA FILE=HCAPLUS ABB=ON L57 AND POLYOXYALKYLENE?/IT L59 44 SEA FILE=HCAPLUS ABB=ON L56 OR L58	L51	19	SEA	FILE=HCAPLUS	ABB=ON	L48 AND CEMENT#
L56 42 SEA FILE=HCAPLUS ABB=ON L49 OR L51 OR L55 L57 30 SEA FILE=HCAPLUS ABB=ON L50 AND POLYAMINE/IT L58 8 SEA FILE=HCAPLUS ABB=ON L57 AND POLYOXYALKYLENE?/IT L59 44 SEA FILE=HCAPLUS ABB=ON L56 OR L58	L54	55	SEA	FILE=HCAPLUS	ABB=ON	L50 AND POLYAMINES/IT
L57 30 SEA FILE=HCAPLUS ABB=ON L50 AND POLYAMINE/IT L58 8 SEA FILE=HCAPLUS ABB=ON L57 AND POLYOXYALKYLENE?/IT L59 44 SEA FILE=HCAPLUS ABB=ON L56 OR L58	L55	8	SEA	FILE=HCAPLUS	ABB=ON	L54 AND POLYOXYALKYLENE?/IT
L58 8 SEA FILE=HCAPLUS ABB=ON L57 AND POLYOXYALKYLENE?/IT L59 44 SEA FILE=HCAPLUS ABB=ON L56 OR L58	L56	42	SEA	FILE=HCAPLUS	ABB=ON	L49 OR L51 OR L55
L59 44 SEA FILE=HCAPLUS ABB=ON L56 OR L58	L57	30	SEA	FILE=HCAPLUS	ABB=ON	L50 AND POLYAMINE/IT
	L58	8	SEA	FILE=HCAPLUS	ABB=ON	L57 AND POLYOXYALKYLENE?/IT
	L59	44	SEA	FILE=HCAPLUS	ABB=ON	L56 OR L58
L60 44 SEA FILE=HCAPLUS ABB=ON L59 NOT L23	L60	. 44	SEA	FILE=HCAPLUS	ABB=ON	L59 NOT L23

=> D L60 1-44 BIB ABS HITIND HITSTR

L60 ANSWER 1 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 2003:673980 HCAPLUS

DN 139:198525

TI Acrylic polymer-based adhesive compositions for adhesive sheets

IN Okochi, Naoki; Ando, Masahiko

PA Nitto Denko Corp., Japan

SO Jpn. Kokai Tokkyo Koho, 9 pp.

```
CODEN: JKXXAF
DT
     Patent
LA
     Japanese
FAN.CNT 1
                      KIND DATE
                                           APPLICATION NO. DATE
     PATENT NO.
                                           _____
     JP 2003238922
                       A2
                            20030827
                                           JP 2002-44758
                                                             20020221
PΙ
PRAI JP 2002-44758
                            20020221
     The composition comprises (A) a hydroxyl group-free acrylic polymer, (B) an
     amine compound containing plural hydroxyl groups, and (C) a polyisocyanate
     compound, wherein preferably, the composition has gel fraction 10-70% after
dried
     or cured. The sheets such as labels and adhesive tapes are obtained by
     applying supports on ≥1 side with the adhesives. Thus, a composition
     comprising 100/5 Bu acrylate-acrylic acid copolymer 100,
     phenolic resin tackifier 20, xylene resin 30, EDP 1100
     (ethylenediamine-propylene oxide adduct) 1.5 and Coronate L
     (polyisocyanate) 4 parts was coated on both sides of a nonwoven fabric to
     give an adhesive tape showing gel fraction 19% and adhesion strength 15.2
     N/20 \text{ mm}.
IC
     ICM C09J133-00
     ICS C09J007-02; C09J175-04
CC
     38-3 (Plastics Fabrication and Uses)
IT
     Polyurethanes, uses
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (acrylic-polyamine-polyoxyalkylene-; acrylic
        polymer-based adhesive compns. for adhesive sheets)
IT
     Polyoxyalkylenes, uses
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (acrylic-polyamine-polyurethane-; acrylic polymer-based
        adhesive compns. for adhesive sheets)
TΨ
     Polyamines
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (acrylic-polyoxyalkylene-polyurethane-; acrylic polymer-based
        adhesive compns. for adhesive sheets)
IT
     585565-52-2P
     RL: IMF (Industrial manufacture); TEM (Technical or engineered
    material use); PREP (Preparation); USES (Uses)
        (acrylic polymer-based adhesive compns. for adhesive sheets)
IT
     585565-52-2P
     RL: IMF (Industrial manufacture); TEM (Technical or engineered
    material use); PREP (Preparation); USES (Uses)
        (acrylic polymer-based adhesive compns. for adhesive sheets)
RN
     585565-52-2 HCAPLUS
CN
     2-Propenoic acid, polymer with butyl 2-propenoate, Coronate L and
     \alpha, \alpha', \alpha'', \alpha''' - [1, 2-ethanediylbis[nitrilobis(methyl-
     2,1-ethanediyl)]]tetrakis[ω-hydroxypoly[oxy(methyl-1,2-ethanediyl)]]
     (9CI)
           (CA INDEX NAME)
     CM
          1
          51178-86-0
    CRN
          (C3 H6 O)n (C3 H6 O)n (C3 H6 O)n (C3 H6 O)n C14 H32 N2 O4
     CMF
     CCI
         IDS, PMS
```

PAGE 1-B

CM 2

CRN 39278-79-0

CMF Unspecified

CCI PMS, MAN

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

CM 3

CRN 141-32-2

CMF C7 H12 O2

CM 4

CRN 79-10-7 CMF C3 H4 O2

L60 ANSWER 2 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 2003:506596 HCAPLUS

DN 139:70148

TI Use of aqueous adhesives for bonding wood sheets

IN Fuessl, Ruediger; Meyer-Roscher, Bernd; Gerst, Matthias; Smink, Eduard

PA BASF AG, Germany

SO Ger. Offen., 20 pp.

CODEN: GWXXBX

DT Patent

LA German

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE

DE 10253498 A1 20030703 DE 2002-10253498 20021116

PRAI DE 2001-10160150 A1 20011207

AB Aqueous adhesives for bonding wood sheets contain dispersed particles of ≥1 polymer (A1) obtained by radical polymerization in the presence of polymers (A2) formed from 50-99.5% ≥1 ethylenically unsatd. mono-and(or) dicarboxylic acid, 0.5-50% ≥1 ethylenically unsatd. compound selected from esters of ethylenically unsatd. monocarboxylic acids and semiesters and diesters of ethylenically unsatd. diacids and ≥1 OH-containing amine, and ≤20% ≥1 other monomer. Optionally, the adhesives are based on an aqueous solution containing (A) a polymer prepared by radical

polymerization of 5-100% of an ethylenical unsatd. acid anhydride or an anhydride-formable ethylenically unsatd. diacid, and (B) an alkanolamine with ≥2 OH groups. A typical adhesive contained Acrodur A502/121 [aqueous emulsion prepared by polymerization of styrene 70, Me methacrylate 25, and

hydroxyethyl acrylate 5 parts in the presence of 100 parts polymer prepared from 56% acrylic acid, 24% maleic anhydride, and 20% Uniperol AC (amine emulsifier) and 9 parts triethanolamine] and 30% Epikote 828 (bisphenol A diglycidyl ether polymer) crosslinker.

IC ICM C09D005-12

ICS C09J133-02

CC 38-3 (Plastics Fabrication and Uses)

Section cross-reference(s): 43

IT Wood

(aqueous adhesives based on amine-containing **polycarboxylic** acid ester and crosslinkers for bonding wood sheets)

IT Adhesives

(water-thinned; aqueous adhesives based on amine-containing **polycarboxylic** acid ester and crosslinkers for bonding wood sheets)

IT 551960-34-0P, Acrylic acid-bisphenol A-epichlorohydrin-2-hydroxyethyl acrylate-maleic anhydride-methyl methacrylate-styrene copolymer triethanolamine ester

RL: IMF (Industrial manufacture); TEM (Technical or engineered material

use); PREP (Preparation); USES (Uses)
 (cured adhesive; aqueous adhesives based on amine-containing
 polycarboxylic acid ester and crosslinkers for bonding wood
 sheets)

IT 165670-41-7P

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(cured adhesive; crosslinkable aqueous amine-containing polymeric carboxylic acid adhesive compns. for bonding wood sheets)

IT 165670-41-7P

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(cured adhesive; crosslinkable aqueous amine-containing polymeric carboxylic acid adhesive compns. for bonding wood sheets)

RN 165670-41-7 HCAPLUS

CN 2-Butenedioic acid (2Z)-, polymer with 2,2',2''-nitrilotris[ethanol] and 2-propenoic acid (9CI) (CA INDEX NAME)

CM 1

CRN 110-16-7 CMF C4 H4 O4

Double bond geometry as shown.

CM 2

CRN 102-71-6 CMF C6 H15 N O3

$$_{\rm CH_2-CH_2-OH}$$
 но- $_{\rm CH_2-CH_2-N-CH_2-CH_2-OH}$

CM 3

CRN 79-10-7 CMF C3 H4 O2

L60 ANSWER 3 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN AN 2003:469556 HCAPLUS

PEZZUTO 10/025588 10/28/03 Page 30 DN 139:185518 Novel Biodegradable Ternary Copolymers hy-PEI-g-PCL-b-PEG: TΙ Synthesis, Characterization, and Potential as Efficient Nonviral Gene Delivery Vectors Shuai, Xintao; Merdan, Thomas; Unger, Florian; Wittmar, Matthias; Kissel, ΑU Department of Pharmaceutics and Biopharmacy, Philipps-University of CS Marburg, Marburg, D-35032, Germany Macromolecules (2003), 36(15), 5751-5759 SO CODEN: MAMOBX; ISSN: 0024-9297 American Chemical Society PB DTJournal English LΑ Diblock copolymers (MPEG-b-PCLs) of poly(\varepsilon-caprolactone) AΒ (PCL) and monomethoxyl poly(ethylene glycol) (MPEG) were synthesized by the conventional ring-opening polymerization of ϵ -caprolactone using MPEG as a macroinitiator. The monohydroxy-bearing diblock copolymers were reacted first with maleic anhydride and then with N-hydroxysuccinimide (NHS) to yield activated succinimidyl carbonate derivs. that are reactive with the primary amino group. Subsequently, a new class of biodegradable amphiphilic copolymer (hy-PEI-g-PCL-b-PEG) was prepared by grafting the activated PCL-b-PEG onto the hyperbranched poly(ethylene imine) (hy-PEI). Thermal properties of bulk graft copolymers were investigated using differential scanning calorimetry and thermogravimetric anal. Depending on their compns., these polymers are completely soluble in water or form micelles of tens to hundreds of nanometers in size in the studied concentration range, as revealed by surface tension and dynamic light scattering measurements of copolymer solns. Complexation of plasmid DNA (pDNA) with various copolymers was investigated to achieve particles of ca. 200 nm diameter (N/P = 7). Copolymer composition was found to affect

administration.

CC 63-6 (Pharmaceuticals)

Section cross-reference(s): 35

efficiency.

IT

- ST PEI PEG ternary copolymer DNA gene delivery
- DNA
 RL: PEP (Physical, engineering or chemical process); PRP (Properties); PYP
 (Physical process); THU (Therapeutic use); BIOL (Biological study); PROC
 (Process); USES (Uses)

also demonstrated by the viability of cells in the transfection expts.

promising candidates for gene delivery, featuring good biocompatibility,

Their neutral surface charge offers potential for i.v.

significantly the gene transfection efficiency of polyplexes. In general, low graft d. and high mol. weight of PEI blocks favor high gene transfection

efficiency. All DNA/copolymer complexes (N/P = 7) showed a much lower ξ -potential (i.e., neutral or neg.) than the DNA/PEI25 kDa complex (21 mV), indicating lower toxicity of copolymer-based complexes. Lower cytotoxicity of DNA/copolymer complexes was

potential biodegradability, and relatively high gene transfection

These results indicate that these ternary copolymers are

(plasmid; synthesis, characterization, and potential as efficient nonviral gene delivery vectors of a novel biodegradable ternary copolymers hy-PEI-g-PCL-b-PEG)

Polyoxyalkylenes, biological studies
RL: PRP (Properties); SPN (Synthetic preparation); THU (Therapeutic use);
BIOL (Biological study); PREP (Preparation); USES (Uses)
(polyamine-polyester-, block; synthesis, characterization,
and potential as efficient nonviral gene delivery vectors of a novel

```
biodegradable ternary copolymers hy-PEI-g-PCL-b-PEG)
     Polyesters, biological studies
IT
     RL: PRP (Properties); SPN (Synthetic preparation); THU (Therapeutic use);
     BIOL (Biological study); PREP (Preparation); USES (Uses)
        (polyamine-polyoxyalkylene-, block; synthesis,
        characterization, and potential as efficient nonviral gene delivery
        vectors of a novel biodegradable ternary copolymers
        hy-PEI-g-PCL-b-PEG)
TT
     Polyamines
     RL: PRP (Properties); SPN (Synthetic preparation); THU (Therapeutic use);
     BIOL (Biological study); PREP (Preparation); USES (Uses)
        (polyester-polyoxyalkylene-, block; synthesis,
        characterization, and potential as efficient nonviral gene delivery
        vectors of a novel biodegradable ternary copolymers
        hy-PEI-g-PCL-b-PEG)
     Critical micelle concentration
     Genetic vectors
     Glass transition temperature
     Particle size
     Solubility
     Surface tension
     Thermal decomposition
     Transformation, genetic
    Zeta potential
        (synthesis, characterization, and potential as efficient nonviral gene
        delivery vectors of a novel biodegradable ternary copolymers
        hy-PEI-q-PCL-b-PEG)
IT
     9014-00-0, Luciferase
     RL: PEP (Physical, engineering or chemical process); PRP (Properties); PYP
     (Physical process); THU (Therapeutic use); BIOL (Biological study); PROC
     (Process); USES (Uses)
        (synthesis, characterization, and potential as efficient nonviral gene
        delivery vectors of a novel biodegradable ternary copolymers
        hy-PEI-g-PCL-b-PEG)
IT
     579439-08-0P
     RL: PRP (Properties); SPN (Synthetic preparation); THU
     (Therapeutic use); BIOL (Biological study); PREP (Preparation);
     USES (Uses)
        (synthesis, characterization, and potential as efficient nonviral gene
        delivery vectors of a novel biodegradable ternary copolymers
        hy-PEI-g-PCL-b-PEG)
     6066-82-6D, N-Hydroxy succinimide, reaction products with PEG-caprolactone
TΨ
                 9002-98-6
     copolymer
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (synthesis, characterization, and potential as efficient nonviral gene
        delivery vectors of a novel biodegradable ternary copolymers
        hy-PEI-g-PCL-b-PEG)
     108-31-6DP, Maleic anhydride, reaction products with PEG-caprolactone
TΤ
                 263237-87-2P
                                579439-05-7P
                                               579439-07-9P
     copolymer
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (synthesis, characterization, and potential as efficient nonviral gene
        delivery vectors of a novel biodegradable ternary copolymers
        hy-PEI-g-PCL-b-PEG)
IT
     579439-08-0P
     RL: PRP (Properties); SPN (Synthetic preparation); THU
     (Therapeutic use); BIOL (Biological study); PREP (Preparation);
     USES (Uses)
```

(synthesis, characterization, and potential as efficient nonviral gene delivery vectors of a novel biodegradable ternary **copolymers** hy-PEI-g-PCL-b-PEG)

RN 579439-08-0 HCAPLUS

CN 2-Oxepanone, polymer with oxirane, mono[hydrogen (2Z)-2-butenedioate], methyl ether, block, polymer with aziridine, graft (9CI) (CA INDEX NAME)

CM 1

CRN 151-56-4 CMF C2 H5 N



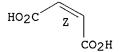
CM 2

CRN 579439-05-7 CMF (C6 H10 O2 . C2 H4 O)x . C4 H4 O4 . C H4 O

CM 3

CRN 110-16-7 CMF C4 H4 O4

Double bond geometry as shown.



CM 4

CRN 67-56-1 CMF C H4 O

нзс-он

CM 5

CRN 107596-21-4

CMF (C6 H10 O2 . C2 H4 O) x

CCI PMS

CM 6

CRN 502-44-3 CMF C6 H10 O2



CM 7

CRN 75-21-8 CMF C2 H4 O

RE.CNT 40 THERE ARE 40 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L60 ANSWER 4 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 2003:430745 HCAPLUS

DN 139:214851

TI Synthesis and cholesterol-lowering effect of poly(ethylene glycol)-polyamine graft and block polymers

AU Nakamura, Tohru; Hamamich, Yoshiko; Uehara, Keiji; Ishii, Takehiko; Hayashi, Hisato; Nagasaki, Yukio; Kataoka, Kazunori

CS Self Medication Laboratories, Taisho Pharmaceutical Co., Ltd., Saitamashi, Saitama, 330-8530, Japan

SO Kobunshi Ronbunshu (2003), 60(5), 238-240 CODEN: KBRBA3; ISSN: 0386-2186

PB Kobunshi Gakkai

DT Journal

LA Japanese

AB Both water-soluble graft and block copolymers consisting of poly(ethylene glycol) segment and polyamine segment were synthesized via our original synthetic technique. Oral administrations of these copolymers to high cholesterol diet fed rats showed an effective reduction of plasma cholesterol, indicating that bile acids were effectively bound to the polymers and excreted from the GI tract. The PEG/polyamine copolymer thus prepared is promising as a high performance cholesterol-lowering agent.

CC 35-8 (Chemistry of Synthetic High Polymers)
Section cross-reference(s): 1

IT Polyoxyalkylenes, preparation

RL: BSU (Biological study, unclassified); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation)

(block; synthesis and cholesterol-lowering effect of poly(ethylene glycol)-polyamine graft and block polymers)

IT Polyoxyalkylenes, preparation

RL: BSU (Biological study, unclassified); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation)

(graft polymers; synthesis and cholesterol-lowering effect of

poly(ethylene glycol)-polyamine graft and block polymers)

IT Anticholesteremic agents

```
(synthesis and cholesterol-lowering effect of poly(ethylene glycol)-
       polyamine graft and block polymers)
     260426-24-2P 590424-06-9P, Polyethylene glycol methyl
IT
     ether methacrylate-diethylaminoethyl methacrylate graft copolymer
     590424-07-0P
                   591206-05-2P 591206-08-5P, Diethylaminoethyl
    methacrylate-ethylene oxide graft copolymer methyl ether
     RL: BSU (Biological study, unclassified); SPN (Synthetic
    preparation); BIOL (Biological study); PREP (Preparation)
        (synthesis and cholesterol-lowering effect of poly(ethylene glycol)-
       polyamine graft and block polymers)
     260426-24-2P 590424-06-9P, Polyethylene glycol methyl
     ether methacrylate-diethylaminoethyl methacrylate graft copolymer
     591206-08-5P, Diethylaminoethyl methacrylate-ethylene oxide graft
     copolymer methyl ether
    RL: BSU (Biological study, unclassified); SPN (Synthetic
    preparation); BIOL (Biological study); PREP (Preparation)
        (synthesis and cholesterol-lowering effect of poly(ethylene glycol)-
       polyamine graft and block polymers)
     260426-24-2 HCAPLUS
RN
CN
     2-Propenoic acid, 2-methyl-, 2-(diethylamino)ethyl ester, polymer with
     oxirane, monomethyl ether, block (9CI) (CA INDEX NAME)
     CM
          1
     CRN
         67-56-1
     CMF C H4 O
нзс-он
     CM
          214957-23-0
     CRN
          (C10 H19 N O2 . C2 H4 O)x
     CMF
     CCI
          PMS
          CM
               3
          CRN
               105-16-8
          CMF
              C10 H19 N O2
 H<sub>2</sub>C
Me-C-C-O-CH_2-CH_2-NEt_2
          CM
          CRN
               75-21-8
          CMF C2 H4 O
```

 $\stackrel{\circ}{\triangle}$

RN 590424-06-9 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-(diethylamino)ethyl ester, polymer with α -(2-methyl-1-oxo-2-propenyl)- ω -methoxypoly(oxy-1,2-ethanediyl), graft (9CI) (CA INDEX NAME)

CM 1

CRN 26915-72-0

CMF (C2 H4 O)n C5 H8 O2

CCI PMS

$$\begin{array}{c|c} ^{H_2C} & \text{O} \\ \parallel & \parallel \\ \text{Me} - \text{C} - \text{C} & - \\ \hline \end{array} \begin{array}{c} \text{O} - \text{CH}_2 - \text{CH}_2 \\ \hline \end{array} \begin{array}{c} \text{OMe} \\ \end{array}$$

CM 2

CRN 105-16-8 CMF C10 H19 N O2

RN 591206-08-5 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-(diethylamino)ethyl ester, polymer with oxirane, methyl ether, graft (9CI) (CA INDEX NAME)

CM 1

CRN 67-56-1 CMF C H4 O

нзс-он

CM 2

CRN 591206-07-4

CMF (C10 H19 N O2 . C2 H4 O) \times

CCI PMS

CM 3

CRN 105-16-8

Page 36 PEZZUTO 10/025588 10/28/03

CMF C10 H19 N O2

H₂C 0 Ш Me-C-C-O-CH2-CH2-NEt2

CM

CRN 75-21-8 CMF C2 H4 O



L60 ANSWER 5 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

2003:132559 HCAPLUS ΑŃ

138:172295 DN

Laundering detergent compositions with suppressed resoiling of clothes by TΤ clays during washing

Akasaki, Kazumoto; Morimoto, Masakazu; Saeki, Takuya; Fujii, Yoshikazu; IN Yamaguchi, Shigeru

PA Nippon Shokubai Co., Ltd., Japan

Jpn. Kokai Tokkyo Koho, 13 pp. SO

CODEN: JKXXAF

DTPatent

Japanese T.A

FAN.CNT 1

PΙ

PATENT NO. KIND DATE APPLICATION NO. DATE -----_____ ----_____ A2 JP 2001-241464 20010808 20030221 JP 2003049194 20010808 PRAI JP 2001-241464

The compns. contain polymers having anionic groups and cationic groups that are manufactured by aminoethylation of some of the anionic groups in anionic polymers. Thus, polyethylene glycol monomethyl ether was reacted with maleic anhydride, polymerized with acrylic acid, further reacted with ethyleneimine, and mixed with surfactants to give a detergent compns. showing good clay dispersibility and washing power for cotton fabrics.

ICM C11D003-37 IC

ICS C08F008-32

46-5 (Surface Active Agents and Detergents) CC

IT Polyamines

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(acrylic, graft; acrylic polymers containing cationic and anionic groups manufactured by aminoethylation for anti-resoiling laundering detergent)

IT Polyoxyalkylenes, uses

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(polyamine-, graft, acrylic; acrylic polymers containing cationic and anionic groups manufactured by aminoethylation for anti-resoiling laundering detergent)

IT Polyamines RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(polyoxyalkylene-, graft, acrylic; acrylic polymers containing cationic and anionic groups manufactured by aminoethylation for anti-resoiling laundering detergent)

IT 497221-97-3P 497221-98-4P 497818-85-6P, Acrylic acid-ethyleneimine-ethylene oxide graft copolymer methyl ether RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(acrylic polymers containing cationic and anionic groups manufactured by aminoethylation for anti-resoiling laundering detergent)

IT 497221-97-3P 497818-85-6P, Acrylic acid-ethyleneimine-

ethylene oxide graft copolymer methyl ether

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(acrylic polymers containing cationic and anionic groups manufactured by aminoethylation for anti-resoiling laundering detergent)

RN 497221-97-3 HCAPLUS

CN 2-Propenoic acid, polymer with aziridine and α -[(2Z)-3-carboxy-1-oxo-2-propenyl]- ω -methoxypoly(oxy-1,2-ethanediyl), graft (9CI) (CA INDEX NAME)

CM 1

CRN 31833-82-6 CMF (C2 H4 O)n C5 H6 O4 CCI PMS

$$HO_2C-CH=CH-C-CH_2-CH_2-CH_2$$
 OMe

CM 2

CRN 151-56-4 CMF C2 H5 N



CM 3

CRN 79-10-7 CMF C3 H4 O2

PEZZUTO 10/025588 10/28/03 Page 38

RN 497818-85-6 HCAPLUS

CN 2-Propenoic acid, polymer with aziridine and oxirane, methyl ether, graft (9CI) (CA INDEX NAME)

CM 1

CRN 67-56-1 CMF C H4 O

нзс-он

CM 2

CRN 497818-84-5

CMF (C3 H4 O2 . C2 H5 N . C2 H4 O) x

CCI PMS

CM 3

CRN 151-56-4 CMF C2 H5 N

H N /\

CM 4

CRN 79-10-7 CMF C3.H4 O2

о || но- с- сн== сн₂

CM 5

CRN 75-21-8 CMF C2 H4 O

 $^{\circ}$

L60 ANSWER 6 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN AN 2002:538098 HCAPLUS

KATHLEEN FULLER EIC 1700/PARKER LAW 308-4290

```
DN
    137:110740
    Compositions for the production of molded shapes from finely divided
ΤI
    materials
     Kroner, Matthias; Gerst, Matthias; Reck, Bernd
IN
     Basf A.-G., Germany
PA
SO
     Ger. Offen., 20 pp.
     CODEN: GWXXBX
     Patent
DT
     German
LA
FAN.CNT 1
                    KIND DATE
                                         APPLICATION NO. DATE
     PATENT NO.
                           -----
                     ____
    DE 10101944
                            20020718
                                          DE 2001-10101944 20010117
                     A1
PΙ
                     A1
                                          US 2002-44948
                            20020919
                                                            20020115
    US 2002130439
                                          EP 2002-1203
     EP 1225193
                      A2
                            20020724
                                                           20020116
                     A3
                           20030423
     EP 1225193
            AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
            IE, SI, LT, LV, FI, RO, MK, CY, AL, TR
                            20010117
PRAI DE 2001-10101944 A
    MARPAT 137:110740
OS
    Compns. containing reaction products of HOCOCHYCXRCO2H (I, R = H or CH2CO2H, X
AΒ
     = OH or NH2 when Y = H, Y = OH or NH2 when X = H, XY = \pi bond) and(or)
     their anhydrides with NH3 and, optionally, primary amines and(or) polyols
     or an mixture of I and (or) their anhydride and heat-sensitive, NH3-forming
     compound and, optionally, primary amines and(or) polyols are useful as
     thermally hardenable binders for manufacture of moldings from finely divided
    materials (such as wood fibers) and strengthening of sheets from fibrous
    materials (such as natural fiber mats). A typical particleboard composition
    was manufacture by spraying a 40% aqueous solution of maleic acid monoamide
(II) onto
     180 g wood fiber and drying to 7% moisture content, so that the composition
     contained 20 g II.
IC
     ICM C09D005-12
     ICS C09D177-00; C09D123-00; C08J005-04; C08J003-24; D04H001-42
     43-9 (Cellulose, Lignin, Paper, and Other Wood Products)
CC
     Section cross-reference(s): 58
    polycarboxylic ammonia adduct polymer binder finely divided
ST
     material molding; maleic acid monoamide polymer binder particleboard
     29996-04-1P, Maleic acid monoamide homopolymer
IT
                                                    31586-29-5P,
     Poly(2,5-dioxo-1,3-pyrrolidinediyl) 39444-67-2P, Maleic acid ammonium
                      172280-28-3P, Ammonium carbonate-maleic anhydride
     salt homopolymer
                             442844-23-7P, Ammonium carbonate-malic
     copolymer 442844-21-5P
     acid-triethanolamine copolymer
                                      442844-25-9P, Ammonium carbonate-citric
     acid-triethanolamine copolymer 442844-27-1P, Maleic acid
    monoamide-triethanolamine copolymer
     RL: IMF (Industrial manufacture); PRP (Properties); TEM
     (Technical or engineered material use); PREP (Preparation); USES
     (Uses)
        (compns. containing succinimide polymer-forming compds. for thermal
        of molded shapes from finely divided materials)
     442844-21-5P 442844-27-1P, Maleic acid
     monoamide-triethanolamine copolymer
     RL: IMF (Industrial manufacture); PRP (Properties); TEM
     (Technical or engineered material use); PREP (Preparation); USES
     (Uses)
        (compns. containing succinimide polymer-forming compds. for thermal
production
```

PEZZUTO 10/025588 10/28/03 Page 40

of molded shapes from finely divided materials)

RN 442844-21-5 HCAPLUS

CN 2-Butenedioic acid (2Z)-, ammonium salt, polymer with 2,2',2''- nitrilotris[ethanol] (9CI) (CA INDEX NAME)

CM 1

CRN 13716-99-9

CMF C4 H4 O4 . x H3 N

Double bond geometry as shown.

●x NH3

CM 2

CRN 102-71-6 CMF C6 H15 N O3

$$\begin{array}{c} \text{CH}_2-\text{CH}_2-\text{OH} \\ | \\ \text{HO-CH}_2-\text{CH}_2-\text{N-CH}_2-\text{CH}_2-\text{OH} \end{array}$$

RN 442844-27-1 HCAPLUS

CN 2-Butenoic acid, 4-amino-4-oxo-, (2Z)-, polymer with 2,2',2''nitrilotris[ethanol] (9CI) (CA INDEX NAME)

CM :

CRN 557-24-4 CMF C4 H5 N O3

Double bond geometry as shown.

CM 2

CRN 102-71-6 CMF C6 H15 N O3

```
CH2-CH2-OH
 HO-CH_2-CH_2-N-CH_2-CH_2-OH
 L60 ANSWER 7 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN
 AN
      2002:536415 HCAPLUS
 DN
      137:94558
     Unsaturated polyfunctional amide-based polymer dental material
 ΤI
     Moszner, Norbert; Zeuner, Frank; Rheinberger, Volker; Angermann, Jorg;
 IN
      Voelkel, Thomas
 PA
      Ivoclar Vivadent AG, Liechtenstein
      Eur. Pat. Appl., 18 pp.
      CODEN: EPXXDW
 DT
      Patent
 T.A
      German
 FAN.CNT 1
      PATENT NO.
                      KIND DATE
                                          APPLICATION NO.
                                                            DATE
                      ----
                                           -----
                                                            -----
PΤ
      EP 1222910
                      A2 20020717
                                           EP 2002-118
                                                            20020103
         R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, SI, LT, LV, FI, RO, MK, CY, AL, TR
     DE 10101523
                      A1
                            20020801
                                           DE 2001-10101523 20010115
     JP 2002212019
                       A2
                            20020731
                                           JP 2002-5383
                                                            20020111
     US 2002143138
                      A1
                            20021003
                                           US 2002-45358
                                                            20020114
PRAI DE 2001-10101523 A
                            20010115
     US 2001-289097P
                      P
                            20010507
     Polymers based on compds. having 2-5 \alpha, \beta-unsatd. amide groups
AΒ
     are hydrolysis resistant and are useful for dental materials. A typical
     polymer was manufactured by polymerization of ethylenebisacrylamide.
IC
     ICM A61K006-083
     37-3 (Plastics Manufacture and Processing)
CC
     Section cross-reference(s): 63
IT
     Dental materials and appliances
        (cements; unsatd. polyfunctional amide-based polymer with
        good hydrolysis resistance for dental materials)
TΤ
     442573-55-9P
     RL: IMF (Industrial manufacture); TEM (Technical or engineered
    material use); PREP (Preparation); USES (Uses)
        (cured dental adhesive; unsatd. polyfunctional amide-based polymer with
        good hydrolysis resistance for dental materials)
IT
     442573-55-9P
    RL: IMF (Industrial manufacture); TEM (Technical or engineered
    material use); PREP (Preparation); USES (Uses)
        (cured dental adhesive; unsatd. polyfunctional amide-based polymer with
       good hydrolysis resistance for dental materials)
RN
    442573-55-9 HCAPLUS
    2-Butenoic acid, 4-(1-phosphonoethoxy)-, 1-ethyl ester, polymer with
CN
    2-hydroxyethyl 2-methyl-2-propenoate, N,N'-1,3-propanediylbis[N-ethyl-2-
    propenamide] and 1,2,3-propanetriol bis(2-methyl-2-propenoate) (9CI) (CA
    INDEX NAME)
    CM
         1
    CRN 442573-54-8
    CMF C8 H15 O6 P
```

$$\begin{array}{c|c} \text{O} & \text{PO}_3\text{H}_2 \\ || & | \\ \text{EtO-C-CH} = \text{CH-CH}_2\text{-O-CH-Me} \end{array}$$

CRN 442200-41-1 CMF C13 H22 N2 O2

CM 3

CRN 868-77-9 CMF C6 H10 O3

CM 4

CRN 28497-59-8 CMF C11 H16 O5 CCI IDS

CM 5

CRN 79-41-4 CMF C4 H6 O2

CM 6

CRN 56-81-5 CMF C3 H8 O3

ОН но-сн2-сн-сн2-он

L60 ANSWER 8 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 2002:169789 HCAPLUS

DN 136:233650

resin compositions for storage-stable and low-temperature-curable coatings TT having good adhesion to various substrates

Kageishi, Ichiji; Osanai, Yoshitaka; Ando, Yumi IN

Toray Industries, Inc., Japan PA

Jpn. Kokai Tokkyo Koho, 23 pp. SO CODEN: JKXXAF

DT Patent.

Japanese LΑ

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE --------------JP 2002069363 A2 20020308 JP 2000-255467 20000825 PRAI JP 2000-255467 20000825

Title compns. comprise hardeners and resin blends consisting of alkoxy-containing unsatd. compound-polymerized acrylic resins with a number-average mol.

weight (Mn) of 500-80,000 and epoxy group-containing unsatd. compound (EU)-polymerized

acrylic resins with Mn of 500-80,000 at unreacted EU of \leq 5,000 ppm in total resin blends. A composition comprising acetylacetone, Al tris(acetylacetonate), and a blend of 20% Me methacrylate (I)-SZ 6030 copolymer and 80% I-Bu acrylate-Bu methacrylate-Cyclomer A 200 copolymer (unreacted Cyclomer A 200 of 50 ppm) was diluted with organic solvents to form a clear coat with good storage stability at 23° or 40° for 1 mo, which was sprayed on an Al alloy, glass, ABS, or slate plate to a 40-μm thickness, and baked at 80° for 30 min to form a film showing good adhesion to the plate and good blocking and weather resistance.

ICM C09D143-04 IC

ics c09D133-04; c09D151-00; c09D163-00; c09D175-04

42-10 (Coatings, Inks, and Related Products)

(asbestos, slate, substrates; alkoxysilyl-containing and epoxy-containing acrylic resin-based low-temperature-curable coatings with good adhesion to various substrates)

26936-30-1P, Methyl methacrylate-SZ 6030 copolymer ΤТ 146241-49-8P, Butyl acrylate-butyl methacrylate-methyl methacrylate-Cyclomer A 200 copolymer 318988-59-9P, Butyl acrylate-butyl methacrylate-2-hydroxyethyl methacrylate-methyl methacrylate-Cyclomer A 200 copolymer Butyl acrylate-butyl methacrylate-3-glycidoxypropyltrimethoxysilane-2-403500-04-9P, hydroxyethyl methacrylate-3-methacryloxypropyltrimethoxysilane-methyl methacrylate-Cyclomer A 200 copolymer 403500-05-0P, Butyl acrylate-butyl methacrylate-2-hydroxyethyl methacrylate-3-methacryloxypropyltrimethoxysil ane-methyl methacrylate-3,4-epoxycyclohexylmethyl acrylate-Sumidur N 3500 copolymer 403500-06-1P, Acrylic acid-butyl acrylate-glycidyl methacrylate-2-hydroxyethyl methacrylate-methyl methacrylate copolymer triethylamine salt 403500-07-2P, 2-Hydroxyethyl methacrylate-3methacryloxypropyltrimethoxysilane-methyl methacrylate-trifluoroethyl

```
methacrylate-Cyclomer A 200 copolymer
                                              403500-08-3P, Butyl acrylate-butyl
       methacrylate-glycidyl methacrylate-methyl methacrylate-SZ 6030 graft
                   403500-10-7P, Butyl acrylate-butyl methacrylate-glycidyl
       methacrylate-3-glycidoxypropyltrimethoxysilane-2-hydroxyethyl
       methacrylate-3-methacryloxypropyltrimethoxysilane-methyl
       methacrylate-trifluoroethyl methacrylate-Cyclomer A 200 copolymer
       403500-11-8P, Butyl acrylate-butyl methacrylate-glycidyl
       methacrylate-3-glycidoxypropyltrimethoxysilane-2-hydroxyethyl
       methacrylate-3-methacryloxypropyltrimethoxysilane-methyl
       methacrylate-trifluoroethyl methacrylate-phenyltrimethoxysilane-
       phenylmethyldimethoxysilane-Cyclomer A 200 copolymer
       acrylate-butyl methacrylate-glycidyl methacrylate-2-hydroxyethyl
                                                             403500-12-9P, Butyl
       methacrylate-3-methacryloxypropyltrimethoxysilane-methyl
      methacrylate-trifluoroethyl methacrylate-Cyclomer A 200 copolymer
       403500-13-0P, Butyl acrylate-butyl methacrylate-3,4-epoxycyclohexylmethyl
      acrylate-glycidyl methacrylate-2-hydroxyethyl methacrylate-3-
      methacryloxypropyltrimethoxysilane-methyl methacrylate-trifluoroethyl
      methacrylate-Sumidur N 3500 copolymer
      RL: IMF (Industrial manufacture); POF (Polymer in formulation);
      TEM (Technical or engineered material use); PREP (Preparation);
         (alkoxysilyl-containing and epoxy-containing acrylic resin-based
         low-temperature-curable coatings with good adhesion to various substrates)
      403500-06-1P, Acrylic acid-butyl acrylate-glycidyl
      methacrylate-2-hydroxyethyl methacrylate-methyl methacrylate copolymer
      RL: IMF (Industrial manufacture); POF (Polymer in formulation);
      TEM (Technical or engineered material use); PREP (Preparation);
         (alkoxysilyl-containing and epoxy-containing acrylic resin-based
         low-temperature-curable coatings with good adhesion to various substrates)
     403500-06-1 HCAPLUS
     2-Propenoic acid, 2-methyl-, 2-hydroxyethyl ester, polymer with butyl
     2-propenoate, methyl 2-methyl-2-propenoate, oxiranylmethyl
     2-methyl-2-propenoate and 2-propenoic acid, compd. with
     N, N-diethylethanamine (9CI) (CA INDEX NAME)
     CM
     CRN
         121-44-8
     CMF C6 H15 N
   Εt
Et-N-Et
    CM
    CRN
         69645-73-4
         (C7 H12 O2 . C7 H10 O3 . C6 H10 O3 . C5 H8 O2 . C3 H4 O2)x
    CCI
         CM
              3
         CRN 868-77-9
```

CMF C6 H10 O3

CM 4

CRN 141-32-2 CMF C7 H12 O2

$$\begin{array}{c} \circ \\ \parallel \\ \text{n-BuO-C-CH-----} \text{CH}_2 \end{array}$$

CM 5

CRN 106-91-2 CMF C7 H10 O3

$$\overset{\text{O}}{\longleftarrow} \overset{\text{O}}{\underset{\text{CH}_2-\text{O}-\text{C}-\text{C}-\text{Me}}{\text{Me}}} \overset{\text{CH}_2}{\underset{\text{CH}_2-\text{O}-\text{C}-\text{C}-\text{Me}}{\text{Me}}}$$

CM 6

CRN 80-62-6 CMF C5 H8 O2

CM 7

CRN 79-10-7 CMF C3 H4 O2

L60 ANSWER 9 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

KATHLEEN FULLER EIC 1700/PARKER LAW 308-4290

```
2002:63622 HCAPLUS
AN
     136:122323
DN
    Asphalt emulsification dispersant and its use in oil-in-water asphalt
ΤI
     emulsion for paving
     Iizuka, Masanori; Sasaki, Hirotaka; Tamaoki, Ryoichi; Honma, Yuichi
IN
     Kao Corp., Japan
PA
     Jpn. Kokai Tokkyo Koho, 8 pp.
SO
     CODEN: JKXXAF
DT
     Patent
     Japanese
LA
FAN.CNT 1
     PATENT NO.
                     KIND DATE
                                           APPLICATION NO. DATE
     JP 2002020626
                            20020123
                                           JP 2000-204688 20000706
                      A2
PΙ
PRAI JP 2000-204688
                            20000706
     The dispersant contains water-soluble copolymers comprising (a) vinyl
     monomers containing average 2-300 mol of C2-3 oxyalkylenes and (b) vinyl
     containing CO2H, SO3H, amide, and/or their water-soluble salts. Cement
     and aggregates are mixed homogeneously with the emulsion using the
     dispersant, and pavement formed with the emulsion has high durability.
IC
     ICM C08L095-00
     ICS C04B024-26; C04B026-26; C04B028-02; C08F290-06; C08K005-00;
          C04B024-36; C04B103-40
     58-4 (Cement, Concrete, and Related Building Materials)
CC
     Section cross-reference(s): 37
IT
     111740-39-7P, Methacrylic acid-polyethylene glycol methyl ether
     methacrylate graft copolymer 221881-27-2P, Methacrylic acid-polyethylene
     glycol methyl ether methacrylate graft copolymer sodium salt
     221882-30-0P, Ethylene oxide-methacrylic acid graft copolymer methyl ether
     sodium salt 223122-81-4P, Ethylene oxide-methacrylic acid graft
     copolymer methyl ether 286007-97-4P, Maleic acid-polyethylene glycol
     allyl methyl ether graft copolymer 354137-33-0P, Ethylene
     oxide-methacrylic acid-propylene oxide graft copolymer methyl ether
     359793-25-2P, Ethylene oxide-maleic acid graft copolymer methyl ether
     374622-03-4P
                   390417-45-5P 390745-31-0P, Ethylene
     oxide-methacrylic acid graft copolymer methyl ether triethanolamine salt
     390745-33-2P, Ethylene oxide-methacrylsulfonic acid-sodium methacrylate
     graft copolymer methyl ether
     RL: MOA (Modifier or additive use); PNU (Preparation, unclassified); PRP
     (Properties); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (dispersant; water-soluble polymer dispersant for asphalt emulsification
        and its use in emulsion for paving)
ΙT
     374622-03-4P 390745-31-0P, Ethylene oxide-methacrylic
     acid graft copolymer methyl ether triethanolamine salt
     RL: MOA (Modifier or additive use); PNU (Preparation, unclassified); PRP
     (Properties); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (dispersant; water-soluble polymer dispersant for asphalt emulsification
        and its use in emulsion for paving)
RN
     374622-03-4 HCAPLUS
CN
     2-Propenoic acid, 2-methyl-, polymer with \alpha-(2-methyl-1-oxo-2-
     propenyl)-\omega-methoxypoly(oxy-1,2-ethanediyl), graft, compd. with
     2,2',2''-nitrilotris[ethanol] (9CI) (CA INDEX NAME)
     CM
```

KATHLEEN FULLER EIC 1700/PARKER LAW 308-4290

PEZZUTO 10/025588 10/28/03 Page 47

CRN 102-71-6 CMF C6 H15 N O3

CM 2

CRN 111740-39-7

CMF (C4 H6 O2 . (C2 H4 O)n C5 H8 O2)x

CCI PMS

CM 3

CRN 26915-72-0

CMF (C2 H4 O)n C5 H8 O2

CCI PMS

$$H_2C$$
 O H_2C H_2C

CM 4

CRN 79-41-4 CMF C4 H6 O2

$$^{\mathrm{CH_2}}_{\parallel}$$
 Me-C-CO₂H

RN 390745-31-0 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, polymer with oxirane, methyl ether, graft, compd. with 2,2',2''-nitrilotris[ethanol] (9CI) (CA INDEX NAME)

CM 1

CRN 102-71-6 CMF C6 H15 N O3

CM 2

PEZZUTO 10/025588 10/28/03 Page 48

CRN 223122-81-4
CMF (C4 H6 O2 . C2 H4 O) x . x C H4 O

CM 3

CRN 67-56-1
CMF C H4 O

H3C-OH

CM 4

CRN 167763-01-1

CMF (C4 H6 O2 . C2 H4 O) x

CCI PMS

5

CM

CRN 79-41-4 CMF C4 H6 O2

 $\begin{array}{c} \text{CH}_2 \\ || \\ \text{Me-C-CO}_2 \text{H} \end{array}$

CM 6

CRN 75-21-8

CMF C2 H4 0

 $^{\circ}$

L60 ANSWER 10 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN AN 2001:300791 HCAPLUS DN 134:312217 Method for preparation of functional polymeric surface with improved TI adhesion and surface properties Bilyk, Alexander; Li, Sheng; Yang, Wei Dong; Hoobin, Pamela Maree; IN Russell, Lee Joy; Gutowski, Wojciech Stanislaw Commonwealth Scientific and Industrial Research Organisation, Australia PΑ PCT Int. Appl., 44 pp. SO CODEN: PIXXD2 DTPatent LΑ English FAN.CNT 1 PATENT NO. KIND DATE APPLICATION NO.

```
PΙ
      WO 2001029118
                        A1
                              20010426
                                             WO 2000-AU1272
                                                               20001019
          W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,
              CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR,
              HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT,
              LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU,
              SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN,
              YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
          RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ,
              CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
      BR 2000014904
                        Α
                              20020611
                                             BR 2000-14904
                                                               20001019
      EP 1242516
                        A1
                              20020925
                                             EP 2000-971135
                                                               20001019
              AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
              IE, SI, LT, LV, FI, RO, MK, CY, AL
      JP 2003512490
                        T2
                             20030402
                                             JP 2001-531910
                                                               20001019
      US 2003194504
                        Α1
                             20031016
                                             US 2002-120252
                                                              20020411
PRAI AU 1999-3521
                        Α
                             19991019
     AU 1999-4905
                        Α
                             19991230
     AU 1999-9935
                        A
                             19991019
     AU 1999-9949
                        Α
                             19991230
     WO 2000-AU1272
                             20001019
                        W
     Title method for providing a crosslinked network grafted to the polymeric
AΒ
     substrate surface includes: (i) providing the polymeric surface with
     functional groups; and (ii) contacting the surface with (a) a polyamine
     compound reactive with the surface functional groups, wherein the polyamine
     comprises \geq 4 amine groups including \geq 2 amine groups selected
     from primary and secondary amine groups, and (b) a crosslinking agent
     reactive with the polyamine. Thus a mineral filled automotive grade
     polypropylene of Corton PDR 1054/2 HS Natural was flame treated with an
     Aerogem FT Lab Model flame treatment unit and sprayed with a Lupasol
     FC/glutaraldehyde formulation, then painted with a Heron White base coat
     and a clear polyurethane top coat and cured, showing pull off strength
     after 3 days at 80° 5.1 MPa and cohesive failure of the substrate
     49%.
     ICM C08J007-12
     38-3 (Plastics Fabrication and Uses)
     Section cross-reference(s): 37, 42
IT
     Polyamines
     RL: TEM (Technical or engineered material use); USES (Uses)
        (method for preparation of functional polymeric surface with improved
        adhesion and surface properties)
ΙT
     Coordination compounds
     Halides
     Oxides (inorganic), miscellaneous
       Polyoxyalkylenes, miscellaneous
     Polysaccharides, miscellaneous
    RL: MSC (Miscellaneous)
        (method for preparation of functional polymeric surface with improved
       adhesion and surface properties with functional mols.)
    106400-60-6P, Acrylic acid-propylene graft copolymer
    335259-44-4P, Allylamine-3-chloro-2-hydroxypropyl acrylate-propylene
               335259-45-5P 335259-46-6P
    copolymer
                                               335259-47-7P
    335259-48-8P, Acrylic acid-allylamine-3-chloro-2-hydroxypropyl
    acrylate-propylene copolymer
                                   335277-16-2P,
    Allylamine-3-chloro-2-hydroxypropyl acrylate-dextran glycidyl
    acrylate-propylene copolymer 335277-17-3P, Dextran glycidyl
    acrylate-propylene graft copolymer
                                         335318-07-5P
                                                        335318-08-6P
    335446-98-5P 335446-99-6P 335447-00-2P 335447-01-3P
```

335447-02-4P 335447-03-5P 335447-04-6P 335447-05-7P RL: IMF (Industrial manufacture); PRP (Properties); TEM

(Technical or engineered material use); PREP (Preparation); USES (Uses)

(method for preparation of functional polymeric surface with improved adhesion and surface properties)

IT 335447-00-2P 335447-01-3P 335447-02-4P

RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(method for preparation of functional polymeric surface with improved adhesion and surface properties)

RN 335447-00-2 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-[(1-oxo-2-propenyl)oxy]ethyl ester, polymer with aziridine and 1-propene (9CI) (CA INDEX NAME)

CM 1 .

CRN 69040-48-8 CMF C9 H12 O4

CM 2

CRN 151-56-4 CMF C2 H5 N



CM 3

CRN 115-07-1 CMF C3 H6

 $H_3C-CH=CH_2$

RN 335447-01-3 HCAPLUS

CN 2-Propenoic acid, 3-chloro-2-hydroxypropyl ester, polymer with aziridine and 1-propene (9CI) (CA INDEX NAME)

CM 1

CRN 3326-90-7 CMF C6 H9 C1 O3

CRN 151-56-4 CMF C2 H5 N



CM 3

CRN 115-07-1 CMF C3 H6

 $H_3C-CH=CH_2$

RN 335447-02-4 HCAPLUS
CN 2-Propenoic acid, oxiranylmethyl ester, polymer with aziridine and
1-propene (9CI) (CA INDEX NAME)

CM 1

CRN 151-56-4 CMF C2 H5 N



CM 2

CRN 115-07-1 CMF C3 H6

 $H_3C-CH=CH_2$

CM 3

CRN 106-90-1 CMF C6 H8 O3

THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD RE.CNT 7 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L60 ANSWER 11 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

2000:747064 HCAPLUS AN

DN 133:310786

High-concentration solutions of amino group-containing TI α, β -unsaturated nitrile-conjugated diene rubber compositions with low viscosity and good storage stability

Koshimura, Katsuo; Tsuneyoshi, Minoru; Yamazaki, Takao IN

PΑ JSR Co., Ltd., Japan

Jpn. Kokai Tokkyo Koho, 8 pp. SO

CODEN: JKXXAF

DT Patent

LΑ Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE		
PI	JP 2000297182	A2	20001024	JP 1999-105801	19990413		
PRAI	JP 1999-105801		19990413				

- The compns., useful as rubber cements for printing blankets, AB metal gasket coatings, friction materials, etc., contain amino group-containing α,β -unsatd. nitrile-conjugated diene rubbers [having α , β -unsatd. nitrile units 10-50, conjugated diene units 30-89.9, and CH2:CRCO2XNR1R2 units (R = H, C1-4 alkyl; R1, R2 = C1-8 hydrocarbyl; X = divalent linkage group) 0.1-20%] dissolved in polar solvents. Thus, 100 parts 60:30:10 butadiene-acrylonitrilediethylaminoethyl methacrylate rubber was kneaded with 40 parts Nipsil VN 3 (wet silica) and other additives, vulcanized, and dissolved in PhMe [solubility parameter 8.9 (cal/cm3)1/2] to give a 30% rubber cement with viscosity change 5% after 7-day storage at 25°, which was press-cured to give a test piece with tensile strength 25.0 MPa and elongation at break 850%.
- IC ICM C08L013-00

C08F220-42; C08F236-04; C08J003-09; C08K003-04; C08K003-34; C08L009-02; C08L011-00; C08L033-18; C08F220-34

39-9 (Synthetic Elastomers and Natural Rubber) CC

rubber acrylonitrile butadiene ethylaminoethyl methacrylate soln; nitrile ST conjugated diene rubber silica reinforced; blanket gasket damper rubber soln low viscosity; cement rubber amino nitrile conjugated diene

IT 91277-34-8P, Acrylonitrile-butadiene-diethylaminoethyl

methacrylate copolymer

RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(rubber, vulcanized; solns. of amino group-containing α, β unsatd. nitrile-conjugated diene rubber compns. with good storage stability)

91277-34-8P, Acrylonitrile-butadiene-diethylaminoethyl IT methacrylate copolymer

RL: IMF (Industrial manufacture); POF (Polymer in formulation);

PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (rubber, vulcanized; solns. of amino group-containing α,β unsatd. nitrile-conjugated diene rubber compns. with good storage stability) RN 91277-34-8 HCAPLUS 2-Propenoic acid, 2-methyl-, 2-(diethylamino)ethyl ester, polymer with CN 1,3-butadiene and 2-propenenitrile (9CI) (CA INDEX NAME) CM CRN 107-13-1 CMF C3 H3 N $H_2C = CH - C = N$ CM 2 CRN 106-99-0 CMF C4 H6 $H_2C = CH - CH = CH_2$ CM 3 CRN 105-16-8 CMF C10 H19 N O2 H₂C 0 Me-C-C-O-CH2-CH2-NEt2 L60 ANSWER 12 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN 2000:742150 HCAPLUS ANDN 133:310443 Aqueous superabsorbent polymer solution, its manufacture and use in ΤI agricultural and consumer goods Anderson, Stewart C.; Miller, Wayne P. IN H.B. Fuller Licensing and Financing, Inc, USA PA SO PCT Int. Appl., 23 pp. CODEN: PIXXD2 DT Patent LA English FAN.CNT 1

WO 2000061642 20001019 WO 2000-US10134 20000413 A1 PΙ W: AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, CU, CZ, DE, DK, DM, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL,

APPLICATION NO. DATE

KIND DATE

PATENT NO.

```
IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW, AM, AZ,
             BY, KG, KZ, MD, RU, TJ, TM
         RW: GH, GM, KE, LS, MW, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE,
             DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF,
             CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
                                                               20000413
                                             EP 2000-923378
                        A1
                             20020724
     EP 1224226
            AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, SI, LT, LV, FI, RO, MK, CY, AL
                             19990414
PRAI US 1999-129145P
                       Ρ
     WO 2000-US10134
                        W
                             20000413
     A crosslinking aqueous solution (viscosity 50-20,000 cP) superabsorbent polymer
AΒ
     (SAP) composition consists of .apprx.15-50% \geq1 water-soluble monomer,
     preferably an \alpha,\beta-ethylenically unsatd. carboxylic acid monomer
     and a crosslinking agent. The polymer solution has low application
     viscosity, yet after crosslinking possesses a fast rate of acquisition and
     high absorption capacity. The SAP enhances the absorbency of various
     articles, increasing the humectancy and/or absorbency of a fiber or
     fibrous matrix, improving the H2O retention of soil and other agricultural
     methods, and increasing the open time of cement. Thus, Bacote
     20-crosslinked aqueous polyacrylic acid solution (viscosity 100-2000 cP) had
     water absorbency rating 9 (0 = worst, 10 = best).
     C08F008-00; A61L015-00
IC
     37-3 (Plastics Manufacture and Processing)
CC
     cement additive superabsorbent polymer; fiber improved
ST
     absorbency superabsorbent polymer; soil amendment superabsorbent polymer;
     diaper component superabsorbent polymer; crosslinked polyacrylate
     superabsorbent
     109665-04-5P, Acrylic acid-Neocryl CX 100 copolymer
IT
     302543-59-5P, Acrylic acid-Bacote 20 copolymer
     RL: IMF (Industrial manufacture); TEM (Technical or engineered
     material use); PREP (Preparation); USES (Uses)
        (aqueous superabsorbent polymer solution for use in agricultural and
consumer
        goods)
     109665-04-5P, Acrylic acid-Neocryl CX 100 copolymer
IT
     RL: IMF (Industrial manufacture); TEM (Technical or engineered
     material use); PREP (Preparation); USES (Uses)
        (aqueous superabsorbent polymer solution for use in agricultural and
consumer
        goods)
     109665-04-5 HCAPLUS
RN
     1-Aziridinepropanoic acid, 2-methyl-, 2-ethyl-2-[[3-(2-methyl-1-
CN
     aziridinyl)-1-oxopropoxy]methyl]-1,3-propanediyl ester, polymer with
     2-propenoic acid (9CI) (CA INDEX NAME)
     CM
          1
     CRN 64265-57-2
     CMF C24 H41 N3 O6
```

Me CH₂-CH₂-C-O-CH₂-C-CH₂-O-C-CH₂-CH₂-N Me CH₂

$$\begin{array}{c|c} CH_2 & CH_$$

CRN 79-10-7 CMF C3 H4 O2

RE.CNT 7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L60 ANSWER 13 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 2000:686296 HCAPLUS

DN 133:267265

TI Water-soluble or water-dispersible polymer salts and their use in cosmetic and pharmaceutical formulations

IN Nguyen, Kim Son; Sanner, Axel; Hossel, Peter

PA BASF Aktiengesellschaft, Germany

SO Eur. Pat. Appl., 31 pp.

CODEN: EPXXDW

DT Patent

LA German

FAN. CNT 1

PAN.	JNT.	T																	
	PAT	CENT	NO.		KI	ND	DATE			ΑI	PLI	CATI	ON NC	ο.	DATE				
PI	ΕP	1038	891		A	2	2000	0927		E	20	00-1	0647	0	2000	0324			
	ΕP	1038	891		A.	3	2001	0801											
	ΕP	1038	891		В	1	2003	0122											
		R:	ΑT,	BE,	CH,	DE,	DK,	ES,	FR,	GB,	GR,	IT,	LI,	LU,	NL,	SE,	MC,	PT,	
			IE,	SI,	LT,	LV,	FΙ,	RO											
	DE	1991	.3875		A	1	2000	0928		DE	19	99-1	9913	875	1999	0326			
	US	6407	158		В	1	2002	0618		US	20	00-5	3146	9	2000	0320			
	JP	2000	3028	37	A.	2	2000	1031		JI	20	00-83	2459		2000	0323			
	ΑТ	2315	25		F.		2003	0215		Αſ	2.0	00 - 10	0647	0	2000	0324			

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ES 2191580
                         Т3
                              20030916
                                             ES 2000-106470
                                                              20000324
       CN 1269377
                         Α
                              20001011
                                             CN 2000-104817
                                                              20000327
 PRAI DE 1999-19913875 A
                              19990326
      The salts, especially useful in hair sprays, consist of a polymer with free
      amino or acid groups and, resp., compds. with \geq 2 acid (or a
      polybasic inorg. acid) or amino groups, where the latter compound also
      contains a hydrophilic group. Thus, a polyester diol (from adipic acid,
      1,6-hexanediol, and isophthalic acid) 1.0, neopentyl glycol 1.2,
      dimethylolpropionic acid 2.7, and IPDI 5.0 mol were polymerized to give a
      carboxy group-containing polyurethane, which was neutralized with
      N-methyldipropylenetriamine.
 IC
      ICM C08F220-00
      35-8 (Chemistry of Synthetic High Polymers)
 CC
      Section cross-reference(s): 62, 63
 IT
      Polysiloxanes, properties
      RL: BUU (Biological use, unclassified); PRP (Properties); BIOL (Biological
      study); USES (Uses)
         (polyamine-polyoxyalkylene-, block; water-soluble or
         water-dispersible polymer salts for use in cosmetic and pharmaceutical
         formulations)
 TΤ
      Polyoxyalkylenes, properties
      RL: BUU (Biological use, unclassified); PRP (Properties); BIOL (Biological
      study); USES (Uses)
         (polyamine-polysiloxane-, block; water-soluble or
         water-dispersible polymer salts for use in cosmetic and pharmaceutical
         formulations)
 IT
     Polyurethanes, preparation
     RL: BUU (Biological use, unclassified); IMF (Industrial manufacture); PRP
      (Properties); BIOL (Biological study); PREP (Preparation); USES (Uses)
         (polyester-, block, carboxy-containing, salts with polyamines;
        water-soluble or water-dispersible polymer salts for use in cosmetic and
        pharmaceutical formulations)
TΤ
     Polyurethanes, preparation
     RL: BUU (Biological use, unclassified); IMF (Industrial manufacture); PRP
     (Properties); BIOL (Biological study); PREP (Preparation); USES (Uses)
         (polyoxyalkylene-, block, amino-containing, salts with polybasic
        acids; water-soluble or water-dispersible polymer salts for use in
        cosmetic and pharmaceutical formulations)
     Polysiloxanes, properties
IΤ
     Polysiloxanes, properties
     RL: BUU (Biological use, unclassified); PRP (Properties); BIOL (Biological
     study); USES (Uses)
        (polyoxyalkylene-, graft, amino group-containing; water-soluble or
        water-dispersible polymer salts for use in cosmetic and pharmaceutical
        formulations)
IT
     Polyamines
     RL: BUU (Biological use, unclassified); PRP (Properties); BIOL (Biological
     study); USES (Uses)
        (polyoxyalkylene-polysiloxane-, block; water-soluble or
        water-dispersible polymer salts for use in cosmetic and pharmaceutical
        formulations)
IT
     Polyoxyalkylenes, properties
       Polyoxyalkylenes, properties
    RL: BUU (Biological use, unclassified); PRP (Properties); BIOL (Biological
     study); USES (Uses)
        (polysiloxane-, graft, amino group-containing; water-soluble or
       water-dispersible polymer salts for use in cosmetic and pharmaceutical
        formulations)
```

```
154838-98-9DP, Butyl acrylate-tert-butyl acrylate-methacrylic acid
IT
     copolymer, compound with Dinoramox S 7 292621-96-6DP, Adipic
     acid-dimethylolpropionic acid-1,6-hexanediol-IPDI-isophthalic
     acid-neopentyl glycol block copolymer, compound with Dinoramox S 7
     297168-78-6P, Adipic acid-dimethylolpropionic acid-1,6-hexanediol-IPDI-
     isophthalic acid-neopentyl glycol block copolymer
                                                       297168-81-1P
    N-methyldipropylenetriamine salt 297168-79-7P
     297168-82-2P 297168-83-3P, Butyl acrylate-tert-butyl
     acrylate-methacrylic acid copolymer N-methyldipropylenetriamine
                        297168-85-5P
                                        297168-86-6P
     salt 297168-84-4P
     297168-90-2P
    RL: BUU (Biological use, unclassified); IMF (Industrial
    manufacture); PRP (Properties); BIOL (Biological study); PREP
     (Preparation); USES (Uses)
        (water-soluble or water-dispersible polymer salts for use in cosmetic and
       pharmaceutical formulations)
     297168-83-3P, Butyl acrylate-tert-butyl acrylate-methacrylic acid
ΙT
     copolymer N-methyldipropylenetriamine salt 297168-84-4P
     RL: BUU (Biological use, unclassified); IMF (Industrial
     manufacture); PRP (Properties); BIOL (Biological study); PREP
     (Preparation); USES (Uses)
        (water-soluble or water-dispersible polymer salts for use in cosmetic and
       pharmaceutical formulations)
     297168-83-3 HCAPLUS
RN
     2-Propenoic acid, 2-methyl-, polymer with butyl 2-propenoate and
CN
     1,1-dimethylethyl 2-propenoate, compd. with N-(2-aminomethylethyl)-1,2-
     propanediamine N-methyl deriv. (9CI) (CA INDEX NAME)
     CM
         1
         11071-12-8
     CRN
         C7 H19 N3
     CMF
     CCI IDS
H2N-CH2-CH2-NH-CH2-CH2-NH2
         3 (D1-Me)
     CM
         2
         154838-98-9
     CRN
          (C7 H12 O2 . C7 H12 O2 . C4 H6 O2)x
     CMF
     CCI
         PMS
          CM
               3
         CRN 1663-39-4
          CMF C7 H12 O2
```

CRN 141-32-2 CMF C7 H12 O2

CM 5

CRN 79-41-4 CMF C4 H6 O2

$$^{\text{CH}_2}_{||}$$

Me-C-CO₂H

RN 297168-84-4 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, polymer with butyl 2-propenoate and 1,1-dimethylethyl 2-propenoate, compd. with α,α' -[[(9Z)-9-octadecenylimino]di-2,1-ethanediyl]bis[ω -hydroxypoly(oxy-1,2-ethanediyl)] (9CI) (CA INDEX NAME)

CM 1

CRN 26635-93-8

CMF (C2 H4 O)n (C2 H4 O)n C22 H45 N O2

CCI PMS

CRN 154838-98-9

CMF (C7 H12 O2 . C7 H12 O2 . C4 H6 O2) \times

CCI PMS

CM 3

CRN 1663-39-4 CMF C7 H12 O2

0 \parallel t-BuO-C-CH=CH2

CM

CRN 141-32-2 CMF C7 H12 O2

0 $n-BuO-C-CH=CH_2$

> CM 5

CRN 79-41-4 CMF C4 H6 O2

CH₂ Me-C-CO2H

L60 ANSWER 14 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

2000:631920 HCAPLUS AN

DN 133:209050

Manufacture of crosslinked polyurethane foams for automobile parts with ΤI excellent mechanical properties

Watanabe, Hiroyuki; Satake, Shuichi IN

Sanyo Chemical Industries, Ltd., Japan PA

SO Jpn. Kokai Tokkyo Koho, 17 pp.

CODEN: JKXXAF

DT Patent

LΑ Japanese

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE

```
JP 1999-375723
                                                            19991228
    JP 2000248033
                      A2
                            20000912
PΙ
                            19981229
PRAI JP 1998-377241
                      Α
    The foams, useful for shock absorbers, steering wheels, seat cushions, and
    headrests, are manufactured by reaction of addition polymerizable active H
compds.
     and organic polyisocyanates in the presence of polymerization initiators
(containing
    peroxides, transition metal compds., and chelating agents and/or reducing
     agents) and optionally other additives. Thus, a composition containing 100
parts
     glycerin dimethacrylate, 172.7 parts cruder MDI, disodium EDTA, Fyrol CEF
     (P-type fireproofing agent), Rongalit, and Percumyl H (cumene
     hydroperoxide) was molded into foams with compression strength 2.2 kg/cm2,
     thermal conductivity 0.0230 kcal/mh-°C, and good fire resistance.
     ICM C08G018-09
TC:
     ICS C08G018-22; C08G018-09; C08G101-00
     38-3 (Plastics Fabrication and Uses)
CC
     Polyurethanes, uses
     Polyurethanes, uses
     Polyurethanes, uses
     RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP
     (Properties); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (acrylic-polyamine-polyoxyalkylene-; manufacture of
        crosslinked polyurethane foams for automobile parts with good mech.
       properties)
IT
     Polyoxyalkylenes, uses
       Polyoxyalkylenes, uses
       Polyoxyalkylenes, uses
     RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP
     (Properties); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (acrylic-polyamine-polyurethane-; manufacture of crosslinked
        polyurethane foams for automobile parts with good mech. properties)
IT
     Polyamines
       Polyamines
       Polyamines
     RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP
     (Properties); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (acrylic-polyoxyalkylene-polyurethane-; manufacture of crosslinked
        polyurethane foams for automobile parts with good mech. properties)
IT
     Polyurethanes, uses
     RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP
     (Properties); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (polyoxyalkylene-, acrylic; manufacture of crosslinked
        polyurethane foams for automobile parts with good mech. properties)
     101-68-8DP, MDI, urethane-modified, reaction products with glycerin
IT
     monomethacrylate and polyoxyalkylene
                                            50853-28-6DP, Glycerin
     monomethacrylate, reaction products with polyoxyalkylene and
     urethane-modified MDI
                             107498-00-0DP, Ethylene oxide-propylene oxide
     block copolymer glycerin ether, reaction products with glycerin
     monomethacrylate and urethane-modified MDI
                                                  139385-69-6P, Crude
                                             289705-87-9P, Crude
     MDI-glycerin dimethacrylate copolymer
     MDI-ethylene oxide-propylene oxide block copolymer glycerin
     ether-glycerin monomethacrylate-TDI copolymer
     289705-88-0P, Crude MDI-diethylene glycol monoacrylate-ethylene
```

oxide-propylene oxide block copolymer glycerin ether-triethanolamine copolymer RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (manufacture of crosslinked polyurethane foams for automobile parts with good mech. properties) 289705-88-0P, Crude MDI-diethylene glycol monoacrylate-ethylene oxide-propylene oxide block copolymer glycerin ether-triethanolamine copolymer (Preparation); USES (Uses)

RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PREP

(manufacture of crosslinked polyurethane foams for automobile parts with good mech. properties)

289705-88-0 HCAPLUS RN

2-Propenoic acid, 2-(2-hydroxyethoxy)ethyl ester, polymer with CN methyloxirane block polymer with oxirane ether with 1,2,3-propanetriol (3:1), 2,2',2''-nitrilotris[ethanol] and polymethylenepolyphenylene isocyanate (9CI) (CA INDEX NAME)

CM 1

ΙT

CRN 13533-05-6 CMF C7 H12 O4

2 CM

CRN 9016-87-9 CMF Unspecified CCI PMS, MAN

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

CM 3

CRN 102-71-6 CMF C6 H15 N O3

CM 4

107498-00-0 CRN C3 H8 O3 . 3 (C3 H6 O . C2 H4 O)x CMF

PEZZUTO 10/025588 10/28/03 Page 62

CM 5

CRN 56-81-5 CMF C3 H8 O3

 $\begin{array}{c} \text{OH} \\ | \\ \text{HO-} \, \text{CH}_2\text{--} \, \text{CH-} \, \text{CH}_2\text{--} \, \text{OH} \end{array}$

CM 6

CRN 106392-12-5

CMF (C3 H6 O . C2 H4 O) \times

CCI PMS

CM 7

CRN 75-56-9 CMF C3 H6 O



CM 8

CRN 75-21-8 CMF C2 H4 O



L60 ANSWER 15 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 2000:181937 HCAPLUS

DN 133:22373

TI Synthesis and NMR characterization of methacrylate copolymers with potential uses as bone **cements**

AU Martinez-Richa, Antonio; Cauich-Rodriguez, Juan V.; Vera-Graziano, Ricardo

CS Facultad de Quimica, Universidad de Guanajuato, Guanajuato, 36050, Mex.

SO Polymeric Materials Science and Engineering (2000), 82, 17-18 CODEN: PMSEDG; ISSN: 0743-0515

PB American Chemical Society

DT Journal

LA English

AB Methacrylate copolymers which are compatible with hydroxylapatite and have potential applications as bone **cements** were prepared and characterized. Stereochem. structure, comonomer-sequence and end groups were assigned.

CC 63-7 (Pharmaceuticals)

Section cross-reference(s): 36, 37 methacrylate copolymer NMR bone cement prepn ST IT Medical goods (bone cements; preparation and NMR characterization and tacticity of methacrylate copolymers for bone cements) IT NMR spectroscopy Tacticity (preparation and NMR characterization and tacticity of methacrylate copolymers for bone cements) 9011-14-7P, Poly(methyl methacrylate) 25685-29-4P, Ethyl ΙT methacrylate-methyl methacrylate copolymer 141699-08-3P, Diethylaminoethyl methacrylate-ethyl methacrylate-methyl methacrylate copolymer RL: PRP (Properties); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses) (preparation and NMR characterization and tacticity of methacrylate copolymers for bone cements) 141699-08-3P, Diethylaminoethyl methacrylate-ethyl ΙT methacrylate-methyl methacrylate copolymer RL: PRP (Properties); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses) (preparation and NMR characterization and tacticity of methacrylate copolymers for bone cements) 141699-08-3 HCAPLUS RN 2-Propenoic acid, 2-methyl-, 2-(diethylamino)ethyl ester, polymer with CN ethyl 2-methyl-2-propenoate and methyl 2-methyl-2-propenoate (9CI) INDEX NAME) CM 1 105-16-8 CRN C10 H19 N O2 CMF H₂C 0 Me-C-C-O-CH2-CH2-NEt2 2 CMCRN 97-63-2 CMF C6 H10 O2 H₂C 0 Me-C-C-OEt

3

80-62-6

C5 H8 O2

CM

CRN

CMF

10/025588

PEZZUTO

10/28/03

Page 63

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RE.CNT 19 THERE ARE 19 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L60 ANSWER 16 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 1999:751670 HCAPLUS

DN 132:3912

TI Resin vehicles for asphalt emulsions and asphalt compositions therewith for ambient-temperature blending

IN Endo, Toshio; Isobe, Tomohisa; Iwamoto, Naohisa

PA Daicel Chemical Industries, Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 10 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE					
PI	JP 11323140	A2	19991126	JP 1999-60097	19990308					
PRAT	TP 1998-73257		19980306							

AB The vehicles comprise (A) H2O-soluble epoxy resins having ≥2 epoxy groups in a mol. and epoxy equivalent 100-2000 and/or their (meth)acrylates, (B) flexible epoxy resins, and (C) aliphatic polyamines, aromatic polyamines, alicyclic polyamines, polyamide-amines, and/or their modified products in their weight ratio of A:B = 5-100:95-0 and equivalent ratio of (A + B):C = 100:50-200. Thus, a composition of a 60% asphalt emulsion 3.00, Denacol EX 512 (polyglycerol polyglycidyl ether; epoxy equivalent 167) 3.00, diethylenetriamine 0.30, melaminesulfonic acid-type water reducing agent 1.20, and aggregate 100 parts showed improved strength, durability, and water resistance.

IC ICM C08L095-00

ICS C08G059-50; C09J195-00

CC 38-3 (Plastics Fabrication and Uses) Section cross-reference(s): 51, 58

IT Cement (construction material)

Paving materials

Water-resistant materials

(ambient-temperature-curable water-soluble epoxy resin vehicles for asphalt emulsions)

TΨ 133842-79-2P, Denacol EX 830-diethylenetriamine copolymer 193896-29-6P, Denacol EX 512-diethylenetriamine copolymer 250691-69-1P, Denacol EX 830 acrylate-diethylenetriamine copolymer 250708-70-4P, Denacol EX 512-diethylenetriamine-Epikote 871 copolymer Denacol EX 512-Denacol EX 931-diethylenetriamine copolymer 250708-74-8P, Denacol EX 512 acrylate-diethylenetriamine copolymer 250708-75-9P, Denacol EX 512-diethylenetriamine-ethylenediamine copolymer 250708-76-0P, Denacol EX 512-hexamethylenediamine copolymer 250708-77-1P, Denacol EX 512-diaminodiphenylmethane copolymer 250708-78-2P, Denacol EX 512-EH 651 copolymer 250708-79-3P, Denacol EX 512-Laromin C copolymer 250708-80-6P, Denacol EX 512-Versamid 140 251297-80-0P, Bisphenol A diglycidyl ether-Denacol EX copolymer

512-diethylenetriamine copolymer 251297-81-1P, N-Aminoethylpiperazine-

```
Denacol EX 512-piperazine-triethanolamine copolymer
     RL: IMF (Industrial manufacture); MOA (Modifier or additive
     use); PRP (Properties); PREP (Preparation); USES (Uses)
        (ambient-temperature-curable water-soluble epoxy resin vehicles for asphalt
        emulsions)
     250691-69-1P, Denacol EX 830 acrylate-diethylenetriamine copolymer
IT
     250708-74-8P, Denacol EX 512 acrylate-diethylenetriamine copolymer
     RL: IMF (Industrial manufacture); MOA (Modifier or additive
     use); PRP (Properties); PREP (Preparation); USES (Uses)
        (ambient-temperature-curable water-soluble epoxy resin vehicles for asphalt
        emulsions)
     250691-69-1 HCAPLUS
RN
     1,2-Ethanediamine, N-(2-aminoethyl)-, polymer with \alpha-
CN
     (oxiranylmethyl) - \omega - (oxiranylmethoxy) poly (oxy-1, 2-ethanediyl)
     homopolymer 2-propenoate (9CI) (CA INDEX NAME)
     CM
          1
     CRN
         111-40-0
     CMF C4 H13 N3
H2N-CH2-CH2-NH-CH2-CH2-NH2
     CM
          2
         104220-34-0
     CRN
          C3 H4 O2 . x ((C2 H4 O)n C6 H10 O3)x
               3
          CM
              79-10-7
          CRN
          CMF C3 H4 O2
   0
но- с- сн сн2
          CM
               4
              58782-18-6
          CRN
          CMF
               ((C2 H4 O)n C6 H10 O3)x
          CCI
               PMS
               CM
                    5
                   26403-72-5
               CRN
               CMF
                    (C2 H4 O)n C6 H10 O3
               CCI
                    PMS
```

$$CH_2-O$$
 CH_2-CH_2-O CH_2 CH_2

RN 250708-74-8 HCAPLUS

CN Tetraglycerol, tetrakis(oxiranylmethyl) ether, homopolymer, 2-propenoate, polymer with N-(2-aminoethyl)-1,2-ethanediamine (9CI) (CA INDEX NAME)

CM 1

CRN 111-40-0 CMF C4 H13 N3

 $_{\rm H_2N-CH_2-CH_2-NH-CH_2-CH_2-NH_2}$

CM 2

CRN 250708-73-7

CMF (C24 H42 O13) \times . \times C3 H4 O2

CM 3

CRN 79-10-7 CMF C3 H4 O2

CM 4

CRN 105521-63-9

CMF (C24 H42 O13)x

CCI PMS

CM 5

CRN 112477-66-4

CMF C24 H42 O13

CCI IDS

CM 6

CRN 56491-53-3

CMF C12 H26 O9

CCI IDS, MAN

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

CM 7

CRN 556-52-5 CMF C3 H6 O2

сн2-он

L60 ANSWER 17 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

1999:705021 HCAPLUS AN

DN 131:323408

Formaldehyde-free, accelerated cure, aqueous composition for bonding heat-resistant nonwoven glass fibers

Arkens, Charles Thomas; Egolf, Scott Lind IN

PA Rohm and Haas Company, USA

U.S., 8 pp. SO

CODEN: USXXAM

ידים Patent

English LΑ

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE _____ _____ PI US 5977232 A 19991102 PRAI US 1997-904713 19970801 US 1997-904713 19970801

A formaldehyde-free curable aqueous composition comprises (a) a polyacid comprising

at least two carboxylic acid groups, anhydride groups, or salts thereof; (b) an active hydrogen compound containing at least two active hydrogen groups selected from hydroxyl, primary amino, secondary amino, and mixts. thereof; and (c) a fluoroborate accelerator; and (d) a phosphorus-containing accelerator, where the ratio of the number of equivalent of carboxylic acid groups, anhydride groups, or salts thereof to the number of equivalent of hydroxyl groups is 1:0.01-3, and the carboxylic acid groups, anhydride groups, or salts thereof are <35% neutralized with a fixed base. A method for curing formaldehyde-free polyacids and a method for bonding glass fiber heat-resistant nonwovens with a formaldehyde-free binder are further disclosed. The nonwovens are useful in insulation batts, reinforcing mat for roofing or flooring, battery separators, etc. (no data). Thus, a cured nonwoven sheet containing a composition comprising poly(acrylic acid)

50.11,

triethanolamine 7.07, sodium hypophosphite (I) 0.18, sodium fluoroborate (II) 0.15, Z 6040 14.92, and water 227.54 g showed dry tensile strength 9.7 lb/in., wet tensile strength 7.3 lb/in, and retention ($W/D \times 100$) 75.3%, compared with 10.6, 2.6, and 24.5, resp., without II, and 10.3, 5.5, and 53.4, resp., without I.

ICM C08K003-38 IC

ICS C08K003-16

NCL 524404000

CC 37-6 (Plastics Manufacture and Processing)

ST polycarboxylic acid compn curing formaldehyde free; polyacrylic acid curing formaldehyde free; fluoroborate accelerator curing polycarboxylic acid compn; phosphorus accelerator curing polycarboxylic acid compn; sodium hypophosphite curing accelerator polycarboxylic acid; glass fiber nonwoven bonding formaldehyde free

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PEZZUTO

IT 52
RI
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IT 52
RI
ma

RN 52
CN 2-
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52880-57-6P, Poly(acrylic acid) triethanolamine salt
RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(formaldehyde-free, accelerated cure, aqueous composition for

(formaldehyde-free, accelerated cure, aqueous composition for bonding heat-resistant nonwoven glass fibers)

IT 52880-57-6P, Poly(acrylic acid) triethanolamine salt

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(formaldehyde-free, accelerated cure, aqueous composition for bonding heat-resistant nonwoven glass fibers)

RN 52880-57-6 HCAPLUS

CN 2-Propenoic acid, homopolymer, compd. with 2,2',2''-nitrilotris[ethanol] (9CI) (CA INDEX NAME)

CM 1

CRN 102-71-6 CMF C6 H15 N O3

$$_{\rm CH_2-CH_2-OH}$$
 $_{\rm HO-CH_2-CH_2-N-CH_2-CH_2-OH}$

CM 2

CRN 9003-01-4 CMF (C3 H4 O2)x CCI PMS

CM 3

CRN 79-10-7 CMF C3 H4 O2

RE.CNT 15 THERE ARE 15 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L60 ANSWER 18 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 1999:629948 HCAPLUS

DN . 131:258429

Graft copolymers, their manufacture from polyalkylenimines, alkylene oxides, and unsaturated monomers, and their use as cement additives, scale inhibitors, and detergent compositions

IN Maita, Takeshi; Tanaka, Hiromichi; Nagare, Koichiro

PA Nippon Shokubai Kagaku Kogyo Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 12 pp. CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1 PATENT NO. KIND DATE APPLICATION NO. DATE JP 11269239 A2 ΡI 19991005 JP 1998-70695 19980319 PRAI JP 1998-70695 19980319 Graft polymers are manufactured by addition polymerization of polyalkylenimines with excess alkylene oxides (based on the active H-containing amino groups) to prepare polyamine-polyethers, followed by solventless graft polymerization with unsatd. monomers at 80-160° in the presence of organic peroxides. The graft polymers are used as additives for cement compns., scale inhibitors, and detergent compns. Thus, 194 g polyethylenimine polyethylene oxide adduct reacted with 14.0 g maleic acid and 20.2 g acrylic acid at 130° for 2 h to give an aqueous solution of a graft copolymer with Mn 4800 and Mw 16,000, which improved dispersion stability of portland cements, scale inhibition of Ca phosphate and Zn(OH)2 to glass containers, and detergency of surfactant-containing detergent compns. IC ICM C08F283-06 ICS C02F005-10; C04B024-26; C09K003-00; C11D003-37; C04B103-40 37-6 (Plastics Manufacture and Processing) Section cross-reference(s): 46, 58 polyalkylenimine polyoxyethylene unsatd monomer graft copolymer; cement dispersant polyalkylenimine polyoxyalkylene acrylic graft copolymer; scale inhibitor graft acrylic polyethylenimine polyoxyethylene; detergent graft acrylic polyethylenimine ethylene oxide adduct TΨ Dispersing agents (for cements; manufacture of graft copolymers from polyalkylenimines, alkylene oxides, and unsatd. monomers for use as cement additives, scale inhibitors, and detergent compns.)

Polymerization

(graft; manufacture of graft copolymers from polyalkylenimines, alkylene oxides, and unsatd. monomers for use as cement additives, scale inhibitors, and detergent compns.)

IT Detergents

Scale inhibitors

(manufacture of graft **copolymers** from polyalkylenimines, alkylene oxides, and unsatd. monomers for use as **cement** additives, scale inhibitors, and detergent compns.)

IT Polyoxyalkylenes, preparation

Polyoxyalkylenes, preparation

RL: IMF (Industrial manufacture); MOA (Modifier or additive use); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(polyamine-, acrylic, graft; manufacture of graft copolymers from polyalkylenimines, alkylene oxides, and unsatd. monomers for use as cement additives, scale inhibitors, and detergent compns.)

IT Polyamines

Polyamines '

RL: IMF (Industrial manufacture); MOA (Modifier or additive use); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(polyoxyalkylene-, acrylic, graft; manufacture of graft copolymers from polyalkylenimines, alkylene oxides, and unsatd. monomers for use as cement additives, scale inhibitors, and detergent compns.)

```
IT
     Cement (construction material)
        (portland; manufacture of graft copolymers from polyalkylenimines,
        alkylene oxides, and unsatd. monomers for use as cement
        additives, scale inhibitors, and detergent compns.)
     245053-05-8P, Acrylic acid-ethylene oxide-ethylenimine-maleic acid
IT
     graft copolymer
     RL: IMF (Industrial manufacture); PRP (Properties); TEM
     (Technical or engineered material use); PREP (Preparation); USES
     (Uses)
        (manufacture of graft copolymers from polyalkylenimines, alkylene
        oxides, and unsatd. monomers for use as cement additives,
        scale inhibitors, and detergent compns.)
     10103-46-5, Calcium phosphate
                                     20427-58-1, Zinc hydroxide
TΨ
     RL: MSC (Miscellaneous)
        (scale inhibitor for; manufacture of graft copolymers from
        polyalkylenimines, alkylene oxides, and unsatd. monomers for use as
        cement additives, scale inhibitors, and detergent compns.)
     98-11-3D, Benzenesulfonic acid, C12-16 alkyl derivs., sodium salts, uses
IT
     7664-93-9D, Sulfuric acid, higher alkyl esters, Na salts, uses
     RL: MOA (Modifier or additive use); USES (Uses)
        (surfactant in detergents; manufacture of graft copolymers from
        polyalkylenimines, alkylene oxides, and unsatd. monomers for use as
        cement additives, scale inhibitors, and detergent compns.)
     245053-05-8P, Acrylic acid-ethylene oxide-ethylenimine-maleic acid
TT
     graft copolymer
     RL: IMF (Industrial manufacture); PRP (Properties); TEM
     (Technical or engineered material use); PREP (Preparation); USES
        (manufacture of graft copolymers from polyalkylenimines, alkylene
        oxides, and unsatd. monomers for use as cement additives,
        scale inhibitors, and detergent compns.)
RN
     245053-05-8 HCAPLUS
CN
     2-Butenedioic acid (2Z)-, polymer with aziridine, oxirane and 2-propenoic
     acid, graft (9CI) (CA INDEX NAME)
     CM
          1
     CRN
        151-56-4
     CMF C2 H5 N
```



CRN 110-16-7 CMF C4 H4 O4

Double bond geometry as shown.

CRN 79-10-7 CMF C3 H4 O2

CM 4

CRN 75-21-8 CMF C2 H4 O



L60 ANSWER 19 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 1999:380817 HCAPLUS

DN 131:46270

TI Treatment for improving the surface strength and printability of paper

IN Taniquchi, Masahide; Takeuchi, Kunio; Ohwatari, Junya

PA Seiko Kagaku Kogyo Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 8 pp. CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

CC

PATENT NO. KIND DATE APPLICATION NO. DATE
PI JP 11158800 A2 19990615 JP 1997-364263 19971127
PRAI JP 1997-364263 19971127

AB The treatment is obtained from a mixture of (A) polyamide-polyamine-epichlorohydrin adducts, (B) (meth)acrylamide-based polymers and (C) polyalkylene glycols. Thus, heating diethylenetriamine 101 with adipic acid 146 in water 55 g at 170-180° while removing water for 4 h, then mixing the resulting polyaminamide with 138.9 g epichlorohydrin and 1530 g water and heating at 60-65° for 3 h gave an adduct. A paper surface treatment was obtained from the adduct, a copolymer of acrylamide and acrylic acid and a polyethylene glycol.

IC ICM D21H027-00 ICS D21H019-16

43-7 (Cellulose, Lignin, Paper, and Other Wood Products)

ST polyaminamide epichlorohydrin adduct paper surface treatment;

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polyalkylenepolyamine polycarboxylic acid amide paper surface
     treatment; polyalkylene glycol paper surface treatment; size surface paper
     polyaminamide epichlorohydrin adduct compn
     25212-19-5P, Adipic acid-diethylenetriamine-epichlorohydrin copolymer
IT
     227011-77-0P, Adipic acid-diethylenetriamine-epichlorohydrin-
    itaconic acid copolymer 227011-78-1P, Adipic acid-diethylenetriamine-
     epichlorohydrin-glycerin triglycidyl ether copolymer
     RL: IMF (Industrial manufacture); POF (Polymer in formulation);
     PRP (Properties); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (surface treatment; compns. containing acrylamide based polymers for
        improving surface strength and printability of paper)
     227011-77-0P, Adipic acid-diethylenetriamine-epichlorohydrin-
ΤТ
     itaconic acid copolymer
     RL: IMF (Industrial manufacture); POF (Polymer in formulation);
     PRP (Properties); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (surface treatment; compns. containing acrylamide based polymers for
        improving surface strength and printability of paper)
     227011-77-0 HCAPLUS
RN
     Hexanedioic acid, polymer with N-(2-aminoethyl)-1,2-ethanediamine,
CN
     (chloromethyl) oxirane and methylenebutanedioic acid (9CI) (CA INDEX NAME)
     CM
          1
     CRN
         124-04-9
     CMF C6 H10 O4
HO_2C-(CH_2)_4-CO_2H
     CM
          2
        111-40-0
     CRN
     CMF C4 H13 N3
H2N-CH2-CH2-NH-CH2-CH2-NH2
     CM
          3
     CRN 106-89-8
     CMF C3 H5 C1 O
     CH2-Cl
     CM
```

CRN

97-65-4

CMF C5 H6 O4

```
СH<sub>2</sub>
||
HO<sub>2</sub>C-С-СH<sub>2</sub>-СО<sub>2</sub>H
```

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L60 ANSWER 20 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN
AN
    1999:311270 HCAPLUS
    130:339501
DN
    Printing ink compositions containing core-shell binders and additives for
TΙ
    image film having superior smear-fastness
    Nguyen, Khe C.; Ganapathiappan, Sivapackia
IN
    Hewlett-Packard Company, USA
PΑ
    PCT Int. Appl., 70 pp.
    CODEN: PIXXD2
DT
    Patent
LΑ
    English
FAN.CNT 8
                 . KIND DATE
                                         APPLICATION NO. DATE
    PATENT NO.
     _____
                    ____
                                         _____
                    A1 19990514 WO 1998-US23474 19981029
    WO 9923183
PΙ
        W: AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE,
            DK, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IS, JP, KE,
            KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW,
            MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR,
            TT, UA, UG, US, UZ, VN, YU, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
        RW: GH, GM, KE, LS, MW, SD, SZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES,
            FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI,
            CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
                                        US 1997-998164
                          19991123
                                                          19971224
                    Α
    US 5990202
    US 6417249
                           20020709
                                         US 1998-138772
                                                          19980824
                      В1
    AU 9913796
                                         AU 1999-13796
                      A1
                           19990524
                                                          19981029
                           20000816
                                         EP 1998-957567
                                                          19981029
    EP 1027393
                      A1
        R: DE, FR, GB, IT
                    Т2
                                         JP 2000-519048
                                                         19981029
    JP 2001521977
                           20011113
PRAI US 1997-962496
                     Α
                           19971031
                    A .
                           19971224
    US 1997-998164
    US 1998-138772
                     Α
                           19980824
    WO 1998-US23474
                      W
                           19981029
    MARPAT 130:339501
OS
    Core/shell binders such as [AmBnC'p]x are prepared, where A and B are
ΑB
    hydrophobic components in which A exhibits a glass transition temperature Tg
    -150° to 25° and B exhibits a Tg >25°, C' forms a
    hydrophilic or water-soluble component and has an ionic or nonionic
    structure, m <30%, n >40%, and p <30%, m + n + p = 100%, and x =
    1-100,000, and the weight-average mol. weight .apprx.1000-2,000,000.
binder
    polymer is used in conjunction with additives comprising either (a) amine
    alcs. R1R2N(RX)OH(R1, R2 = H, alkyl, alkoxy, aryl, and phenoxy, R =
    alkyl, X = H, alkyl, aryl, OH, CO2H, CHO, and substituted groups) or (b)
    organic acids (water-soluble or water-dispersive), including polymeric acids,
    optionally amines, polyalcs., polyamines, and polyesters, and the
    binder/colorant ratio ≥10. Thus, hexyl acrylate-Me
    methacrylate-polyethylene glycol Me ether acrylate copolymer
    emulsion, Et acetate (0.05%), and water was cast on glass and dried as a
```

acrylate-N-vinylmaleimide-polyethylene glycol methyl ether acrylate graft

224045-50-5P, Hexyl acrylate-methyl methacrylate-acrylic acid graft

copolymer 224045-48-1P, Hexyl acrylate-methyl
methacrylate-N-vinyl-4-methylpyrrolidone graft copolymer

224045-51-6P, Hexyl acrylate-methyl

```
methacrylate-methacrylic acid graft copolymer
                                               224045-52-7P,
Hexyl acrylate-methyl methacrylate-maleic acid graft copolymer
224045-53-8P, Hexyl acrylate-methyl methacrylate-vinylbenzoic acid graft
           224045-54-9P, Hexyl acrylate-methyl
methacrylate-vinylsulfonamide graft copolymer 224045-55-0P,
Hexyl acrylate-methyl methacrylate-sodium acrylate graft copolymer
224045-56-1P, Ethyl acrylate-methyl methacrylate-
(acrylamidopropyl) triethylammonium chloride graft copolymer
224045-57-2P, Hexyl acrylate-methyl methacrylate-ammonium acrylate graft
           224045-58-3P, Hexyl acrylate-methyl
methacrylate-ammonium methacrylate graft copolymer
224045-60-7P, Hexyl acrylate-methyl methacrylate-sodium styrenesulfonate
                 224045-61-8P, Methyl methacrylate-hexyl
graft copolymer
acrylate-polyethylene glycol methyl ether acrylate-acrylic acid graft
            224047-92-1P, Hydroxyoctadecyl acrylate-methyl
copolymer
methacrylate-polyethylene glycol methyl ether acrylate graft
           224047-99-8P, Propyl acrylate-methyl
copolymer
methacrylate-vinylpyridine hydrochloride graft copolymer
224048-41-3P, Hexyl acrylate-tetrafluoropropyl methacrylate-polyethylene
glycol methyl ether acrylate graft copolymer
                                              224184-44-5P,
Lauryl methacrylate-methyl methacrylate-polyethylene glycol methyl ether
acrylate graft copolymer 224184-45-6P, Octadecyl
methacrylate-methyl methacrylate-polyethylene glycol methyl ether acrylate
                224184-46-7P, Hydroxylauryl
graft copolymer
methacrylate-methyl methacrylate-polyethylene glycol methyl ether acrylate
graft copolymer 224184-47-8P, 2-Aminopropyl acrylate-methyl
methacrylate-polyethylene glycol methyl ether acrylate graft
           224184-48-9P, 6-Aminohexyl acrylate-methyl
copolymer
methacrylate-polyethylene glycol methyl ether acrylate graft
          224184-50-3P, 12-Aminolauryl methacrylate-methyl
copolymer
methacrylate-polyethylene glycol methyl ether acrylate graft
          224184-51-4P, Hexyl acrylate-trifluoromethylstyrene-
copolymer
polyethylene glycol methyl ether acrylate graft copolymer
224184-52-5P, Hexyl acrylate-methyl methacrylate-1-vinyl-2-pyrrolidone
                224184-53-6P, Hexyl acrylate-methyl
graft copolymer
methacrylate-vinylimidazole graft copolymer 224184-54-7P,
Hexyl acrylate-methyl methacrylate-2-methylacrylamide graft
copolymer 224184-61-6P, Butyl acrylate-methyl
methacrylate-sodium vinyl phosphate graft copolymer
RL: SPN (Synthetic preparation); TEM (Technical or engineered
material use); PREP (Preparation); USES (Uses)
   (core shell; printing ink compns. containing core-shell binders and
   additives for image film having superior smear-fastness and water
   fastness)
                             57-55-6, 1,2-Propanediol, uses
                                                               77-92-9,
57-10-3, Palmitic acid, uses
                              96-80-0 104-15-4, uses
Citric acid, uses
                    96-20-8
                                  107-21-1, 1,2-Ethanediol, uses
107-15-3, Ethylenediamine, uses
110-15-6, Butanedioic acid, uses
                                 110-85-0, Piperazine, uses
                   110-97-4, Diisopropanolamine
                                                 111-55-7, Ethylene
Piperidine, uses
                                                 123-31-9,
                   123-25-1, Diethyl succinate
glycol diacetate
1,4-Benzenediol, uses
                       124-40-3, Dimethylamine, uses
                                                       141-43-5, uses
616-47-7, 1-Methylimidazole 628-67-1, Butylene glycol diacetate
                                2454-37-7, 3-(1-Hydroxyethyl)aniline
1484-84-0, 2-Piperidineethanol
3040-44-6, 1-Piperidineethanol
                               3179-63-3
                                             3433-37-2,
                       4606-65-9, 3-Piperidinemethanol
                                                         5138-18-1,
2-Piperidinemethanol
                   6425-32-7, 3-Morpholino-1,2-propanediol
                                                              7209-38-3,
Sulfosuccinic acid
1,4-Bis(3-aminopropyl) piperazine 9002-89-5, Poly(vinyl alcohol)
            9003-01-4, Polyacrylic acid 9003-13-8 9003-20-7,
9002-98-6
```

ΙT

```
9004-74-4
                                       9046-31-5, Poly(vinylbenzoic acid)
     Poly(vinyl acetate)
     25087-26-7, Polymethacrylic acid 25265-75-2, Butylene glycol
                                        25322-68-3
                                                    25322-69-4, Polypropylene
     25321-41-9, Xylene sulfonic acid
                                        26062-79-3, Poly
             25736-86-1
                           25988-97-0
     glycol
                                          26403-58-7 26762-52-7, Hexanediol
     (diallyldimethylammonium chloride)
     28574-59-6, Poly(dimethylaminoethyl acrylate) 37286-64-9, Polypropylene
     glycol methyl ether 39420-45-6, Polypropylene glycol monomethacrylate
     50858-51-0, Polypropylene glycol monoacrylate 70378-21-1 98961-84-3
     116770-99-1, Aziridine-ethylene oxide graft copolymer
     154976-19-9
                  201989-32-4
                                224322-41-2
     RL: MOA (Modifier or additive use); USES (Uses)
        (printing ink compns. containing colorant, core-shell binders and additives
        for image film having superior smear-fastness and water fastness)
                 25852-37-3, Butyl acrylate-methyl methacrylate
IT
     9004-98-2
                 35545-57-4, Solsperse 27000
     copolymer
     RL: POF (Polymer in formulation); TEM (Technical or engineered material
     use); USES (Uses)
        (printing ink compns. containing core-shell binders and additives for image
        film having superior smear-fastness and water fastness)
     224045-37-8P, N,N-Dihexylacrylamide-methyl methacrylate-
IT
     polyethylene glycol methyl ether acrylate graft copolymer
     RL: SPN (Synthetic preparation); TEM (Technical or engineered
     material use); PREP (Preparation); USES (Uses)
        (core shell; printing ink compns. containing core-shell binders and
        additives for image film having superior smear-fastness and water
        fastness)
RN
     224045-37-8 HCAPLUS
CN
     2-Propenoic acid, 2-methyl-, methyl ester, polymer with
     N,N-dihexyl-2-propenamide and \alpha-(1-oxo-2-propenyl)-\omega-
     methoxypoly(oxy-1,2-ethanediyl), graft (9CI) (CA INDEX NAME)
     CM
          1
     CRN 35143-36-3
     CMF C15 H29 N O
Me-(CH_2)_5-N-(CH_2)_5-Me
     CM
          2
     CRN
          32171-39-4
     CMF
          (C2 H4 O)n C4 H6 O2
     CCI
          PMS
```

H₂C== CH-C O-CH₂-CH₂-

CRN 80-62-6 CMF C5 H8 O2

H₂C O $\parallel \parallel$ Me-C-C-OMe

RE.CNT 7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L60 ANSWER 21 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

1998:314742 HCAPLUS

DN 129:28952

Cationized polymer emulsions and their preparation ΤI IN

Arai, Takeo; Tamura, Akira; Miyata, Masanori; Okita, Yasuo PA

Mitsui Cyanamid K. K., Japan

Jpn. Kokai Tokkyo Koho, 7 pp. SO CODEN: JKXXAF

Patent T.A

Japanese

FAN.CNT 1

DT

PATENT NO. KIND DATE APPLICATION NO. DATE ---------PRAI JP 1996-306968 1006110 JP 1996-306968 19961101

Title emulsions having highly cationized particle surface and reduced agglomerates, useful for paper strengthening agents, mortar and cement mixing agents, etc., are prepared by neutralizing seed latexes of carboxy-modified synthetic rubbers or resin emulsions to pH \geq 6, adding H20-insol. or hardly soluble monomers CH2:CR1AR2NR3R4 (R1 = H, Me; R2 = C2-5 alkylene; R3, R4 = H, C1-5 alkyl; A = CO2, CONH) in the latexes at an amount greater than the colloid equivalent (absolute value) of the

latexes, radically polymerizing the monomers, adding nonionic surfactants in the resulting seed polymers, and neutralizing the polymers with acids or bases or quaternizing amino groups. Thus, carboxy-modified SBR latex 701.3 (pH 8.3, 48% solid, anion colloid equivalent -0.18 mequiv/g), N,N'-methylenebisacrylamide 0.4, H2O 159.7, and diethylaminoethyl methacrylate (I) $37.\overline{5}$ g were treated with K2S2O8 at 50° for 2 h to obtain polymer emulsion, which was then mixed with 0.5% (solid) Emulgen 840S (nonionic surfactant) and quaternized with equivalent amount (vs. I) of Me2SO4 to give emulsion containing 0.032% agglomerates.

ICM C08F008-44

ICS C08F002-22; C08F279-02

38-3 (Plastics Fabrication and Uses) CC Section cross-reference(s): 43, 58

Cement (construction material)

(mixing agents; manufacture of cationized polymer emulsions with reduced agglomerates)

ΙT 123723-36-4P 208103-13-3P 208103-14-4P RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (manufacture of cationized polymer emulsions with reduced agglomerates) IT 123723-36-4P 208103-14-4P

RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES

(manufacture of cationized polymer emulsions with reduced agglomerates)

RN 123723-36-4 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-(diethylamino)ethyl ester, polymer with N,N'-methylenebis[2-propenamide], compd. with dimethyl sulfate (9CI) (CA INDEX NAME)

CM 1

CRN 77-78-1 CMF C2 H6 O4 S

CM 2

CRN 123723-35-3

CMF (C10 H19 N O2 . C7 H10 N2 O2)x

CCI PMS

CM 3

CRN 110-26-9 CMF C7 H10 N2 O2

CM 4

CRN 105-16-8 CMF C10 H19 N O2

RN 208103-14-4 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-(diethylamino)ethyl ester, polymer with N,N'-methylenebis[2-propenamide], hydrochloride (9CI) (CA INDEX NAME)

PEZZUTO 10/025588 10/28/03 Page 79 CM CRN 123723-35-3 CMF (C10 H19 N 0.2 . C7 H10 N2 0.2) x CCI PMS 2 CM CRN 110-26-9 CMF C7 H10 N2 O2 0 H2C == CH - C - NH - CH2 - NH - C - CH == CH2 CM CRN 105-16-8 CMF C10 H19 N O2 H₂C 0 Me-C-C-O-CH2-CH2-NEt2 L60 ANSWER 22 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN 1998:256346 HCAPLUS AN 129:16782 DN Dispersing agents useful for stable inorganic pigments with low viscosity TIKyuda, Nobuo; Ida, Yoshimi; Yamauchi, Sunao ΙN PA Sanyo Chemical Industries, Ltd., Japan SO Jpn. Kokai Tokkyo Koho, 11 pp. CODEN: JKXXAF Patent DT LΑ Japanese FAN.CNT 1 KIND DATE APPLICATION NO. DATE PATENT NO. ____ ----------JP 10110015 A2 19980428 JP 1996-286078 19961007 ΡI PRAI JP 1996-286078 19961007 The dispersants comprise (a) 10-99.9 mol% (based on acid groups) alkali metal salts or alkaline earth metal salts of (co)polymers mainly containing α,β -unsatd. carboxylic acids, (b) 0.01-50 mol% ≥ 1 of the (co)polymer salts selected from quaternary ammonium salts, ammonium, and organic amines, and (c) 0-89.9 mol% free acid groups. Thus, acrylic acid (I) was polymerized and neutralized with NaOH and triethylmethylammonium Me carbonate to give I homopolymer salts, 0.6 part of which were dispersed in 75 parts Ca(HCO3)2. The obtained aqueous slurry showed viscosity 230 mPa-s at 25° and 60 rpm and 2050 mPa-s after 7 days storage at 25°. ICM C08F020-04 IC ICS C08F022-02 CC 37-6 (Plastics Manufacture and Processing) Carboxylic acids, preparation

```
RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM
     (Technical or engineered material use); PREP (Preparation); USES (Uses)
        (polycarboxylic, ammonium salts; acrylic (co)polymer salt
        mixts. as inorg. pigment dispersants)
IT
     Carboxylic acids, preparation
     RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM
     (Technical or engineered material use); PREP (Preparation); USES (Uses)
        (polycarboxylic, salts; acrylic (co)polymer salt mixts. as
        inorg. pigment dispersants)
     9003-01-4P, Acrylic acid homopolymer
                                            9003-03-6P, Poly(acrylic acid)
IT
                     9003-04-7P, Acrylic acid homopolymer sodium salt
     ammonium salt
     25608-12-2P, Poly(acrylic acid) potassium salt
                                                     25608-13-3P, Poly(acrylic
     acid) tetramethylammonium salt, preparation 27936-81-8P,
                                                           175344-08-8P,
     Poly(acrylic acid) diethanolamine salt
                                              41206-71-7P
                                  207723-94-2P
                                                 207723-95-3P
                                                                207723-98-6P,
                 207723-93-1P
     preparation
                   207723-99-7P, preparation
                                               207724-00-3P
     preparation
     RL: IMF (Industrial manufacture); POF (Polymer in formulation);
     TEM (Technical or engineered material use); PREP (Preparation);
     USES (Uses)
        (acrylic (co)polymer salt mixts. as inorg. pigment dispersants)
IT
     27936-81-8P, Poly(acrylic acid) diethanolamine salt
     RL: IMF (Industrial manufacture); POF (Polymer in formulation);
     TEM (Technical or engineered material use); PREP (Preparation);
     USES (Uses)
        (acrylic (co)polymer salt mixts. as inorg. pigment dispersants)
RN
     27936-81-8 HCAPLUS
CN
     2-Propenoic acid, homopolymer, compd. with 2,2'-iminobis[ethanol] (9CI)
     (CA INDEX NAME)
     CM
          1
     CRN 111-42-2
     CMF C4 H11 N O2
HO-CH2-CH2-NH-CH2-CH2-OH
     CM
          2
         9003-01-4
     CRN
          (C3 H4 O2)x
     CMF
     CCI
         PMS
          CM
               3
              79-10-7
         CRN
         CMF C3 H4 O2
   O
HO-C-CH=CH_2
```

KATHLEEN FULLER EIC 1700/PARKER LAW 308-4290

L60 ANSWER 23 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

```
1998:28722 HCAPLUS
AN
DN
     128:131520
    Admixtures for concrete and their use, and method for dispersing
TI
     cement with, and concrete containing, the admixtures
     Satoh, Haruyuki; Yamato, Fujio; Kono, Yoshinao; Nakamura, Sayuri
IN
     Kao Corp., Japan; Satoh, Haruyuki; Yamato, Fujio; Kono, Yoshinao;
PA
    Nakamura, Sayuri
     PCT Int. Appl., 42 pp.
SO
     CODEN: PIXXD2
DT
    Patent
LΑ
    English
FAN.CNT 1
                                           APPLICATION NO. DATE
     PATENT NO.
                            DATE
                      KIND
                      ____
                            19971224
                                           WO 1997-JP2095
                                                            19970618
    WO 9748656
PI
                       A1
        W: CN, US, VN
         RW: AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE
                                                            19970609
     JP 10081549
                       A2
                            19980331
                                           JP 1997-150709
                                           TW 1997-86108064 19970611
                            20000111
     TW 379208
                       В
                            19980610
                                           EP 1997-927384
                                                             19970618
     EP 846090
                       A1
     EP 846090
                      В1
                            20031001
        R: DE, FR, GB
                            19990615
                                           US 1998-29031
                                                             19980220
     US 5911820
                       Α
PRAI JP 1996-161287
                       Α
                            19960621
    WO 1997-JP2095
                       W
                            19970618
     The admixt. comprise a copolymer comprising, as structural units, units
     derived from an ethylenically unsatd. monomer (a) containing 25-300 mols
     C2-3-oxyalkylene groups, and units derived from a monomer (b) of an alkyl,
     alkenyl or hydroxyalkyl ester of an ethylenically unsatd. mono- or
     dicarboxylic acid. The admixts. are effective in imparting fluidity to
     hydraulic compns., e.g., cement pastes, mortar, and concrete,
     especially in maintaining fluidity of the compns., and hardly retard hardening
     of the compns. A cement composition containing 0.27% methoxypolyethylene
     glycol methacrylate-Me acrylate copolymer telomer with 2-mercaptoethanol
     gave slump after 120 min 22.5 cm and initial and final setting time 4 h 52
    min and 6 h 31 min, vs. 7.0 cm, and 7 h 20 min and 8 h 58 min for a
     control.
     ICM C04B024-26
IC
     58-2 (Cement, Concrete, and Related Building Materials)
CC
     acrylic polymer telomer concrete slump; cement dispersant
ST
     acrylic polymer telomer
IT
     Cement (construction material)
     Concrete
        (acrylic telomers as dispersants for)
                                                  201793-14-8P
                                                                  201793-16-0P
ΙT
     201793-10-4P 201793-11-5P
                                   201793-12-6P
                                                                  201793-24-0P
                                                  201793-23-9P
     201793-18-2P
                    201793-20-6P
                                   201793-21-7P
     201793-25-1P
                   201793-26-2P 201793-30-8P
                                                201872-78-8P
     RL: IMF (Industrial manufacture); SPN (Synthetic
     preparation); PREP (Preparation)
        (dispersant; for concrete)
IT
     201793-30-8P
     RL: IMF (Industrial manufacture); SPN (Synthetic
    preparation); PREP (Preparation)
        (dispersant; for concrete)
RN
     201793-30-8 HCAPLUS
     Butanedioic acid, mercapto-, sodium salt, telomer with methyl
CN
     2-propenoate, 2-methyl-2-propenoic acid and \alpha,\alpha'-[[(1-oxo-2-
     propenyl)imino]di-2,1-ethanediyl]bis[ω-hydroxypoly(oxy-1,2-
```

PEZZUTO 10/025588 10/28/03 Page 82

ethanediyl)] (9CI) (CA INDEX NAME)

CM 1

CRN 22275-72-5

CMF C4 H6 O4 S . x Na

$$\begin{array}{c} & \text{SH} \\ | \\ \text{HO}_2\text{C--} \text{CH---} \text{CH}_2\text{---} \text{CO}_2\text{H} \end{array}$$

●x Na

CM 2

CRN 201793-29-5

CMF (C4 H6 O2 . C4 H6 O2 . (C2 H4 O)n (C2 H4 O)n C7 H13 N O3)x

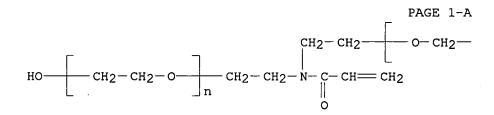
CCI PMS

CM 3

CRN 51601-34-4

CMF (C2 H4 O)n (C2 H4 O)n C7 H13 N O3

CCI PMS



PAGE 1-B

CM 4

CRN 96-33-3 CMF C4 H6 O2

CRN 79-41-4 CMF C4 H6 O2

СH₂ || ме-с-со₂н

55:45

IC

CC

solution 46 g.

ICM C08F008-30

```
L60 ANSWER 24 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN
AN
     1997:805770 HCAPLUS
     128:49011
DN
     Thermosetting aqueous compositions for binders for molded products based
ΤI
     on fibers, chips, etc.
     Reck, Bernd; Wistuba, Eckehardt; Beckerle, Wilhelm Friedrich; Mohr,
TN
     Jurgen; Kistenmacher, Axel; Roser, Joachim
     BASF Aktiengesellschaft, Germany; Reck, Bernd; Wistuba, Eckehardt;
PA
     Beckerle, Wilhelm Friedrich; Mohr, Jurgen; Kistenmacher, Axel; Roser,
     Joachim
     PCT Int. Appl., 47 pp.
SO
     CODEN: PIXXD2
DТ
     Patent
LA
    German
FAN.CNT 1
     PATENT NO.
                     KIND DATE
                                          APPLICATION NO. DATE
                     A1 19971204
                                           WO 1997-EP2796
                                                          19970528
PΙ
    WO 9745461
        W: AU, BR, CA, CN, JP, KR, MX, US
        RW: AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE
                                          DE 1996-19621573 19960529
                           19971204
     DE 19621573
                     A1
                                           AU 1997-30929
    AU 9730929
                      A1
                            19980105
                                                            19970528
                                           EP 1997-925964
     EP 902796
                      A1
                            19990324
                                                            19970528
     EP 902796
                            20000329
                      В1
           AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
        R:
            IE, FI
    AT 191225
                      Ε
                           20000415
                                           AT 1997-925964
                                                            19970528
                      Т3
                           20000716
                                           ES 1997-925964
                                                            19970528
    ES 2146097
     JP 2000511572
                      T2
                           20000905
                                           JP 1997-541621
                                                            19970528
                                           US 1998-147310
    US 6114464
                      Α
                           20000905
                                                            19981125
                                           KR 1998-709722
     KR 2000016154
                      Α
                            20000325
                                                            19981128
PRAI DE 1996-19621573 A
                           19960529
    WO 1997-EP2796
                      W
                           19970528
     Thermosetting mixts. containing hydroxyalkylated polyamines and polymers
AB
    prepared from 5-100% ≥1 ethylenically unsatd. mono- or dicarboxylic
     acid are useful as HCHO-free aqueous binders for the title products. Va
     typical binder composition tested in wood fiberboards contained 50% aqueous
```

acrylic acid-maleic acid copolymer solution 400, 76% aqueous hydroxyethylated hexamethylenediamine (1.05 ethylene oxide) solution 79, and 25% aqueous NH40H

ICS C08K005-17; C08J005-04; C09J133-06

37-6 (Plastics Manufacture and Processing)

```
Section cross-reference(s): 43
     thermosetting waterborne binder polymeric polycarboxylic acid;
ST
     hydroxyethylated hexamethylenediamine crosslinker polymeric
     polycarboxylic acid; maleic acid copolymer waterborne
     thermosetting binder; acrylic acid copolymer waterborne thermosetting
     binder; fiberboard binder formaldehyde free; hydroxyalkylated polyamine
     thermosetting waterborne binder
IT
     Binders
        (polymeric polycarboxylic acids and hydroxylated polyamines;
        thermosetting aqueous compns. for binders for molded products based on
        fibers and chips)
     200057-02-9P 200057-04-1P 200057-06-3P
TΤ
     200057-08-5P 200057-10-9P
                                 200057-12-1P
     200057-14-3P 200057-16-5P 200057-18-7P
     RL: IMF (Industrial manufacture); PRP (Properties); TEM
     (Technical or engineered material use); PREP (Preparation); USES
     (Uses)
        (thermosetting aqueous compns. for binders for molded products based on
        fibers and chips)
IT
     200057-02-9P 200057-04-1P 200057-06-3P
     200057-08-5P 200057-10-9P 200057-14-3P
     200057-16-5P 200057-18-7P
     RL: IMF (Industrial manufacture); PRP (Properties); TEM
     (Technical or engineered material use); PREP (Preparation); USES
     (Uses)
        (thermosetting aqueous compns. for binders for molded products based on
        fibers and chips)
RN
     200057-02-9 HCAPLUS
CN
     2-Butenedioic acid (2Z)-, polymer with \alpha,\alpha',\alpha'',\alpha'
     ''-[1,6-hexanediylbis(nitrilodi-2,1-ethanediyl)]tetrakis[ω-
     hydroxypoly(oxy-1,2-ethanediyl)] and 2-propenamide (9CI) (CA INDEX NAME)
     CM
          1
          39968-51-9
     CRN
          (C2 H4 O)n (C2 H4 O)n (C2 H4 O)n (C2 H4 O)n C14 H32 N2 O4
     CMF
     CCI
```

PAGE 1-B

$$- CH_2 \longrightarrow_n OH$$
 $- CH_2 - CH_2 \longrightarrow_n OH$
 $- CH_2 - CH_2 \longrightarrow_n OH$

CM 2

CRN 110-16-7 CMF C4 H4 O4

Double bond geometry as shown.

CM 3

CRN 79-06-1 CMF C3 H5 N O

RN 200057-04-1 HCAPLUS

CN 2-Butenedioic acid (2Z)-, polymer with aziridine, oxirane and 2-propenoic acid (9CI) (CA INDEX NAME)

CM 1

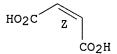
CRN 151-56-4 CMF C2 H5 N

H N

CM 2 ·

CRN 110-16-7 CMF C4 H4 O4

Double bond geometry as shown.



CM 3

CRN 79-10-7 CMF C3 H4 O2

CM 4

CRN 75-21-8 CMF C2 H4 O



RN 200057-06-3 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, polymer with α,α',α'',.alph
a.'''-[1,2-ethanediylbis(nitrilodi-2,1-ethanediyl)]tetrakis[ωhydroxypoly(oxy-1,2-ethanediyl)] and methyl 2-methyl-2-propenoate (9CI)
(CA INDEX NAME)

CM 1

CRN 27014-42-2 CMF (C2 H4 O)n (C2 H4 O)n (C2 H4 O)n (C2 H4 O)n C10 H24 N2 O4

CCI PMS

PAGE 1-A

$$CH_2-CH_2-CH_2-O$$
 $CH_2-CH_2-CH_2-O$
 CH_2-CH_2-O
 CH_2-CH_2-O
 CH_2-CH_2-O
 CH_2-CH_2-O
 CH_2-CH_2-O
 CH_2-CH_2-O
 CH_2-CH_2-O
 CH_2-O

PAGE 1-B

$$-CH_2$$
 OH

 $-CH_2$ OH

 $-CH_2$ OH

 $-CH_2$ OH

 $-CH_2$ OH

 $-CH_2$ OH

 $-CH_2$ OH

CM 2

CRN 80-62-6

CMF C5 H8 O2

CM 3

CRN 79-41-4 CMF C4 H6 O2

RN 200057-08-5 HCAPLUS CN 2-Propenoic acid, 2-methyl-, polymer with α -hydro- ω -hydroxypoly(oxy-1,2-ethanediyl) ether with 2,2',2'',2'''-[[(2-hydroxyethyl)imino]bis(2,1-ethanediylnitrilo)]tetrakis[ethanol] (5:1) and PEZZUTO 10/025588 10/28/03 Page 88

2-propenoic acid (9CI) (CA INDEX NAME)

CM 1

CRN 39940-23-3

CMF (C2 H4 O)n C14 H33 N3 O5

CCI PMS

PAGE 1-B

CM 2

CRN 79-41-4 CMF . C4 H6 O2

CM 3

CRN 79-10-7 CMF C3 H4 O2

RN 200057-10-9 HCAPLUS

CN 2-Propenoic acid, polymer with ethyl 2-propenoate and α -hydrowypoly(oxy-1,2-ethanediyl) ether with 2,2',2'',2'''-[[(2-hydroxyethyl)imino]bis(2,1-ethanediylnitrilo)]tetrakis[ethanol] (5:1) (9CI) (CA INDEX NAME)

CM 1

CRN 39940-23-3

CMF (C2 H4 O)n C14 H33 N3 O5 CCI PMS

PAGE 1-A

HO
$$CH_2 - CH_2 - O$$
 $CH_2 - CH_2$ $CH_2 - CH_2$ $CH_2 - CH_2 - O$ $CH_2 - CH_2 - O$

PAGE 1-B

CM 2

CRN 140-88-5 CMF C5 H8 O2 PEZZUTO 10/025588 10/28/03 Page 90

CM 3

CRN 79-10-7 CMF C3 H4 O2

$$0 \\ || \\ HO-C-CH \longrightarrow CH_2$$

RN 200057-14-3 HCAPLUS

CN 2-Propenoic acid, polymer with $\alpha,\alpha',\alpha'',\alpha'''-[1,6-hexanediylbis(nitrilodi-2,1-ethanediyl)]tetrakis[<math>\omega$ -hydroxypoly(oxy-1,2-ethanediyl)] (9CI) (CA INDEX NAME)

CM 1

CRN 39968-51-9

CMF (C2 H4 O)n (C2 H4 O)n (C2 H4 O)n (C2 H4 O)n C14 H32 N2 O4 CCI PMS

PAGE 1-A
$$CH_{2}-CH_{2}-CH_{2}-O-CH_{2$$

PAGE 1-B

$$-$$
 СН2 $-$ ОН $-$ СН2 $-$ СН2 $-$ ОН $-$ ОН $-$ СН2 $-$ СН2 $-$ ОН

CRN 79-10-7 CMF C3 H4 O2

RN 200057-16-5 HCAPLUS

CN 2-Butenedioic acid (2Z)-, polymer with aziridine, 2,2',2''- nitrilotris[ethanol], oxirane and 2-propenoic acid (9CI) (CA INDEX NAME)

CM 1

CRN 151-56-4 CMF C2 H5 N



CM 2

CRN 110-16-7 CMF C4 H4 O4

Double bond geometry as shown.

CM 3

CRN 102-71-6 CMF C6 H15 N O3

CM 4

CRN 79-10-7

PEZZUTO 10/025588 10/28/03 Page 92

CMF C3 H4 O2

CM 5

CRN 75-21-8 CMF C2 H4 O



RN 200057-18-7 HCAPLUS

CN 2-Butenedioic acid (2Z)-, polymer with $\alpha,\alpha',\alpha'',\alpha''$ ''-[1,2-ethanediylbis[nitrilobis(methyl-2,1-ethanediyl)]]tetrakis[ω hydroxypoly[oxy(methyl-1,2-ethanediyl)]] and 2-propenoic acid (9CI) (CA
INDEX NAME)

CM 1

CRN 51178-86-0

CMF (C3 H6 O)n (C3 H6 O)n (C3 H6 O)n (C3 H6 O)n C14 H32 N2 O4 CCI IDS, PMS

4 (D1-Me)

PAGE 1-B

CM 2

CRN 110-16-7 CMF C4 H4 O4

Double bond geometry as shown.

CM 3

CRN 79-10-7 CMF C3 H4 O2

L60 ANSWER 25 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 1997:90304 HCAPLUS

DN 126:105517

TI Reactive lubricating agent-containing hydrophilic coating compositions for heat-exchanger fins, fins therewith, and their manufacture

IN Kato, Masashi; Katsumata, Tsuyoshi; Masago, Chihiro; Kawahara, Rieko

PA Mitsubishi Aluminium, Japan; Kyoeisha Kagaku Kk

SO Jpn. Kokai Tokkyo Koho, 16 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 08302277	A2	19961119	JP 1995-106657	19950428
PRAI	JP 1995-106657		19950428		

KATHLEEN FULLER EIC 1700/PARKER LAW 308-4290

```
Title aqueous compns. comprise aqueous epoxides, hydrophilic resins, and
reactive
     lubricating agents prepared by reacting nonionic polymeric surfactants
     (having a m.p. of ≥50° and a Karabinos cloudy point of
     ≥15.5) with polycarboxylic acids and with organic bases. An
    Al fin was sprayed with an aqueous composition containing acrylic
acid-acrylonitrile-n-
     amyl methacrylate-Bu acrylate-crotonic acid-cyclohexyl
     methacrylate-2,3-dihydroxypropyl acrylate-2-hydroxyethyl
    methacrylate-itaconic acid-maleic anhydride-Me acrylate-potassium
     2-acrylamido-2-methylpropane-1-sulfonate-potassium 2-hydroxy-3-
     acryloxypropane-1-sulfonate-sodium allylsulfonate copolymer ammonium
     sodium and Et3N salt, diethylene glycol diglycidyl ether, and
     polyoxyethylene 2-ethylhexyl ether monosuccinate NH4+ and Et3N salt to
     form a fin with water-contact angle ≤30° after soaking in
     water for 240 h, dynamic friction coefficient (without spreading with press.
     oils) \leq 0.1, and good oil-resistant adhesion ad press
    processability.
IC
    ICM C09D163-00
     ICS C09D005-00
     42-10 (Coatings, Inks, and Related Products)
     Section cross-reference(s): 46
     Carboxylic acids, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (polycarboxylic, for manufacture of reactive lubricants; reactive
        lubricant-containing aqueous epoxide-crosslinked hydrophilic resin coatings
for
        heat changers)
     185424-98-0P
IT
                    185425-00-7P
                                   185425-01-8DP, hydrogenated castor oil
    monoethers
                 185425-03-0P 185425-04-1P
                                              185425-06-3P
     185425-08-5P
                   185468-78-4P
                                  185531-74-2P
    RL: IMF (Industrial manufacture); MOA (Modifier or additive
    use); PREP (Preparation); USES (Uses)
        (reactive lubricant; reactive lubricant-containing aqueous
epoxide-crosslinked
       hydrophilic resin coatings for heat changers)
TT
     185425-04-1P
     RL: IMF (Industrial manufacture); MOA (Modifier or additive
    use); PREP (Preparation); USES (Uses)
        (reactive lubricant; reactive lubricant-containing aqueous
epoxide-crosslinked
       hydrophilic resin coatings for heat changers)
     185425-04-1 HCAPLUS
    Ethanol, 2,2',2''-nitrilotris-, compd. with \alpha-(3-carboxy-1-oxo-2-
CN
    propenyl) -\omega - [(1-oxooctadecyl) oxy] poly(oxy-1, 2-ethanediyl) and
    N, N-diethylethanamine, ammonium salt (9CI) (CA INDEX NAME)
    CM
         1
    CRN
         32106-74-4
    CMF
         (C2 H4 O)n C22 H38 O5
    CCI
         PMS
```

PEZZUTO 10/025588 10/28/03 Page 95

$$_{\text{HO}_2\text{C}-\text{CH}=\text{CH}-\text{C}-\text{C}-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CH}_2}$$

CM 2

CRN 121-44-8 CMF C6 H15 N

Et | Et-N-Et

CM 3

CRN 102-71-6 CMF C6 H15 N O3

L60 ANSWER 26 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 1996:569304 HCAPLUS

DN 125:228957

TI Water-soluble polymer-based cement dispersant compositions

IN Tanaka, Akira; Myauchi, Haruyoshi; Ooshima, Nobuo; Nakamura, Yoshinobu

PA Toho Chem Ind Co Ltd, Japan

SO Jpn. Kokai Tokkyo Koho, 9 pp. CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE

PI JP 08165156 A2 19960625 JP 1993-48797 19930216

PRAI JP 1993-48797 19930216

The compns. contain the polymers obtained by copolymg. CH2:CRCO2M (R = H, Me; M = monovalent alkali metal, alkanolammonium) 0-20, CH2:CRCO2(C2H4O)x(R'O)yH (x = 5-30; y = 1-5; R' = propylene, butylene) 10-80, and sulfonates 10-70%. When the low-foaming dispersant compns. are used, the cement compns. obtained have low air content and high initial-fluidity, stability, and dispersibility.

IC ICM C04B024-26

ICS C08L033-02; C08L033-14

CC 58-1 (Cement, Concrete, and Related Building Materials) Section cross-reference(s): 37

ST water sol polymer cement dispersant; acrylic polyoxyalkylene

C6 H15 N O3

CRN 102-71-6

CM 2

CMF

CM 3

CRN 1184-84-5 CMF C2 H4 O3 S

 $H_2C = CH - SO_3H$

CRN 79-10-7 CMF C3 H4 O2

CM 5

CRN 181487-76-3

CMF (C4 H8 O . C2 H4 O) x . C3 H4 O2

CM 6

CRN 79-10-7 CMF C3 H4 O2

CM 7

CRN 9064-31-7

CMF (C4 H8 O . C2 H4 O) \times

CCI PMS

CM 8

CRN 26249-20-7

CMF C4 H8 O

CCI IDS

$$_{\rm H3C-CH_2-CH_2-CH_3}$$

CM 9

CRN 75-21-8 CMF C2 H4 O



```
L60 ANSWER 27 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN
ΑN
     1996:567696 HCAPLUS
DN
     125:255282
     Study on synthesis and application of polymer dispersion for
TI
     cement modifier (II) -waterproofing effect on cement
     mortar using acrylic copolymer-
ΑU
     Kim, Hong-Dai; Kim, Young-Geun; Kim, Seung-Jin; Park, Hong-Soo
     Korea Institute of Construction Materials, Seoul, 152-023, S. Korea
CS
SO
     Kongop Hwahak (1996), 7(4), 679-690
     CODEN: KOHWE9; ISSN: 1225-0112
PΒ
     Korean Society of Industrial and Engineering Chemistry
DΤ
     Journal
LΑ
     Korean
AB
     Acrylic copolymer was synthesized from 2-dimethylaminoethyl methacrylate
     and alkylmethacrylate containing long chain hydrocarbon groups. To facilitate
     emulsification in water, acrylic copolymer was treated with acetic acid,
     and therefore acetated acrylic copolymer was produced. Acetated acrylic
     copolymer was perfectly emulsified in water and showed increased emulsion
     stability. Polymer as a cement dispersion agent (PDCM-PSD) was
     prepared by blending the newly synthesized acetated acrylic copolymer with
     sodium gluconate, oleic acid, and triethanolamine. The applicability of
     the blended polymer was examined, and it was found that the effects on
     dispersion and water-proofing (0.3.apprx.0.5) were excellent.
CC
     58-2 (Cement, Concrete, and Related Building Materials)
     cement mortar acrylic copolymer waterproofing additive;
ST
     waterproofing agent acrylic copolymer cement mortar; dispersing
     agent acrylic copolymer cement mortar
IT
     Dispersing agents
        (acrylic copolymer; synthesis and dispersion/waterproofing effects of
        acrylic copolymer additive for cement mortar)
IT
    Mortar
        (cement; synthesis and dispersion/waterproofing effects of
        acrylic copolymer additive for cement mortar)
IT
    Waterproofing
        (agents, acrylic copolymer; synthesis and dispersion/waterproofing
        effects of acrylic copolymer additive for cement mortar)
     102-71-6P, Ethanol, 2,2',2''-nitrilotris-, preparation
IT
     2-Diethylaminoethyl methacrylate
                                       112-80-1P, Oleic acid, preparation
     527-07-1P, Sodium gluconate 25267-71-4P 32360-05-7P, Stearyl
    Methacrylate
    RL: PRP (Properties); SPN (Synthetic preparation); TEM
     (Technical or engineered material use); PREP (Preparation); USES
        (dispersion/waterproofing agent; synthesis and dispersion/waterproofing
       effects of acrylic copolymer additive for cement mortar)
IT
    25267-71-4P
    RL: PRP (Properties); SPN (Synthetic preparation); TEM
     (Technical or engineered material use); PREP (Preparation); USES
     (Uses)
        (dispersion/waterproofing agent; synthesis and dispersion/waterproofing
       effects of acrylic copolymer additive for cement mortar)
     25267-71-4 HCAPLUS
RN
    2-Propenoic acid, 2-methyl-, 2-(diethylamino)ethyl ester, polymer with
CN
```

PEZZUTO 10/025588 10/28/03 Page 99

octadecyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM

CRN 32360-05-7 CMF C22 H42 O2

$$\begin{array}{c|c} & \text{O} & \text{CH}_2 \\ & || & || \\ \text{Me-} & (\text{CH}_2)_{17} - \text{O-} \text{C-} \text{C-} \text{Me} \end{array}$$

CM 2

CRN 105-16-8 CMF C10 H19 N O2

$$^{\rm H_2C}_{||}$$
 O $_{||}$ $^{\rm H_2C}_{||}$ Me- C- C- O- CH₂- CH₂- NEt₂

L60 ANSWER 28 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 1996:345444 HCAPLUS

125:13493 DN

Pre-coated steel panels and their manufacture ΤI

Aoki, Susumu; Katsuyama, Hiroki; Mizutani, Hiroki; Ishidoya, Masahiro ΤN

Nippon Oils & Fats Co Ltd, Japan PA

Jpn. Kokai Tokkyo Koho, 29 pp. SO

CODEN: JKXXAF

DΤ Patent

LА Japanese

FAN.CNT 1

containing

PΤ

PATENT NO. APPLICATION NO. DATE KIND DATE _____ ___ ----**-**_____ JP 1994-304227 19941115 JP 08073811 A2 19960319 PRAI JP 1994-168628 19940629

Coated steel panels, with good chemical, impact, soil, solvent, and weather resistance, are prepared by spreading stainless or Zn (alloy) - and/or Al (alloy)-plated steel panels with primers, (optionally with middle compns.) and covering with thermally curable top compns. containing polymers (A)

 \geq 2 (R4Y)CR1(OCO)CHR2R3 groups (R1-R3 = H, C1-18 hydrocarbyl; R4 = C1-18 hydrocarbyl; Y = O, S; R3 and R4 could form rings with Y), polymers (B) containing functional groups reactive to functional groups of A, and/or latent acidic catalysts, or containing self-crosslinkable polymers containing functional groups in A and B and/or the catalysts. A top composition comprised Bu methacrylate-(1-ethoxy)ethyl methacrylate-2-ethylhexyl acrylate-Me methacrylate copolymer, Denacol Ex 421, TiO2, and a leveling agent in xylene and BuOAc and was baked at 230° for 50 s to form a 15- μm film.

ICM C09D201-02 IC

ICS B05D007-14; B05D007-24; B32B015-08

42-10 (Coatings, Inks, and Related Products) CC

77-99-6DP, Trimethylolpropane, reaction products with Me orthoformate and ΙT Bu vinyl ether, polymers with latent polycarboxylic group-containing acrylic polymers 85-42-7DP, Hexahydrophthalic anhydride, reaction products with diols or polyols and alkyl vinyl ethers, polymers with glycidyl-containing acrylic polymers 106-14-9DP, 12-Hydroxystearic acid, reaction products with HMDI trimer and Pr vinyl ether, polymers with 109-53-5DP, Isobutyl vinyl ether, glycidyl-containing acrylic polymers reaction products with (methyl)hexahydrophthalic anhydride and diols or 109-92-2DP, polyols, polymers with glycidyl-containing acrylic polymers Ethyl vinyl ether, reaction products with (methyl)hexahydrophthalic anhydride and diols or polyols, polymers with glycidyl-containing acrylic 111-34-2DP, Butyl vinyl ether, reaction products with Me orthoformate and timethylolpropane, polymers with latent polycarboxylic group-containing acrylic polymers 115-77-5DP, Pentaerythritol, reaction products with methylhexahydrophthalic anhydride and alkyl vinyl ethers, polymers with glycidyl-containing acrylic polymers 126-30-7DP, Neopentyl glycol, reaction products with hexahydrophthalic anhydride and hydroxysiloxanes and Pr vinyl ether, polymers with glycidyl-containing acrylic polymers 149-73-5DP, Methyl orthoformate, reaction products with Bu vinyl ether and timethylolpropane, polymers with latent polycarboxylic group-containing acrylic polymers 764-47-6DP, Propyl vinyl ether, reaction products with HMDI trimer and hydroxystearic acid, polymers with glycidyl-containing acrylic polymers 25550-51-0DP, Methylhexahydrophthalic anhydride, reaction products with diols or polyols and alkyl vinyl ethers, polymers with glycidyl-containing 28574-90-5DP, HMDI trimer, reaction products with acrylic polymers hydroxystearic acid and Pr vinyl ether, polymers with glycidyl-containing acrylic polymers 62695-06-1DP, Butyl methacrylate-2-ethylhexyl acrylate-glycidyl methacrylate-methyl methacrylate copolymer, polymers with latent polycarboxylic compds. 111310-09-9DP, k-Flex 188-50, reaction products with hexahydrophthalic anhydride and alkyl vinyl ethers, polymers with glycidyl-containing acrylic polymers 131715-70-3DP, Flexorez Ud 320, reaction products with hexahydrophthalic anhydride and alkyl vinyl ethers, polymers with glycidyl-containing acrylic polymers 138636-57-4DP, Placcel E 488, reaction products with methylhexahydrophthalic anhydride and alkyl vinyl ethers, polymers with glycidyl-containing acrylic polymers 151165-17-2DP, polymers with hydroxy siloxanes or polyol-polyacetal reaction products 170146-03-9P 170146-07-3P 170146-08-4P 170385-33-8P 170146-04-0P RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (thermally curable functional acrylic polymer top coatings for steel panels) IT 170146-04-0P RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (thermally curable functional acrylic polymer top coatings for steel panels) 170146-04-0 HCAPLUS RN CN 1-Aziridinepropanoic acid, 2-[[3-(1-aziridinyl)-1-oxopropoxy]methyl]-2-

ethyl-1,3-propanediyl ester, polymer with butyl 2-methyl-2-propenoate,

tetrahydro-2H-pyran-2-yl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

2-ethylhexyl 2-propenoate, methyl 2-methyl-2-propenoate and

CM 1

CRN 52858-59-0 CMF C9 H14 O3

CRN 52234-82-9 CMF C21 H35 N3 O6

CM 3

CRN 103-11-7 CMF C11 H20 O2

$$\begin{array}{c} \text{O} & \vdots \\ \parallel & \vdots \\ \text{CH}_2-\text{O-C-CH} \longrightarrow \text{CH}_2 \\ \parallel & \vdots \\ \text{Et-CH-Bu-n} \end{array}$$

CM 4

CRN 97-88-1 CMF C8 H14 O2

$$\begin{array}{c|c} \text{O} & \text{CH}_2 \\ \parallel & \parallel \\ \text{n-BuO-C-C-Me} \end{array}$$

CRN 80-62-6 CMF C5 H8 O2

L60 ANSWER 29 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 1995:995433 HCAPLUS

DN 124:59637

TI Oligoamide-epihalohydrin resins as drainage aids for papermaking

IN Dasgupta, Sunil Priya; Espy, Herbert Hasting

PA Hercules Inc., USA

SO Eur. Pat. Appl., 12 pp.

CODEN: EPXXDW

DT Patent

LA English

FAN. CNT 1

FAN.CNT 1						
	PAT	ENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	ΕP	678620	A2	19951025	EP 1995-105755	19950418
	ΕP	678620	A3	19960717		
		R: AT, DE,	ES, FR	, GB, GR, IT,	NL, SÉ	
	US	5656699	Α	19970812	US 1994-229254	19940418
	CA	2146922	AA	19951019	CA 1995-2146922	19950412
	FI	9501823	Α	19951019	FI 1995-1823	19950413
	NO	9501463	A	19951019	NO 1995-1463	19950418
	AU	9516507	A1	19951026	AU 1995-16507	19950418
	JP	07292102	A2	19951107	JP 1995-92763	19950418
	BR	9501711	Α	19951114	BR 1995-1711	19950418
	CN	1125277	Α	19960626	CN 1995-105749	19950418
PRAI	US	1994-229254		19940418		

- AB A water-soluble drainage-aid composition containing the reaction product of a bifunctional crosslinking agent, e.g. epihalohydrin, with an amine-terminated oligoamide derived from polyalkylenepolyamine and polycarboxylic acid or its derivs. is disclosed. Heating adipic acid with diethylenetriamine at amine/acid molar ratio 1.5 and 170° for 2.6 h gave an oligoamide which was dissolved in water and heated with epihalohydrin at 70° to prepare a draining aid for papermaking.
- IC ICM D21H021-10

ICS D21H017-55

- CC 43-7 (Cellulose, Lignin, Paper, and Other Wood Products)
 Section cross-reference(s): 38
- 25085-15-8P 25212-19-5P, Adipic acid-diethylenetriamine-epichlorohydrin copolymer 26568-79-6P, Adipic acid-epichlorohydrin-triethylenetetramine copolymer 52404-84-9P, Diethylenetriamine-epichlorohydrin-itaconic acid copolymer 52470-29-8P 68867-69-6P, Diethylenetriamine-dimethyl glutarate-epichlorohydrin copolymer 172084-36-5P 172084-37-6P 172084-38-7P

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(oligoamide-epihalohydrin resins as drainage aids for papermaking)
IT 52404-84-9P, Diethylenetriamine-epichlorohydrin-itaconic acid
copolymer

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(oligoamide-epihalohydrin resins as drainage aids for papermaking)

RN 52404-84-9 HCAPLUS

CN Butanedioic acid, methylene-, polymer with N-(2-aminoethyl)-1,2-ethanediamine and (chloromethyl)oxirane (9CI) (CA INDEX NAME)

CM 1

CRN 111-40-0 CMF C4 H13 N3

H2N-CH2-CH2-NH-CH2-CH2-NH2

CM 2

CRN 106-89-8 CMF C3 H5 Cl O

CM 3

CRN 97-65-4 CMF C5 H6 O4

L60 ANSWER 30 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 1995:938441 HCAPLUS

DN 124:95602

TI Slump loss-preventing admixtures for cement

IN Honda, Susumu; Kinoshita, Seigo; Ito, Akinori

PA Nippon Oils & Fats Co Ltd, Japan.

SO Jpn. Kokai Tokkyo Koho, 5 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

 JP 3265784 B2 20020318 PRAI JP 1993-347364 19931224

AB The admixts. contain a copolymer prepared from polyoxyalkylene derivs. having general formula (R1)2N(AO)nR2 (R1 = C1-5-hydrocarbyl; AO = C2-4-oxyalkylene containing ≥50 mol.% oxyethylene; n = 1-200; R2 = C2-5-alkenyl) and ≥1 compds. selected from maleic anhydride, maleic acid, and maleates. Mortar prepared from water 180, cement 360, and sand 1080 g, under addition of 0.15 weight% (based on cement) copolymer prepared from 1.0 mol Me2N(C2H4O)11CH2CH:CH2 and 1.0 mol maleic anhydride had slump value and air content at 0, 30, and 60 min 239, 211, and 186 mm, and 3.0, 3.1, and 3.1%, resp., and initial and final setting time 6 h and 8 h 5 min.

IC ICM C04B024-26

ICS C08F222-02; C08F222-06; C08F290-06

ICI C04B103-30

CC 58-1 (Cement, Concrete, and Related Building Materials)
 Section cross-reference(s): 38

ST slump cement polyoxyalkylene maleate copolymer; maleic acid anhydride polyoxyalkylene cement

IT Cement

(polyoxyalkylene derivative-maleate copolymers for slump loss prevention of cement)

IT 172600-62-3P 172600-63-4P 172600-64-5P 172600-65-6P 172600-66-7P
172600-68-9P 172600-70-3P 172600-72-5P 172600-74-7P
RL: MOA (Modifier or additive use); PNU (Preparation, unclassified);

PREP (Preparation); USES (Uses)

(polyoxyalkylene derivative-maleate copolymers for slump loss prevention of cement)

IT 172600-70-3P 172600-72-5P

RL: MOA (Modifier or additive use); PNU (Preparation, unclassified); PREP (Preparation); USES (Uses)

(polyoxyalkylene derivative-maleate copolymers for slump loss prevention of cement)

RN '172600-70-3 HCAPLUS

CN 2,5-Furandione, polymer with α -[2-(diethylamino)ethyl]- ω -(2-propenyloxy)poly(oxy-1,2-ethanediyl) (9CI) (CA INDEX NAME)

CM 1

CRN 172600-69-0

CMF (C2 H4 O)n C9 H19 N O

CCI PMS

$$H_2C = CH - CH_2 - O - CH_2 - CH_2 - O - CH_2 - CH_2 - CH_2 - NEt_2$$

CM 2

CRN 108-31-6 CMF C4 H2 O3

172600-72-5 HCAPLUS RN

2,5-Furandione, polymer with α -[2-(dibutylamino)ethyl]- ω -(2-CN propenyloxy)poly(oxy-1,2-ethanediyl) (9CI) (CA INDEX NAME)

CM 1

172600-71-4 CRN

(C2 H4 O)n C13 H27 N O CMF

CCI PMS

$${\rm H_2C} = {\rm CH-CH_2-O} - {\rm CH_2-CH_2-O} - {\rm CH_2-CH_2-N\,(Bu-n)\,_2}$$

CM 2

CRN 108-31-6 CMF C4 H2 O3

L60 ANSWER 31 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

1995:909394 HCAPLUS AN

DN 123:315652

Curable compositions containing components having blocked carboxyl groups, TΙ components having groups reactable with carboxyl groups, and acid catalysts for moldings and coatings

Nakane, Yoshinori; Mizutani, Hiroki; Ishibashi, Hayato; Ishidoya, Masahiro IN

PA Nof Corp., Japan

SO Can. Pat. Appl., 157 pp.

CODEN: CPXXEB

Patent DT

LΑ English

FAN.CNT 2						
I	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE	
_						
PI C	CA 2131416	AA	19950307	CA 1994-2131416	19940902	
	JP 08041208	A2	19960213	JP 1994-203026	19940805	
t	JS 5661219	Α	19970826	US 1994-297588	19940829	
J	JS 5922633	Α	19990713	US 1997-844050	19970418	
Į	JS 6030571	Α	20000229	US 1997-862057	19970522	
PRAI 3	JP 1993-243512		19930906			
;	JP 1993-243513		19930906			
5	JP 1994-58368		19940304			

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PEZZUTO 10/025588
                    10/28/03
                                Page 106
     JP 1994-66470
                            19940311
     JP 1994-73778
                            19940322
     JP 1994-79239
                            19940328
     JP 1994-130900
                            19940523
     JP 1994-130901
                            19940523
    US 1994-297588
                            19940829
os
    MARPAT 123:315652
     Compns. containing the title components provide coatings with good chemical and
AΒ
     phys. properties and weather and staining resistance and moldings with
     good mech. and insulating properties and moisture, corrosion, and cracking
     resistance. In 1-component compns. based on the above reactable
     components or a self-crosslinking component, the catalyst is a
     thermal-latent-type based on a combination of an epoxide, a S-containing
     compound, and a Lewis acid, a combination of a compound containing \geq 1 of N,
     O, P, and S atom, a compound containing a halogen, and a Lewis acid containing
     ≥1 of an Al, Zn, and Sn atom, or a combination of a metallic
     chelate, and an organic silicon compound or condensate. A 2-component compns.
     curable at for the above uses contain the above reactable components and
     ≥1 of a Broensted acid, a Lewis acid, and a mixture of a metallic
     chelate and a compound having a silanol group. A typical thermosetting
     molding composition contained 100 parts mixture of 272.3 parts trifunctional
    polycarboxylic acid prepared from 420.4 parts hexahydrophthalic
     anhydride and 134.2 parts trimethylolpropane, 129.2 parts Pr vinyl ether,
     and 0.2 parts 2-ethylhexyl phosphate mixture, 175 parts YD-128 epoxide
     crosslinker, and 7.5 parts thermal latent catalyst containing cyclohexene
     oxide 9.82, Bu2S 14.63, ZnCl2 13.63, and octanoic acid 28.84 parts.
IC
     ICM C08G083-00
     ICS C08L101-02; B05D001-36
CC
     37-6 (Plastics Manufacture and Processing)
     Section cross-reference(s): 42
     170146-03-9P 170146-04-0P 170146-05-1P
                                                170146-06-2P
IT
     170146-07-3P
                  170146-08-4P 170146-09-5P
                                                  170385-33-8P
     RL: IMF (Industrial manufacture); PRP (Properties); TEM
     (Technical or engineered material use); PREP (Preparation); USES
     (Uses)
        (curable compns. containing components having blocked carboxyl groups,
        components having groups reactable with carboxyl groups, and acid
        catalysts for moldings and coatings)
IT
     170146-04-0P
     RL: IMF (Industrial manufacture); PRP (Properties); TEM
     (Technical or engineered material use); PREP (Preparation); USES
     (Uses)
        (curable compns. containing components having blocked carboxyl groups,
        components having groups reactable with carboxyl groups, and acid
        catalysts for moldings and coatings)
     170146-04-0 HCAPLUS
RN
CN
     1-Aziridinepropanoic acid, 2-[[3-(1-aziridinyl)-1-oxopropoxy]methyl]-2-
     ethyl-1,3-propanediyl ester, polymer with butyl 2-methyl-2-propenoate,
     2-ethylhexyl 2-propenoate, methyl 2-methyl-2-propenoate and
     tetrahydro-2H-pyran-2-yl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)
     CM
         1
     CRN 52858-59-0
     CMF C9 H14 O3
```

CRN 52234-82-9 CMF C21 H35 N3 O6

CM 3

CRN 103-11-7 CMF C11 H20 O2

$$\begin{array}{c} \text{O} \\ \parallel \\ \text{CH}_2-\text{O-C-CH} \Longrightarrow \text{CH}_2 \\ \mid \\ \text{Et-CH-Bu-n} \end{array}$$

CM 4

CRN 97-88-1 CMF C8 H14 O2

$$\begin{array}{ccc} & \text{O} & \text{CH}_2 \\ & \parallel & \parallel \\ \text{n-BuO-C-C-Me} \end{array}$$

CM 5

CRN 80-62-6 CMF C5 H8 O2

```
L60 ANSWER 32 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN
AN
    1995:851690 HCAPLUS
DN
    123:259894
    Waterborne crosslinkable coatings using polyoxyalkylene polyamine
TΙ
     crosslinked acetoacetate functional binder
     Serelis, Algirdas Kazimieras; Meekings, Craig David
IN
     ICI Australia Operations Proprietary Ltd., Australia
PA
     PCT Int. Appl., 26 pp.
     CODEN: PIXXD2
DT
    Patent
LA
    English
FAN.CNT 1
                                         APPLICATION NO. DATE
                    KIND DATE
     PATENT NO.
                     ____
                                         ----<del>-</del>
     <del>-----</del>
    WO 9509209 A1 19950406 WO 1994-AU599 19940929
PΙ
        W: AM, AT, AU, BB, BG, BR, BY, CA, CH, CN, CZ, DE, DK, EE, ES, FI,
             GB, GE, HU, JP, KE, KG, KP, KR, KZ, LK, LR, LT, LU, LV, MD, MG,
             MN, MW, NL, NO, NZ, PL, PT, RO, RU, SD, SE, SI, SK, TJ, TT, UA,
             US. UZ
         RW: KE, MW, SD, SZ, AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU,
            MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN,
            TD, TG
     ZA 9407536
                                         ZA 1994-7536
                                                           19940927
                           19950526
                      Α
                      A1
                                          AU 1994-78048
                                                           19940929
                           19950418
     AU 9478048
                           19970522
     AU 678297
                      В2
                           19930929
PRAI AU 1993-1525
                           19940929
     WO 1994-AU599
     The title coating compns., suitable for use as a glossy or semi-gloss
AB
     paint, have as the binder a combination of an aqueous film-forming dispersion
     of addition polymer having acetoacetate functional groups, or two dispersion
     polymers of different glass transition temps., and a water-stable
     polyoxyalkylene polyamine crosslinking agent having ≥2 primary
     amine group with acetoacetate functional groups/primary amine groups mol
     ratio 1-40:4-1. A semi-gloss white paint containing acetoacetoxyethyl
     methacrylate-Bu methacrylate-Me methacrylate copolymer was
     formulated with Jeffamine ED 600, applied to polypropylene panels, and
     dried 1 wk at room temperature to give coated panels having good solvent
     resistance (xylene double rubs ≥200).
IC
     ICM C09D133-08
     ICS C09D133-10; C09D133-12; C09D201-06; C08F246-00
     42-3 (Coatings, Inks, and Related Products)
CC
     polyoxyalkylene polyamine crosslinker acetoacetoxyethyl methacrylate
ST
```

copolymer; waterborne polyoxyalkylene polyamine crosslinkable

coating

IT

Coating materials

(paints, water-thinned, glossy; waterborne crosslinkable coatings using polyoxyalkylene polyamine crosslinked acetoacetate

functional binder)
168898-20-2P 168898-21-3P 169274-66-2P 169274-67-3P

169274-68-4P 169274-69-5P

RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(waterborne crosslinkable coatings using polyoxyalkylene polyamine crosslinked acetoacetate functional binder)

IT 168898-21-3P

IT

RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(waterborne crosslinkable coatings using polyoxyalkylene polyamine crosslinked acetoacetate functional binder)

RN 168898-21-3 HCAPLUS

CN Butanoic acid, 3-oxo-, 2-[(2-methyl-1-oxo-2-propenyl)oxy]ethyl ester, polymer with N,N'-bis(2-aminoethyl)-1,2-ethanediamine, butyl 2-methyl-2-propenoate and methyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 21282-97-3 CMF C10 H14 O5

CM 2

CRN 112-24-3 CMF C6 H18 N4

H2N-CH2-CH2-NH-CH2-CH2-NH-CH2-CH2-NH2

CM 3

CRN 97-88-1 CMF C8 H14 O2

$$\begin{array}{c|c} & \text{O} & \text{CH}_2 \\ & \parallel & \parallel \\ n\text{-BuO-C-C-Me} \end{array}$$

CM 4

CRN 80-62-6 CMF C5 H8 O2

L60 ANSWER 33 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN AN 1995:798751 HCAPLUS 123:294710 DN The study on synthesis and application of polymer dispersion for ΤI cement modifier - the waterproofing effect of cement mortar by poly[DMA-co-DAMA] emulsion -Kim, Young-Geun; Herh, Dong-Seop; Park, Hong-Soo ΑU Korea Institute Construction Materials, Kwachon, 427-010, S. Korea CS Kongop Hwahak (1994), 5(4), 669-80 CODEN: KOHWE9; ISSN: 1225-0112 SO PB Korean Society of Industrial and Engineering Chemistry DT Journal LΑ Korean AB DMA-co-DAMA were synthesized from 2-diethylaminoethyl methacrylate and dodecyl-methacrylate containing long chain hydrocarbon group with hydrophilic and hydrophobic radicals. To facilitate water emulsification, acrylic copolymer was cationized by acetic acid to produce acetated acrylic copolymer. The structures of the synthesized copolymer and acetated copolymers were confirmed by IR, NMR, and mol. weight was measured by GPC, and C. H. N elemental anal. Acetated acrylic copolymers were perfectly emulsified in water and showed increased emulsion stability. Polymer dispersion for cement modifier (PDCM-PDD) was prepared by blending of the quaternized acrylic copolymer synthesized above, sodium silicate, sodium gluconate, oleic acid, and triethanol amine. The result with prepared polymer dispersion of cement modifier was examined, and it was found that excellent waterproofing effect; Water permeability ratio is. 0.44 under the water pressure of 100g/cm2 and 0.55 under 3kg/cm2, and water absorption ratio is 0.36.apprx.0.47 and 1.02 compressive strength ratio at mixed ratio of water/PDCM-PDD is 45 times. 58-1 (Cement, Concrete, and Related Building Materials) Section cross-reference(s): 38 ST acrylic polymer dispersion prepn waterproofing cement IT Cement Mortar (synthesis of 2-diethylaminoethyl methacrylate-dodecylmethacrylate copolymer and use as waterproofing agent for cement mortar) IT Waterproofing (agents, synthesis of 2-diethylaminoethyl methacrylatedodecylmethacrylate copolymer and use as waterproofing agent for cement mortar) IT 167968-84-5P RL: MOA (Modifier or additive use); NUU (Other use, unclassified); SPN (Synthetic preparation); PREP (Preparation); USES (Uses) (waterproofing agent; synthesis of 2-diethylaminoethyl methacrylate-dodecylmethacrylate copolymer and use as waterproofing agent for cement mortar)

IT

167968-84-5P

PEZZUTO 10/025588 10/28/03 Page 111

RL: MOA (Modifier or additive use); NUU (Other use, unclassified);

SPN (Synthetic preparation); PREP (Preparation); USES
(Uses)

(waterproofing agent; synthesis of 2-diethylaminoethyl

methacrylate-dodecylmethacrylate copolymer and use as waterproofing agent for cement mortar)

RN 167968-84-5 HCAPLUS
CN 2-Propenoic acid, 2-methyl-, 2-(diethylamino)ethyl ester, polymer with dodecyl 2-methyl-2-propenoate, acetate (9CI) (CA INDEX NAME)

CM 1

CRN 64-19-7

CMF C2 H4 O2

но- с- сн3 о

CM 2

CRN 29402-61-7 CMF (C16 H30 O2 . C10 H19 N O2)x CCI PMS

CM 3

CRN 142-90-5 CMF C16 H30 O2

 $\begin{array}{c|c} & \text{O} & \text{CH}_2 \\ & || & || \\ \text{Me- (CH}_2)_{11} - \text{O- C- C- Me} \end{array}$

CM 4

CRN 105-16-8 CMF C10 H19 N O2

L60 ANSWER 34 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN AN 1995:542749 HCAPLUS

DN 123:121161

TI The study on synthesis and application of polymer dispersion for **cement** modifier - the waterproofing effect of poly[DMA-co-DAMA] emulsion

Kim, Young-Geun; Lee, Kyong-Woon; Park, Jong-ok; Park, Hong-Soo ΑU Inst. Ceramic Technol., NITI, Seoul, 152-023, S. Korea CS Yongu Pogo - Kungnip Kongop Kisulwon (1993), 43, 748-67 SO CODEN: YPKKED National Industrial Technology Institute PΒ Journal DTKorean LΑ Dodecyl methacrylate-2-diethylaminoethyl methacrylate copolymer AΒ (dMa-co-DAMA) was synthesized from 2-diethylaminoethyl methacrylate and dodecylmethacrylate containing long chain hydrocarbon group with hydrophile radical and hydrophobic radical. To facilitate water emulsification, acrylic copolymer was cationized by acetic acid to produce acetated acrylic copolymer. The structures of the synthesized copolymer and acetated copolymers were confirmed by IR-NMR, and mol. weight was measured by GPC, and C.H.N. elemental anal. Acetated acrylic copolymers were perfectly emulsified in water and showed increased emulsion stability. Polymer dispersion for cement modifier (PDCM-PDD)/were prepared by blending of the quaternized acrylic copolymer synthesized above, sodium silicate, sodium gluconate, oleic acid, and triethanol amine. The result with prepared polymer dispersion of cement modifier was examined, and it was found that it had an excellent waterproofing effect; water permeability ratio is 0.44 under the water pressure of 100 g/cm2, and 0.55 under 3 kg/cm2, and water absorption ratio is 0.36-0.47 and 1.02 compressive strength ratio at mixed ratio of water/PDCM-PDD is 45 times. 58-1 (Cement, Concrete, and Related Building Materials) CC ST copolymer waterproofing agent cement; polymer dispersion cement waterproofing IT Acrylic polymers, uses RL: MOA (Modifier or additive use); USES (Uses) (acetated, emulsifying agents; preparation, dispersion performance and waterproofing ability of 2-diethylaminoethylmethacrylatedodecylmethacrylate copolymer in cement) ΙT Cement (preparation and waterproofing ability of 2-diethylaminoethylmethacrylatedodecylmethacrylate copolymer dispersion in cement) IT Waterproofing (agents, preparation and waterproofing ability of 2diethylaminoethylmethacrylate-dodecylmethacrylate copolymer dispersion IT 29402-61-7P, 2-Diethylaminoethylmethacrylate-dodecylmethacrylate copolymer RL: PRP (Properties); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (dispersing agent; preparation, dispersion performance and waterproofing ability of 2-diethylaminoethylmethacrylate-dodecylmethacrylate copolymer in cement) IT 102-71-6, Triethanol amine., uses 112-80-1, Oleic acid, uses 527-07-1, Sodium gluconate 1344-09-8, Sodium silicate RL: MOA (Modifier or additive use); USES (Uses) (waterproofing agent; preparation and waterproofing ability of 2-diethylaminoethylmethacrylate-dodecylmethacrylate copolymer dispersion in cement)

29402-61-7P, 2-Diethylaminoethylmethacrylate-dodecylmethacrylate

(Technical or engineered material use); PREP (Preparation); USES

RL: PRP (Properties); SPN (Synthetic preparation); TEM

KATHLEEN FULLER EIC 1700/PARKER LAW 308-4290

ΙT

(Uses)

> (dispersing agent; preparation, dispersion performance and waterproofing ability of 2-diethylaminoethylmethacrylate-dodecylmethacrylate copolymer in cement)

29402-61-7 HCAPLUS RN

2-Propenoic acid, 2-methyl-, 2-(diethylamino)ethyl ester, polymer with CN dodecyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM

CRN 142-90-5 CMF C16 H30 O2

$$$^{\circ}$$$
 O CH2 $$\|\ \|\$ Me- (CH2) 11-0-C-C-Me

CM 2

CRN 105-16-8 CMF C10 H19 N O2

$$^{\rm H_2C}$$
 O $^{\rm H_2C}$ $^{$

L60 ANSWER 35 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 1994:484833 HCAPLUS

DN 121:84833

Homogeneous polymeric ammonium salt and its use ΤT

Muehlenbernd, Thomas; Vogt, Heinz; Moench, Dietmar; Hahn, Susanne; Telser, IN Thomas; Kingma, Arend Jouke

BASF A.-G., Germany PA

Eur. Pat. Appl., 10 pp. SO CODEN: EPXXDW

DTPatent

LΑ German

FAN.	CNT	1								•			
	PAT	CENT :	NO.		KI	ND.	DATE			AP	PLICATIO	NO.	DATE
PI	ΕP	5780	82		Αź	2	1994	0112		EP	1993-11	0226	19930626
	ΕP	5780	82		A.	3	1995	0104					
		R:	BE,	DE,	ES,	FR,	, GB,	IT,	NL				
	DΕ	4222	301		A.	1	1994	0113		DE	1992-42	22301	19920708
DDAT	DE	1992	-422	2301			1992	0708					

The salt is based on (a) a polyalkylenimine and/or a polyvinyl amine with >3 amino groups/mol. and (b) a co- or terpolymer of C2H4 with an alkene mono- and/or -dicarboxylic acid of anhydride and/or ester. The salt may be used in medium- or high-voltage cable insulation. Thus, on elastomeric salt was prepared by coextruding 60:20:20 C2H4-lenimine. The ionomeric material had a tear strength 25.7 M/mm2.

ICM C08L023-08 TC

CC 37-6 (Plastics Manufacture and Processing)

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PEZZUTO 10/025588 10/28/03 Page 114
```

Section cross-reference(s): 39

ST polyamine **polycarboxylic** ionomer salt; rubber acrylic polyethylenimine salt

IT 156659-42-6P

RL: PREP (Preparation)

(rubber, preparation and characterization of)

IT 156659-42-6P

RL: PREP (Preparation)

(rubber, preparation and characterization of)

RN 156659-42-6 HCAPLUS

CN 2-Propenoic acid, polymer with ethene and 2-ethylhexyl 2-propenoate, compd. with aziridine homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 26713-20-2

CMF (C11 H20 O2 . C3 H4 O2 . C2 H4) \times

CCI PMS

CM 2

CRN 103-11-7 CMF C11 H20 O2

CM 3

CRN 79-10-7 CMF C3 H4 O2

CM 4

CRN 74-85-1 CMF C2 H4

 $H_2C == CH_2$

CM 5

CRN 9002-98-6

```
CM
               6
          CRN 151-56-4
          CMF C2 H5 N
L60
    ANSWER 36 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN
AN
     1992:430284 HCAPLUS
DN
     117:30284
     Metalworking oils containing carboxylic acid compounds
TI
     Ozeki, Toshio; Sugioka, Michihiro
IN
PA
     Asahi Denka Kogyo K. K., Japan
     Jpn. Kokai Tokkyo Koho, 7 pp.
     CODEN: JKXXAF
DT
     Patent
LΑ
     Japanese
FAN.CNT 1
     PATENT NO.
                     KIND DATE
                                          APPLICATION NO. DATE
     -----
                     ____
                           _____
                                           ______
     JP 03294399
                      A2
                            19911225
                                           JP 1990-96945
PΙ
                                                            19900412
PRAI JP 1990-96945
                            19900412
     The oils contain R[XmOCOR1(CO2H)n]y [R = (partially esterified) polyol
     residue; R1 = polycarboxylic acid residue; X = oxyethylene,
     oxypropylene, their mixture group; m = 2-300; n = 1-5; y = 2-6], their
     alkali metal salts, or their amine salts. The odorless oils have good
     resistance to putrefaction, degradation, and foaming, and are suitable for use
     in drawing, cutting, grinding, pressing, etc.
IC
     ICM C10M129-66
     ICS C10M133-06; C10M145-36
ICI C10N010-02, C10N030-00, C10N030-10, C10N030-16, C10N030-18, C10N040-22,
     C10N040-24
CC
     51-8 (Fossil Fuels, Derivatives, and Related Products)
     Section cross-reference(s): 55, 56
ΙT
     52434-09-0P 141386-43-8P
                                 141386-44-9P 141405-40-5P
     141490-39-3P 141655-86-9P
                                  141655-87-0P 141657-36-5P
                                                                 141911-30-0P
     141911-31-1P
                  141911-32-2P
                                  141976-90-1P 141976-91-2P
     142155-22-4P
                   142155-23-5P
     RL: PREP (Preparation)
        (preparation of, metalworking oils containing, odorless, with good
resistance to
       putrefaction and degradation and foaming)
     141405-40-5P 141976-91-2P
     RL: PREP (Preparation)
        (preparation of, metalworking oils containing, odorless, with good
resistance to
        putrefaction and degradation and foaming)
RN
     141405-40-5 HCAPLUS
     Ethanol, 2,2',2''-nitrilotris-, compd. with (Z,Z,Z)-\alpha,\alpha'-[1-
CN
     [[(1-oxo-9-octadecenyl)oxy]methyl]-1,2-ethanediyl]bis[\omega-[(3-carboxy-
```

PEZZUTO 10/025588

PMS

CMF CCI (C2 H5 N)x

10/28/03

Page 115

1-oxo-2-propenyl)oxy]poly(oxy-1,2-ethanediyl)] (1:1) (9CI) (CA INDEX NAME)

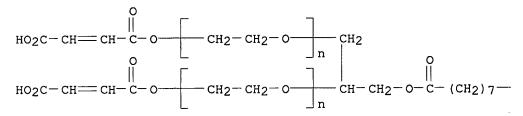
CM 1

CRN 141386-43-8

CMF (C2 H4 O)n (C2 H4 O)n C29 H44 O10

CCI PMS

PAGE 1-A



PAGE 1-B

- CH= CH- (CH₂) 7- Me

CM 2

CRN 102-71-6

CMF C6 H15 N O3

$$\begin{array}{c} {\rm CH_2-CH_2-OH} \\ | \\ {\rm HO-CH_2-CH_2-N-CH_2-CH_2-OH} \end{array}$$

RN 141976-91-2 HCAPLUS

CN Ethanol, 2,2',2''-nitrilotris-, compd. with methyloxirane polymer with oxirane, ether with 1,2,3-propanetriol (3:1), tris[(Z)-hydrogen 2-butenedioate] (9CI) (CA INDEX NAME)

CM 1

CRN 102-71-6 CMF C6 H15 N O3

$$\begin{array}{c} \text{CH}_2-\text{CH}_2-\text{OH} \\ | \\ \text{HO-CH}_2-\text{CH}_2-\text{N-CH}_2-\text{CH}_2-\text{OH} \end{array}$$

CM 2

CRN 52434-09-0

CMF C4 H4 O4 . 1/3 C3 H8 O3 . (C3 H6 O . C2 H4 O) x

CM

CRN 110-16-7 CMF C4 H4 O4

Double bond geometry as shown.

CM 4

CRN 56-81-5 CMF C3 H8 O3

CM5

9003-11-6 CRN

(C3 H6 O . C2 H4 O) \times CMF

CCI PMS

> CM 6

CRN 75-56-9

CMF С3 Н6 О



CM 7 CRN 75-21-8 CMF C2 H4 O



```
L60 ANSWER 37 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN
     1991:248565 HCAPLUS
AN
DN
     114:248565
    Aqueous dispersions of synthetic polymers with increased frost resistance
ΤI
     Stepita, Matej
IN
     Czech.
PA
     Czech., 4 pp.
SO
     CODEN: CZXXA9
DT
     Patent
     Czech
LΆ
FAN.CNT 1
     PATENT NO.
                     KIND DATE
                                           APPLICATION NO.
                                                            DATE
                           _____
                                           ______
                            19900314
                                           CS 1987-8873
                                                            19871207
     CS 268437
                      В1
PΙ
PRAI CS 1987-8873
                            19871207
     The title dispersions, useful as coatings, hydroinsulating compds. and for
     polymer-cement blends, based on copolymers of vinyl chloride
     and/or styrene with acrylic acid esters and 3-20% (meth)acrylic acid
     comonomers, contain mono-, di-, triethylamine, morpholine (I) or I
     derivs., cyclo- or dicyclohexylamine and/or mono-, di-, triethanolamine or
     diethylaminoethanol (II) to form the alkanolamine salt. Thus, 100 parts
     45:52:3 aqueous dispersion of styrene: Bu acrylate: acrylic acid was homogenized
     and neutralized with 3.6 parts II at 20° for 30 min to give a
     dispersion having viscosity 520 mPa-s/20°, frost resistance
     23°, min. film-forming temperature (MFT) 5°, absorption in water
     12.7%/24 h, and elongation 780% compared with 320, 6, 15, 24.9 and 490,
     resp., for dispersion without addition of II.
     ICM C08L033-02
IC
ICA C09D005-02
CC
     37-6 (Plastics Manufacture and Processing)
     Section cross-reference(s): 38, 42, 58
     styrene copolymer salt coating; acrylic acid copolymer salt coating;
ST
     acrylate copolymer salt coating; amine neutralized acrylic acid copolymer;
     cement polymer compn; frost resistance acrylic polymer dispersion
ΙT
     Cement
        (blends with amine-neutralized (meth)acrylic acid-alkyl acrylate
        copolymers, for frost resistance)
     134042-90-3P 134042-91-4P
IT
     RL: PREP (Preparation)
        (aqueous dispersions, preparation of, for frost-resistant coatings)
TI
     134042-91-4P
     RL: PREP (Preparation)
        (aqueous dispersions, preparation of, for frost-resistant coatings)
RN
     134042-91-4 HCAPLUS
     2-Propenoic acid, 2-methyl-, polymer with chloroethene and ethyl
CN
     2-propenoate, compd. with 2-aminoethanol and N,N-diethylethanamine (9CI)
     (CA INDEX NAME)
```

CM

1

CRN 141-43-5 CMF C2 H7 N O

 $_{\rm H_2N-CH_2-CH_2-OH}$

CM 2

CRN 121-44-8 CMF C6 H15 N

Et | Et-N-Et

CM 3

CRN 117647-16-2

CMF (C5 H8 O2 . C4 H6 O2 . C2 H3 C1) \times

CCI PMS

CM 4

CRN 140-88-5 CMF C5 H8 O2

O || EtO-C-CH== CH2

CM 5

CRN 79-41-4 CMF C4 H6 O2

СH₂ || Ме-с-со₂н

CM 6

CRN 75-01-4 CMF C2 H3 Cl $H_2C == CH - C1$

L60 ANSWER 38 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 1990:554356 HCAPLUS

DN 113:154356

TI Oligomeric thixotropic agents and nonaqueous coatings containing them

IN Yun, Han Bo; Smith, Alan

PA Kronos International, Inc., USA

SO Eur. Pat. Appl., 38 pp.

CODEN: EPXXDW DT Patent

LA English

FAN.CNT 1

FAN.		TENT I	. OI		KIN	D	DATE		API	LICATION	NO.	DATE
ΡI	EP	3557	 12		A2	_	19900228		EP	1989-115	181	19890817
•	ΕP	3557	12		A3		19920115					
	ΕP	3557	12		В1		19940727					
		R:	BE,	CH,	DE,	ES,	, FR, GB,	IT,	LI, 1	LU, NL		•
	CA	13352	291		A1		19950418		CA	1989-606	696	19890726
	WO	9002	121		A1		19900308		WO	1989-US3	594	19890824
		W:	BR,	JP,	KR,	SE						
	JP	0450	6527		Т2		19921112		JP	1989-509	065	19890824
	JP	26342	269		В2		19970723					
	US	5034	444		Α		19910723		US	1990-539	545	19900618
	SE	9100	543		Α		19910225		SE	1991-543		19910225
	SE	46572	25		В		19911021					
•	SE	46572	25		С		19920213					
PRAI	US	1988-	-2369	990			19880826					
	WO	1989	-us35	594			19890824					
GT												

Pourable liquid title agents can be added at any stage of paint processing to impart good storage stability and sag resistance with only minimal effects on gloss of a paint. They are compatible with a wide range of polyester, acrylic, and alkyd resin systems, and are especially useful for high-solids coatings. They comprise reaction products of polycarboxylic acids or anhydrides with alkoxylated aliphatic N compds. I and/or II (R = H, Me, Et; R1-2 = C6-30 alkyl;p = 1-20; q, r ≥1, q + r = 2-50; s = 0-1; x + y + z = 1-50), of

polycarboxylic acids or anhydrides with diamines R3NH(CH2)pNH2 (R3 = C6-30 alkyl; p = 1-20), of I and/or II with alkanediol polyepoxide ethers, and/or of R3NH(CH2)pNH2 with alkanediol polyepoxide ethers. Thus, a high-solids Aroplaz 6755-A6-80 (polyester resin) paint containing Cymel 303 (melamine resin hardener), TiO2, and 3.0 lb/100 gal reaction product (III) of 1.0 mol maleic anhydride with 2.0 mol Ethomeon C 20 (ethoxylated coco amine) showed viscosity 570 cP, thixotropy index 1.33, and no sagging when coats up to 7.0 mils thick were applied to inclined surfaces and baked, vs. 301 cP, 0.81, and 3.5 mils without the III, or 410 cP, 0.85, and 4.0 mils using fumed silica instead of III.

IC ICM C09D007-12

CC 42-5 (Coatings, Inks, and Related Products)

88-99-3DP, 1,2-Benzenedicarboxylic acid, polymers with diamines or IT 89-32-7DP, polymers with alkoxylated diamines alkoxylated diamines 105-60-2DP, polymers with maleic anhydride and (alkoxylated) diamines 108-31-6DP, 2,5-Furandione, polymers with diamines or alkoxylated diamines 109-76-2DP, 1,3-Propanediamine, N-fatty alkyl derivs., polymers with dicarboxylic acids or anhydrides or epoxy resins 110-16-7DP, 2-Butenedioic acid (Z)-, polymers with ethoxylated fatty amines 693-23-2DP, Dodecanedioic acid, polymers with ethoxylated fatty amines 2421-28-5DP, polymers with alkoxylated diamines 13281-06-6DP, polymers with dicarboxylic acid anhydrides and ethoxylated fatty amines 26635-92-7DP, Ethomeen 18/12, polymers with dicarboxylic acid anhydrides and ethoxylated fatty amines 73003-90-4DP, polymers with alkoxylated diamines 129613-04-3P 129613-05-4P 129613-06-5P **129613-07-6P** 129613-08-7P **129613-09-8P** 129645-06-3P

RL: PREP (Preparation)

(oligomeric, manufacture of, as thixotropic agents for nonaq. coatings)

129613-07-6P 129613-09-8P

RL: PREP (Preparation)

(oligomeric, manufacture of, as thixotropic agents for nonaq. coatings)

RN 129613-07-6 HCAPLUS

CN 2,5-Furandione, polymer with N-(2-ethylhexyl)-1,3-propanediamine and α,α' -[(octadecylimino)di-2,1-ethanediyl]bis[ω -hydroxypoly(oxy-1,2-ethanediyl)] (9CI) (CA INDEX NAME)

CM 1

ΙŤ

CRN 26635-92-7

CMF (C2 H4 O)n (C2 H4 O)n C22 H47 N O2

CCI PMS

PAGE 1-B

CM 2

CRN 13281-06-6 CMF C11 H26 N2

CM 3

CRN 108-31-6 CMF C4 H2 O3

RN 129613-09-8 HCAPLUS

CN 2-Butenedioic acid (2Z)-, polymer with α,α' - [(octadecylimino)di-2,1-ethanediyl]bis[ω -hydroxypoly(oxy-1,2-ethanediyl)] (9CI) (CA INDEX NAME)

CM 1

CRN 26635-92-7

CMF (C2 H4 O)n (C2 H4 O)n C22 H47 N O2

CCI PMS

PAGE 1-B

CM 2

CRN 110-16-7 CMF C4 H4 O4 Double bond geometry as shown.

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L60 ANSWER 39 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN
AN
         1990:36787 HCAPLUS
DN
         112:36787
         Sulfo group-containing polyesters as plasticizers for cement and
ТT
         plaster of paris and lime
IN
         Wahle, Bernd; Anzinger, Hermann; Ziche, Horst; Schinski, Erhard
PA
         Henkel K.-G.a.A., Fed. Rep. Ger.
SO
         Ger. Offen., 6 pp.
         CODEN: GWXXBX
DT
         Patent
LA
         German
FAN.CNT 1
         PATENT NO.
                                          KIND DATE
                                                                                    APPLICATION NO. DATE
                                                     -----
                                                                                    _____
PΙ
         DE 3743413
                                           A1
                                                       19890629
                                                                                    DE 1987-3743413 19871221
                                         A2
         EP 321818
                                                       19890628
                                                                                     EP 1988-120706 19881212
         EP 321818
                                          A3 19901017
                 R: AT, BE, CH, DE, ES, FR, GB, GR, IT, LI, LU, NL, SE
         US 4880898
                                                     19891114
                                                                                  US 1988-287354 19881220
                                         Α
PRAI DE 1987-3743413
                                                      19871221
         The title polyesters, which help to maintain the fluidity of
         cement and plaster of paris and lime, are prepared by the
         condensation polymerization of fatty acid amides RCON(CkH2kOH)2 (RCO = C8-22
         fatty acid residue; k = 2, 3), dialkanolamino-polyglycolates
         RCO(OCmH2m) nN(CkH2kOH) 2 (m = 2 and/or 3; n = 1-10), or
         \label{eq:conditional} \mbox{dialkanolammoniumalkyl-polyglycolates} \ \mbox{RCO(OCmH2m)} \ \mbox{nN+(CkH2kOH)} \ \mbox{2R1X-} \ \ (\mbox{R1} = \mbox{R1} = \mbox
         C1-4 alkyl; X = alkylation agent anion), with maleic anhydride to acid
         value ≤20, and then are reacted with sulfites. Thus, 1 mol of
         lauric acid diethanolamine amide was polymerized with 1 mol maleic anhydride
         in xylene at 150-160° for 6 h and the polyester reacted with 0.5
         mol Na2S2O5 to produce a title sulfo group-containing polyester.
        ICM C08G063-76
         ICS
                  C08G063-68; C04B028-02; C04B028-14; C04B024-30
ICA C08G063-52
ICI C04B028-02, C04B024-30; C04B028-14, C04B024-30
         35-5 (Chemistry of Synthetic High Polymers)
CC
         Section cross-reference(s): 58
ST
         sulfonated polyester cement plasticizer; plaster paris
         sulfonated polyester plasticizer; lime plasticizer sulfonated polyester;
         maleic anhydride based polyester plasticizer
IT
         Cement
         Lime (chemical)
         RL: USES (Uses)
                (plasticizers for, sulfo group-containing polyesters as)
IT
         Plasticizers
                (sulfo group-containing polyesters, for cement or plaster of
               paris or lime)
```

Amides, polymers

IT

RL: PROC (Process)

(C8-22, N,N-bis(hydroxyalkyl), polymers, with maleic anhydride, reaction products with sulfites, manufacture of, as plasticizers for cement and plaster of paris and lime)

IT Polyesters, preparation

RL: IMF (Industrial manufacture); PREP (Preparation) (sulfo-containing, manufacture of, as plasticizers for **cement** or plaster of paris or lime)

1T 74-87-3DP, Methyl chloride, quaternization products with ethoxylated lauric and myristic acid diethanolamides, polymers with maleic anhydride, reaction products with sodium disulfite 75-21-8DP, Ethylene oxide, reaction products with lauric acid and myristic acid diethanolamide, polymers with maleic anhydride, reaction products with sodium disulfite 7681-57-4DP, Sodium disulfite, reaction products with maleic anhydride-based polyesters 124767-08-4DP, Lauric acid diethanolamide-maleic anhydride-myristic acid diethanolamide copolymer, reaction products with sodium disulfite 124767-09-5DP, reaction products with sodium disulfite

RL: IMF (Industrial manufacture); PREP (Preparation)

(manufacture of, as plasticizers for **cement** or plaster of paris or lime)

1T 124767-08-4DP, Lauric acid diethanolamide-maleic
anhydride-myristic acid diethanolamide copolymer, reaction products with
sodium disulfite 124767-09-5DP, reaction products with sodium
disulfite

RL: IMF (Industrial manufacture); PREP (Preparation)
(manufacture of, as plasticizers for cement or plaster of paris or lime)

RN 124767-08-4 HCAPLUS

CN Tetradecanamide, N,N-bis(2-hydroxyethyl)-, polymer with N,N-bis(2-hydroxyethyl)dodecanamide and 2,5-furandione (9CI) (CA INDEX NAME)

CM 1

CRN 7545-23-5 CMF C18 H37 N O3

HO-
$$CH_2$$
- CH_2 - N - C - $(CH_2)_{12}$ - Me
HO- CH_2 - CH_2

CM 2

CRN 120-40-1 CMF C16 H33 N O3

$$\begin{array}{c} \text{O} \\ || \\ \text{HO-CH}_2\text{--CH}_2\text{--N-C--(CH}_2)}_{10}\text{--Me} \\ || \\ \text{HO-CH}_2\text{--CH}_2 \end{array}$$

CM 3

CRN 108-31-6 CMF C4 H2 O3

RN 124767-09-5 HCAPLUS

CN Dodecanamide, N,N-bis(2-hydroxyethyl)-, polymer with 2,5-furandione (9CI) (CA INDEX NAME)

CM 1

CRN 120-40-1 CMF C16 H33 N O3

$$\begin{array}{c|c} & & & \text{O} \\ || & & \\ \text{HO-CH}_2\text{--CH}_2\text{--N-C-(CH}_2)_{10}\text{--Me} \\ & & \\ \text{HO-CH}_2\text{--CH}_2 \end{array}$$

CM 2

CRN 108-31-6 CMF C4 H2 O3

L60 ANSWER 40 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 1987:637535 HCAPLUS

DN 107:237535

TI Water-soluble film or sheet

IN Fujita, Takeshi; Mori, Shigeo; Kataoka, Hironori; Taniuchi, Akira

PA Daiichi Kogyo Seiyaku Co., Ltd., Japan

SO U.S., 9 pp. Cont.-in-part of U.S. Ser. No. 476,155. CODEN: USXXAM

DT Patent LA English FAN.CNT 2

	PAT	TENT NO.	KIND	DATE	APPLICATION NO.	DATE
ΡI	US	4528360	Α	19850709	US 1984-651486	19840917
	JP	58160316	A2	19830922	JP 1982-43340	19820317
	JP	58179227	A2	19831020	JP 1982-62164	19820413
	JP	05068493	B4	19930929		
	US	4521586	Α	19850604	US 1983-476155	19830317
	JP	04356526	A2	19921210	JP 1991-257726	19911004
	JP	06041511	B4	19940601		
PRAI	JP	1982-43340		19820317		
	JP	1982-62164		19820413		
	US	1983-476155		19830317		

AB Film-formable, water-soluble polyesters of mol. weight ≥50,000 are manufactured by polycondensation of a polycarboxylic acid, its anhydride, or its low-mol.-weight ester with a polyoxyalkylene of average mol. weight ≥3000 prepared by polycondensation of ethylene oxide (I) and optionally, another alkylene oxide with an organic compound having ≥2 active H's. Thus, 106 parts diethylene glycol and 20 parts KOH were gradually treated with 12,000 parts I at 130° and 2 kg/cm2-G to give a polyol having weight-average mol. weight .apprx.10,000. The polyol (100 parts) was treated with 1.85 parts di-Me terephthalate and heated to 120° while distilling MeOH to give a copolymer (II) having weight-average mol. weight .apprx.200,000. II was dissolved in water, and 100 g of the aqueous solution

was evaporated in a dish to give a film having elongation 10.4%, tensile strength 150 kg/cm2, and tear strength 80 kg/cm.

IC ICM C08G063-42

ICS C08G063-66; C08G059-00

NCL 528297000

CC 35-5 (Chemistry of Synthetic High Polymers)
Section cross-reference(s): 37

IT Polyesters, preparation

RL: IMF (Industrial manufacture); PREP (Preparation) (manufacture of, from polyether polyols and polycarboxylic acid derivs.)

75-21-8DP, polymers with alkylene oxides and active H-containing compds. and IT dicarboxylic acid derivs. 75-56-9DP, polymers with alkylene oxides and active H-containing compds. and dicarboxylic acid derivs. 89-32-7DP, reaction products with polyethylenimine-ethylene oxide-propylene oxide 112-57-2DP, ethers with ethylene oxide-propylene copolymers, polymers with di-Me phthalate 120-61-6DP, reaction products with polyethylenimine-ethylene oxide-propylene oxide adduct 131-11-3DP, polymers with tetraethylenepentamine-alkylene oxide reaction products 141-03-7DP, reaction products with polyethylenimine-ethylene oxide adduct 629-11-8DP, polymers with alkylene oxides and dimer acids 9002-98-6DP, reaction products with alkylene oxides and dicarboxylic acid derivs. 9003-11-6DP, ethers with tetraethylenepentamine, polymers with di-Me 25190-06-1DP, reaction products with ethylene oxide and dimer phthalate acids 34937-03-6P 35725-54-3P 37294-00-1P 88077-38-7P 88077-39-8P 88077-40-1P 88077-41-2P 88077-42-3P 88077-44-5P 88077-45-6P **88077-46-7P** 88077-47-8P 88077-48-9P 88077-49-0P 88077-50-3P **88077-51-4P** 88077-54-7P 88077-81-0P 88077-82-1P 88077-83-2P 88077-84-3P 88077-85-4P 88091-62-7P 88169-05-5P 88169-06-6P 88169-07-7P 88248-43-5P 88248-48-0P 95039-05-7P 99651-95-3P 99651-96-4P 99651-97-5P

99692-53-2P 115775-36-5P

RL: PREP (Preparation)

(manufacture of film-formable water-soluble, with high mol. weight)

IT 88077-46-7P 88077-51-4P

RL: PREP (Preparation)

(manufacture of film-formable water-soluble, with high mol. weight)

RN 88077-46-7 HCAPLUS

CN 2-Butenedioic acid (2Z)-, polymer with methyloxirane polymer with oxirane ether with 2,2'-iminobis[ethanol] (2:1) (9CI) (CA INDEX NAME)

CM 1

CRN 110-16-7 CMF C4 H4 O4

Double bond geometry as shown.

CM 2

CRN 79805-70-2

CMF C4 H11 N O2 . 2 (C3 H6 O . C2 H4 O)x

CM 3

CRN 111-42-2 CMF C4 H11 N O2

 $HO-CH_2-CH_2-NH-CH_2-CH_2-OH$

CM 4

CRN 9003-11-6

CMF (C3 H6 O . C2 H4 O) x

CCI PMS

CM 5

CRN 75-56-9 CMF C3 H6 O



CM 6

CRN 75-21-8 CMF C2 H4 O



RN 88077-51-4 HCAPLUS

CN 2-Butenedioic acid (2Z)-, dimethyl ester, polymer with methyloxirane polymer with oxirane ether with 2,2'-iminobis[ethanol] (2:1) (9CI) (CA INDEX NAME)

CM 1

CRN 624-48-6 CMF C6 H8 O4

Double bond geometry as shown.

CM 2

CRN 79805-70-2

CMF C4 H11 N O2 . 2 (C3 H6 O . C2 H4 O) x

CM 3

CRN 111-42-2 CMF C4 H11 N O2

 $HO-CH_2-CH_2-NH-CH_2-CH_2-OH$

CM 4

CRN 9003-11-6

CMF (C3 H6 O . C2 H4 O) \times

CCI PMS

CM 5

CRN 75-56-9 CMF C3 H6 O CH3

CM 6

CRN 75-21-8 CMF C2 H4 O

 $^{\circ}$

L60 ANSWER 41 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 1984:456564 HCAPLUS

DN 101:56564

TI Comb copolymers with polyoxyalkylene and carboxylate salt side chains

IN Tsubakimoto, Tsuneo; Hosoidi, Masahiro; Tahara, Hideyuki

PA Nippon Shokubai Kagaku Kogyo Co., Ltd., Japan

SO Eur. Pat. Appl., 56 pp.

CODEN: EPXXDW

DT Patent

LA English

FAN CNT 1

FAN.		TENT NO.	KIND	DATE	AP	PLICATION NO.	DATE
ΡI	EP	 56627	A2	19820728	EP	1982-100247	19820115
	ΕP	56627	A3	19820804			
	ΕP	56627	B1	19841003			
•		R: DE, FR,	GB, IT				
	JP	57118058	A2	19820722	JP	1981-3776	19810116
	JΡ	58038380	B4	19830823			
	JΡ	57119896	A2	19820726	JP	1981-5913	19810120
	JΡ	59016519	B4	19840416			
	JΡ	58032051	A2	19830224	JP	1981-95928	19810623
	JP	58038381	B4	19830823			
	JP	58006295	A2	19830113	JP	1981-101353	19810701
	JP	59014277	B4	19840403			
	US	4471100	Α	19840911	US	1982-339640	19820115
PRAI	JΡ	1981-3776		19810116			
	JP	1981-5913		19810120			
	JP	1981-95928		19810623			
	JΡ	1981-101353		19810701			

AB Title copolymers, useful as dispersing agents for **cement** in concrete and mortar and for pigments in paints and as scale inhibitors in water, are manufactured by polymerization of polyalkylene glycol monoallyl

25-75, maleic monomer 25-75, and a copolymerizable vinyl monomer 0-50 mol, followed by neutralization with alkali or alkaline earth metal hydroxides, ammonia, or amines. Thus, adding maleic anhydride 139.3, (NH4)2S2O8 14.2, and water 225 parts to 334 parts polyethylene glycol monoallyl ether (average d.p. 5) and 100 parts water in 120 min under N at 95° with

stirring, adding 4.2 parts 20% aqueous (NH4)2S2O8 in 20 min, holding the reaction mixture at 95° for 100 min, and neutralizing with 28% aqueous NH4OH gave 44% solids copolymer salt (I) [91070-72-3] solution with pH 8.0 and viscosity 93 cP. A paste containing 24% I solution 29.9, water 90.8, Emulgen 909 wetting agent 7.1, ethylene oxide 158.5, TiO2 871.2, and Cellosize OP-4400 thickener 36.0 parts was mixed (68.9 parts) with 100 parts Acryset EMN-210E [91196-08-6] acrylic polymer emulsion and 6.5 parts CS-12 film-forming additive to give a paint with viscosities 1714 and 1754 cP immediately after and 1 day after preparation, resp., compared with 3540 and 4490, resp., for a similar paint containing Tamol 731 (diisobutylene-maleic anydride copolymer Na salt) instead of I. C08F216-14; C08F222-00; C04B013-24; C09D007-02; C02F005-00 IC ICI C08F216-14, C08F222-00; C08F222-00, C08F216-14 42-7 (Coatings, Inks, and Related Products) Section cross-reference(s): 35, 58, 61 polyoxyalkylene allyl ether maleate copolymer; polyoxyethylene allyl ether maleate copolymer; dispersant polyoxyethylene allyl maleate copolymer; cement dispersant maleate copolymer; concrete dispersant maleate copolymer; mortar dispersant maleate copolymer; pigment dispersant maleate copolymer; acrylic paint dispersant IT Cement (dispersing agents for, in concrete and mortar, polyethylene glycol allyl ether-maleic copolymer salts as) 84154-79-0P 90819-16-2P 91068-65-4P 91070-72-3P 91070-73-4P IT 91070-75-6P 91070-76-7P RL: PREP (Preparation) (dispersing agents, manufacture of) IT 91070-76-7P RL: PREP (Preparation) (dispersing agents, manufacture of) 91070-76-7 HCAPLUS RN 2-Butenedioic acid (2Z)-, monomethyl ester, polymer with CN α -2-propenyl- ω -hydroxypoly(oxy-1,2-ethanediyl), compd. with 2,2',2''-nitrilotris[ethanol] (9CI) (CA INDEX NAME) CM CRN 102-71-6 CMF C6 H15 N O3 CH2-CH2-OH $HO-CH_2-CH_2-N-CH_2-CH_2-OH$ CM CRN 82940-71-4 CMF (C4 H4 O4 . (C2 H4 O)n C3 H6 O)x CCI PMS

CMF (C2 H4 O)n C3 H6 O

CM

3

CRN . 27274-31-3

CCI PMS

$$HO - CH_2 - CH_2 - O - CH_2 - CH_2 - CH_2 - CH_2$$

CM 4

CRN 110-16-7 CMF C4 H4 O4

Double bond geometry as shown.

L60 ANSWER 42 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 1984:86314 HCAPLUS

DN 100:86314

TI High-molecular-weight compounds

IN Fujita, Takeshi; Mori, Shigeo; Kataoka, Hironori; Taniuchi, Akira

PA Daiichi Kogyo Seiyaku Co., Ltd., Japan

SO Eur. Pat. Appl., 26 pp.

CODEN: EPXXDW

DT Patent

LA English

FAN CNT 2

FAN.	CNT	2									
	PATENT NO.				KIND	DATE		API	PLICATION NO.	DATE	
					-					-	
PI	ΕP	8903	8		A1	19830921		ΕP	1983-102461	19830312	
	ΕP	8903	8		B1	19860917					
		R:	DE,	FR,	GB, NL		•				
	ĴΡ	5816	0316		A2	19830922		JΡ	1982-43340	19820317	
	JP	5817	9227		A2	19831020		JP	1982-62164	19820413	
	JP	0506	8493		B4	19930929					
	JP	0435	6526		A2	19921210		JΡ	1991-257726	19911004	
	JΡ	0604	1511		B4	19940601			•		
PRAI	JP	1982	-4334	40		19820317					
	JΡ	1982	-6216	54		19820413					

AB High-mol.-weight polyesters are prepared by addition polymerizing ethylene oxide

optionally containing an alkylene oxide with an organic compound containing ≥ 2 active H groups to produce a polyhydroxy compound having average mol. weight ≥ 1000 which is then esterified with a **polycarboxylic** acid, anhydride, or lower alkyl ester. Thus, 106 part diethylene glycol and 20 parts KOH were gradually treated with 12,000 parts ethylene oxide at 130° and 2 kg/cm2-G to give a polyol having weight-average mol. weight 10,000. The polyol (100 parts) was treated with 1.85 parts di-Me terephthalate and heated to 120° while distilling MeOH to give a polyester [88077-81-0] having weight-average mol. weight .apprx.200,000. The polymer was dissolved in water and 100 g of the aqueous solution was placed in

dish and evaporated to give a film having elongation 10.0%, tensile strength 150 kg/cm2, and tear strength 80 kg/cm. C08G063-66 IC 35-5 (Chemistry of Synthetic High Polymers) CC Polyesters, preparation IT RL: PREP (Preparation) (preparation of, from polyether polyols and polycarboxylic acids) 75-21-8DP, reaction products with alkylene oxides and polyethylenimine, IT polymers with dicarboxylic acids 75-56-9DP, reaction products with alkylene oxides and polyethylenimine, polymers with dicarboxylic acids 89-05-4DP, polymers with alkylene oxide-phenolic resin reaction products 89-32-7DP, polymers with alkylene oxide-polyethylenimine reaction products 112-57-2DP, ethers with ethylene oxide-propylene oxide copolmers, polymers 120-61-6DP, polymers with polyethylenimine-alkylene with di-Me phthalate 141-03-7DP, polymers with alkylene oxide reaction products oxide-polyethylenimine reaction products 624-48-6DP, polymers with ethylene oxide-propylene oxide copolymer tetraethylenepentamine ethers 9002-98-6DP, reaction products with alkylene oxides, polymers with 9003-11-6DP, ethers with tetraethylenepentamine, dicarboxylic acids 26249-20-7DP, reaction products with polymers with di-Me phthalate alkylene oxides and polyethylenimine, polymers with dicarboxylic acids 72270-44-1DP, polyesters with 37294-00-1P 35725-54-3P 34937-03-6P 88077-40-1P 88077-41-2P 88077-39-8P fatty acids 88077-38-7P 88077-45-6P **88077-46-7P** 88077-42-3P 88077-43-4P 88077-44**-**5P 88077-50-3P 88077-51-4P 88077-48-9P 88077-49-0P 88077-47-8P 88077-54-7P 88077-81-0P 88077-82-1P 88077-83-2P 88077-53-6P ·88077-84-3P 88077-85-4P 88091-62-7P 88169-05-5P 88169-06-6P 88169-07-7P 88248-41-3P 88248-43-5P 88248-48-0P RL: PREP (Preparation) (preparation of high-mol.-weight moldable) IT88077-46-7P 88077-51-4P RL: PREP (Preparation) (preparation of high-mol.-weight moldable) RN 88077-46-7 HCAPLUS 2-Butenedioic acid (2Z)-, polymer with methyloxirane polymer with oxirane CN ether with 2,2'-iminobis[ethanol] (2:1) (9CI) (CA INDEX NAME) CM 1 CRN 110-16-7 C4 H4 O4 CMF Double bond geometry as shown. CO2H 2 CM CRN 79805-70-2 CMF C4 H11 N O2 . 2 (C3 H6 O . C2 H4 O)x

3

CM

10/28/03 Page 133 PEZZUTO 10/025588

> CRN 111-42-2 CMF C4 H11 N O2

но- сн2- сн2- ин- сн2- сн2- он

4 CM

CRN 9003-11-6

CMF (C3 H6 O . C2 H4 O)x

CCI PMS

> CM 5

CRN 75-56-9 CMF СЗ Н6 О

CH3

CM6

CRN 75-21-8 CMF C2 H4 O

88077-51-4 HCAPLUS RN

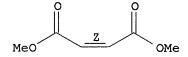
2-Butenedioic acid (2Z)-, dimethyl ester, polymer with methyloxirane polymer with oxirane ether with 2,2'-iminobis[ethanol] (2:1) (9CI) (CA CN INDEX NAME)

CM 1

CRN 624-48-6

C6 H8 O4 CMF

Double bond geometry as shown.



2 CM

CRN 79805-70-2

CMF C4 H11 N O2 . 2 (C3 H6 O . C2 H4 O) \times

CM 3

CRN 111-42-2 CMF C4 H11 N O2

 $HO-CH_2-CH_2-NH-CH_2-CH_2-OH$

CM 4

CRN 9003-11-6

CMF (C3 H6 O . C2 H4 O) \times

CCI PMS

CM 5

CRN 75-56-9 CMF C3 H6 O

СНЗ

CM 6

CRN 75-21-8 CMF C2 H4 O

 $^{\circ}$

L60 ANSWER 43 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 1979:138412 HCAPLUS

DN 90:138412

TI Phosphoric acid esters

IN Reitz, Gunther; Boehmke, Guenther; Jakobs, Karl Hans

PA Bayer A.-G., Fed. Rep. Ger.

SO Ger. Offen., 32 pp.

CODEN: GWXXBX

DT Patent

LA German

FAN.CNT 1

PAN.CNI I				
PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
		-		
PI DE 2726854	A1	19790111	DE 1977-2726854	19770615
EP 125	A1	19790110	EP 1978-100114	19780607
EP 125	B1	19811202		

R: CH, DE, FR, GB

```
JP 54005923
                            19790117
                                           JP 1978-70480
                                                             19780613
                       A2
                            19810331
                                           US 1978-915019
                                                             19780613
    US 4258448
                       Α
PRAI DE 1977-2726854
                            19770615
    Polyesters useful as dispersants, leveling agents, builders, and the like
    were prepared from H3PO4, polyols and(or) epoxides, and optionally
    polycarboxylic acids in mol ratios of 0.05-1.5:1:0-1.5. Thus, 47
     q P2O5 and 33 q H3PO4 were heated, mixed, cooled, treated with 124 g
     ethylene glycol and 1 g NaH2PO2, slowly heated to 180°, heated at
     180° while 35 g water was distilled, cooled, treated with 98 g maleic
     anhydride, and heated 6 h at 190-200° under slowly increasing
     vacuum, distilling an addnl. 10 g liquid The melt was cooled, mixed with 1 L
    water, and adjusted to pH 5-6 with 50% aqueous NaOH, giving a 20% aqueous
solution of
     the polyester [69794-26-9].
IC
     C07F009-09
CC
     35-3 (Synthetic High Polymers)
IT
     Polymerization
        (of phosphoric and phosphorous acids with polyols and
       polycarboxylic acids)
IT
     Polyphosphoric acids
     RL: USES (Uses)
        (polymers with polyols and polycarboxylic acids)
     56-81-5DP, polymers with polyphosphoic acids and polycarboxylic
IT
             75-56-9DP, polymers with polyphosphoric acids and triethanolamine
     102-71-6DP, polymers with polyphosphoric acids and propylene oxide
     13598-36-2DP, polymers with glycerol, phthalic anhydride, and
     polyphosphoric acids
                            15743-39-2DP, polymers with polyphosphoric acids
     69761-10-0P
                   69794-09-8P
                                 69794-10-1P
                                               69794-11-2P
                                                              69794-12-3P
     69794-13-4P
                   69794-14-5P
                                 69794-15-6P
                                               69794-16-7P
                                                              69794-17-8P
                   69794-19-0P
                                 69794-20-3P
                                               69794-21-4P
     69794-18-9P
     69794-22-5P
                   69794-23-6P
                                 69794-24-7P
                                               69794-25-8P
                                                              69794-26-9P
     69822-48-6P
     RL: PREP (Preparation)
        (preparation of)
ΙT
     69794-18-9P
     RL: PREP (Preparation)
        (preparation of)
     69794-18-9 HCAPLUS
RN
     Phosphoric acid, polymer with 2,5-furandione, 2,2',2''-
CN
     nitrilotris[ethanol] and phosphonic acid (9CI) (CA INDEX NAME)
     CM
          1
     CRN 13598-36-2
     CMF H3 O3 P
  0
O- P- O
*** FRAGMENT DIAGRAM IS INCOMPLETE ***
          2
     CM
     CRN
         7664-38-2
     CMF
         H3 O4 P
```

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10/28/03

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CM 3

CRN 108-31-6 CMF C4 H2 O3

CM 4

CRN 102-71-6 CMF C6 H15 N O3

$$_{\rm CH_2-CH_2-OH}^{\rm CH_2-CH_2-OH}$$

HO-CH₂-CH₂-N-CH₂-CH₂-OH

L60 ANSWER 44 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

AN 1967:517717 HCAPLUS

DN 67:117717

TI Substituted acrylamides and polymers therefrom

IN Kelley, Everett J.

PA Rohm and Haas Co.

SO U.S., 7 pp. CODEN: USXXAM

DT Patent

LA English

FAN.CNT 1

PΙ

PATENT NO. KIND DATE APPLICATION NO. DATE
US 3350336 19671031 US 19641229

The title compds. were prepared and used as chemical intermediates and as components in oil-modified alkyd resins to reduce the yellowing tendency of the films. Thus, a mixture containing N-methylolmethacrylamide 173, N,N-bis(carbethoxyethyl)amine 326, C6H6 326, and p-methoxyphenol 0.5 g. was refluxed for 10 hrs., during which 27 g. H2O was collected by a water separator. The C6H6 was removed in vacuo at 100°, and the viscous liquid obtained was cooled and filtered to give a straw-colored N,N-bis(carbethoxyethyl)-N'-methacryloylmethylenediamine (I), n2OD 1.4777 and Br number 6.26. I was homopolymd. by refluxing in C6H6 in the presence of 0.5% azobisisobutyronitrile (II). A mixture containing I 10.5, II 1.4, Me

methacrylate 135.8, and Bu methacrylate 203.7 g. was added during 2 hrs. to 233 g. PhMe at 110-15°. A catalyst solution containing 1.58 g. II in 53 g. PhMe was added to the reaction mixture in 3 equal portions 2, 3, and 4 hrs. after addition of the mixture was completed. The resulting mixture was heated for an addnl. 2 hrs., cooled, and diluted with 180 g. PhMe to give a copolymer solution with 265 cp. viscosity. The copolymer prepared was applied to steel primed with a semigloss Fe2O3-pigmented alkyl coating prepared from a soybeam oil-modified polyester of phthalic anhydride, pentaerythritol, and propylene glycol, and baked for 1 hr. at 200°F. to give an adherent tough coating. An emulsion copolymer composition was prepared by mixing

water, tert-octylhenoxypoly(ethoxyphenol), Et acrylate, II, I, and (NH4)2S2O8. The copolymer prepared was used as an air-drying adherent coating for wood, Al, asbestos **cement**, and stucco panels. N,N-bis(carbomethoxymethyl)-N'-acryloylmethylenediamine and N,N-(carbobutoxyethyl)-N'-methacryloylmethylenediamine were prepared similarly and were copolymd. with styrene, β -hydroxyethyl methacrylate, acrylonitrile, or N-methylolacrylamide.

NCL 260022000

4

CC 36 (Plastics Manufacture and Processing)

IT 15518-24-8P 15518-25-9P 15621-02-0P 30551-82-7P **30551-83-8P**, preparation **30555-30-7P**, preparation **30662-28-3P**

RL: PREP (Preparation)

(preparation of)

IT **30551-83-8P**, preparation **30555-30-7P**, preparation

30662-28-3P

RL: PREP (Preparation)

(preparation of)

RN 30551-83-8 HCAPLUS

CN Methacrylic acid, butyl ester, polymer with diethyl 3,3'[(methacrylamidomethyl)imino]dipropionate and methyl methacrylate (8CI)
(CA INDEX NAME)

CM 1

CRN 15621-02-0 CMF C15 H26 N2 O5

$$\begin{array}{c|c} & \text{O} & \text{CH}_2 \\ || & || \\ \text{O} & \text{CH}_2 - \text{NH} - \text{C} - \text{C} - \text{Me} \\ || & | \\ \text{EtO} - \text{C} - \text{CH}_2 - \text{CH}_2 - \text{N} - \text{CH}_2 - \text{CH}_2 - \text{C} - \text{OEt} \\ || & | \\ \text{O} \end{array}$$

CM 2

CRN 97-88-1 CMF C8 H14 O2

$$\begin{array}{c|c} & \text{O} & \text{CH}_2 \\ & || & || \\ & \text{n-BuO-C-C-Me} \end{array}$$

CM 3

CRN 80-62-6 CMF C5 H8 O2

RN 30555-30-7 HCAPLUS

CN Methacrylic acid methyl ester, polymer with diethyl 3,3'[(methacrylamidomethyl)imino]dipropionate and ethyl acrylate (8CI) (CA INDEX NAME)

CM 1

CRN 15621-02-0 CMF C15 H26 N2 O5

CM 2

CRN 140-88-5 CMF . C5 H8 O2

CM 3

CRN 80-62-6 CMF C5 H8 O2

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RN 30662-28-3 HCAPLUS

CN Methacrylic acid, isobutyl ester, polymer with diethyl 3,3'-[(methacrylamidomethyl)imino]dipropionate (8CI) (CA INDEX NAME)

CM 1

CRN 15621-02-0 CMF C15 H26 N2 O5

CM 2

CRN 97-86-9 CMF C8 H14 O2

$$\begin{array}{c|c} & \text{O} & \text{CH}_2 \\ & \parallel & \parallel \\ \text{i-BuO-C-C-Me} \end{array}$$

=>